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Seasonal and diurnal variations of Hg° over New England

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Abstract. Factors influencing diurnal to interannual variability in Hg° over New England were investigated using multi-year measurements conducted by AIRMAP at the Thompson Farm (TF) coastal site, an inland elevated site at Pac Monadnock (PM), and two month measurements on Appledore Island (AI) in the Gulf of Maine. Mixing ratios of Hg° at TF showed distinct seasonality with maxima in March and minima in October. Hg° at AI tracked the trend at TF but with higher minima, while at PM the diurnal and annual cycles were dampened. In winter, Hg° was correlated most strongly with CO and NOy, indicative of anthropogenic emissions as their primary source. Our analysis indicates that Hg° had a regional background level of \( \sim 160 \text{ fmol/mol} \) in winter, a dry deposition velocity of \( \sim 0.2 \text{ cm s}^{-1} \) with a \( \sim 16 \text{ day} \) lifetime in the coastal boundary layer in summer. The influence of oceanic emissions on ambient Hg° levels was identified using the Hg°-CHBr₃ correlation at both TF and AI. Moreover, the lower Hg° levels and steeper decreasing warm season trend at TF (0.5–0.6 fmol/mol d⁻¹) compared to PM (0.2–0.3 fmol/mol d⁻¹) likely reflected the impact of marine halogen chemistry. Large interannual variability in warm season Hg° levels in 2004 versus 2005/2006 may be due to the role of precipitation patterns in influencing surface evasion of Hg°. In contrast, changes in wintertime maximum levels of Hg° were small compared to drastic reductions in CO, CO₂, NOy, and SO₂ from 2004/2005 to 2006/2007. These trends could be explained by a homogeneous distribution of Hg° over North American in winter due to its long lifetime and/or rapid removal of reactive mercury from anthropogenic sources. We caution that during warmer winters, the Hg°-CO slope possibly reflects Hg° loss relative to changes in CO more than their emission ratio.

1 Introduction

Atmospheric mercury exists in diverse chemical forms comprised of gaseous elemental mercury (Hg°), reactive gaseous mercury (RGM=HgCl₂+HgBr₂+HgOBr+...), and particulate mercury (Hg⁺). The composition of total atmospheric mercury (i.e. the fraction of Hg°, RGM, and Hg⁺) in the continental atmosphere varies geographically due to different land surface types, chemical environments, and human influences.

Numerous studies have measured the ambient level of total gas-phase mercury (TGM=Hg°+RGM) and Hg° at locations world-wide. For example, in the marine atmosphere over the North Atlantic Hg° levels are \( \sim 1.6 \text{ ng m}^{-3} \) (Laurier and Mason, 2007), between 1.6 and 4.7 ng m⁻³ over the North Pacific (Laurier et al., 2003), and 0.4–11.2 ng m⁻³ over the Mediterranean Sea (Sprovieri et al., 2003). At rural and mountainous sites in the northeastern and southeastern US, typical levels are \( \sim 1.6 \text{ ng m}^{-3} \) (Sigler and Lee, 2006; Valente et al., 2007), while concentrations as high as \( 5.1 \text{ ng m}^{-3} \) are present in urban areas such as in Seoul, South Korea (Kim et al., 2005). Valente et al. (2007) summarized RGM levels over the continent, with values ranging from 0.1% (0.003 ng m⁻³) of TGM at Zeppelin, Norway (Berg and Aspmo, 2003) to \( \sim 6\% \) (0.163 ng m⁻³) at Oak Ridge Reservation in Tennessee (Lindberg and Stratton, 1998). Thus, in most regions the concentration of TGM can be approximated as that of Hg°.

Temporal variability in ambient mercury levels reflects the net effect of chemical and physical sources and sinks. The diurnal cycle of atmospheric mercury exhibits varying characteristics at different geographical locations. For example, measurements from 1 January to 31 December 2003 in St. Anicet (Québec, Canada), a flat grassy rural area surrounded by farms and woods, showed a strong diurnal cycle of Hg° with significantly higher concentrations during daytime than in early morning (Poissant et al., 2005).
This was similar to diurnal cycles observed at other locations in Canada and elsewhere (Kellerhals et al., 2003; Kim et al., 2005), and possibly reflects nighttime depletion of TGM by surface deposition. Quite differently, Lee et al. (1998) reported a nighttime enhancement in mercury near a power plant in Harwell, UK, in all seasons which they ascribed to a buildup of emissions under the nocturnal inversion layer. In summary, quantifying diurnal variability in ambient mercury levels is the first step to understanding its regional budget.

In the global budget of mercury, emissions from natural processes comprise 2100 tons Hg° yr⁻¹ compared to 2400 tons yr⁻¹ from anthropogenic sources (Mason and Sheu, 2002). Removal of Hg° from the atmosphere is facilitated by photochemical transformation to RGM with subsequent wet and dry deposition of Hg°, RGM, and Hg² by with efficiencies dependent on the proximity of sources, the level of oxidants, precipitation patterns, and land cover (Schroeder and Munthe, 1998). Laboratory-based studies have suggested that Hg° is converted to RGM and Hg² by varying rates through reactions with O₃, OH, NO₃, and halogen atoms and oxides (Lin et al., 2006). However, chemical loss of Hg° has not been critically evaluated in the context of regional budgets.

Dry depositional loss of atmospheric Hg° appears to be small based on a small number of nighttime flux measurements (Obrist et al., 2006). Based on these limited measurements, the dry deposition velocity (v_d) is commonly set to values ranging from 0.01 cm s⁻¹ down to zero in models (e.g. Lee et al., 2001; Bergan et al., 1999; Xu et al., 1999; Lindberg et al., 1992; Pai et al., 1997). In the Community Multi-scale Air Quality model, CMAQ-Hg, the dry deposition velocity varied over the range of 0.02–0.1 cm s⁻¹ over the forest and coastal surface in the Northeast US during months of June–September (modeling domain and seasonal average: 0.01±0.02 cm s⁻¹) (Talbot et al., 2006). However, field data from Lindberg et al. (1998) places the deposition velocity at one order of magnitude higher. They suggested that the foliar uptake or dry deposition of gas phase Hg° exhibited both cuticular and stomatal pathways, resulting in dry deposition velocities of 1.3±1.8 cm s⁻¹ to wet and 0.4±0.3 cm s⁻¹ to dry forest canopies.

Long-term measurements with high-time resolution are essential for assessing the regional budget of atmospheric mercury. Ideally, a co-located suite of meteorological and chemical measurements could help identify important sources and sinks of atmospheric mercury. In this study we used multi-year continuous measurements of Hg° from an inland site, a coastal site, and one season of measurements from a marine site to assess regional sources and sinks of Hg° in New England. We also explored the possible impact of inter-annual variability in the large-scale mercury background level on the regional characteristics of Hg°.

2 Measurements and data

Long-term continuous measurements with highest resolution possible were conducted at three UNH AIRMAP (www.airmap.unh.edu) Observatory sites: Thompson Farm (43.11° N, 70.95° W, 24 m), Pac Monadnock (42.86° N, 71.88° W, 700 m), and Appledore Island (42.97° N, 70.62° W, 40 m). Detailed information on the measurements can be found in Mao and Talbot (2004a), Talbot et al. (2005), and Fischer et al. (2007). Since January 2004 at TF, approximately 100 different hydrocarbons, halocarbons, and alkyl nitrates have been measured on a daily basis at 16:00 UT. Specifically, the observations of Hg°, CO, CO₂, O₁, SO₂, NO and NO₃ at TF were used in this study, whereas at PM only measurements of Hg° and CO were available. At both TF and AI hourly measurements of volatile organic compounds (VOCs) were conducted during the International Consortium for Atmospheric Research on Transport and Transformation (ICARTT) field campaign in summer 2004 (Fehsenfeld et al., 2006). Information concerning the instrumentation configuration and measurement details are provided in Sive et al. (2005) and Zhou et al. (2005).

The PM and TF sites are 185 and 25 km inland from the Atlantic Ocean respectively, while AI is 10 km offshore in the Gulf of Maine. The locations of the measurement sites form a unique west-east oriented transect with site surroundings composed of heavily forested, coastal, and marine boundary layer environments. Moreover, due to the remote central location of PM in New England and its 700 m elevation (i.e. above the nocturnal inversion and in the middle of the daytime boundary layer), the site is ideally located to determine regional trends in trace gases, including Hg°.

At each site ambient air is delivered to the instruments through a 10 cm OD PFA Teflon-coated aluminum manifold operated at 1500 standard liters per minute. The inlets to the manifolds are located 15 m above ground level at TF and PM, just above the surrounding canopy, while that at AI is located at the top of a WWII-era surveillance tower about 40 m above sea level. The data were obtained as one minute averages for CO, O₃, CO₂, SO₂, NO, and NO₃. These data were merged to the five minute time resolution of the Hg° measurement. All data are presented in Universal Time (UT) with local time corresponding to UT−5 h for non-daylight saving time intervals and UT−4 h when daylight saving was in effect (i.e. April–October).

Measurements of TGM began at TF on 1 November 2003 and 28 February 2005 at PM, whereas measurements were available at AI only for the period of 8 July–6 September 2005. At all three sites a Tekran model 2537A cold vapor atomic fluorescence spectrometer was used to measure Hg° with a 5-min time resolution and a limit of detection of 5–10 fmol/mol. A teflon filter was used to exclude aerosols. The system employed an internal permeation tube calibration (±5% reproducibility) that was verified every six months using syringe injection from the headspace.
of a thermoelectrically cooled Hg\textsuperscript{o} reservoir (Tekran model 2505). Standard additions of Hg\textsuperscript{o} were done using the internal permeation source of the Tekran. The concentration of these standard additions was 2–3 times higher than ambient mixing ratios. Standard additions of Hg\textsuperscript{o} were performed twice daily on ambient air during day and night to capture variations in temperature and specific humidity. In practice the amount of TGM measured over a several minute time interval essentially represents Hg\textsuperscript{o} unless there is an unusually large amount of RGM present. In most environments, a few minute sampling resolution for RGM is too short to detect levels near the detection limit. Independent measurements of RGM with a Tekran model 1130 show that at all three AIRMAP sites during mid-day in summer 2007 the RGM/TGM percentage rarely exceeded 1% (Sigler et al., 2008\textsuperscript{1}). Thus, our TGM data can be approximated to be Hg\textsuperscript{o} at all three sampling sites, and as such we refer to it as Hg\textsuperscript{o} in this paper. In addition, we report Hg\textsuperscript{o} in fmol/mol, and note that at STP 1 ng m\textsuperscript{-3}=112 fmol/mol.

The precision of the Hg\textsuperscript{o} measurements was assessed by sampling ambient air at ground level using three co-located 2537A instruments at TF. These instruments were intercalibrated using syringe injection from a Tekran 2505 unit prior to the ambient air measurement periods. We found that two brand new instruments agreed within ±4–5\%, but that when a one-year old instrument was included the average increased to ±8–10\%. Based on these comparisons, it appears that a conservative estimate of the overall measurement precision is ±10\%. The accuracy of the Hg\textsuperscript{o} measurements should be around ±5\% due to careful calibration with the headspace injections. Future work will include rigorously re-assessing the overall accuracy of the calibrations when gas-phase calibration standards are available from the US National Institute of Standards and Technology.

3 Seasonal and diurnal variation

Time series of 5-min averaged Hg\textsuperscript{o} mixing ratios are displayed in Fig. 1 at TF, PM, and AI for the entire study period. Mixing ratios of Hg\textsuperscript{o} at TF exhibited a distinct and recurring seasonality with an annual maximum in late winter–early spring and a minimum in early fall. This long-term variation was accompanied by precipitous day-to-day dips and peaks that were accentuated during warm seasons. For example, Hg\textsuperscript{o} mixing ratios in early fall were as low as ∼50 fmol/mol compared to the highest levels exceeding 300 fmol/mol. In comparison, Hg\textsuperscript{o} at the PM inland site exhibited much smaller diurnal-to-annual variations and much higher minimum values. At this site, the daily mixing ratio maximum rarely exceeded 250 fmol/mol while the minimum was typically ∼100 fmol/mol. A comparison of Hg\textsuperscript{o} seasonal


![Fig. 1. Five-minute averaged mixing ratios of Hg\textsuperscript{o} for TF (red) during the period of 1 November 2003 to 31 May 2007, for PM (blue) over 1 June 2005–31 May 2007, and for AI (green) over 8 July–6 September 2005. The smooth lines (TF – dark red, PM – grey, and AI – bright green) are 10-day moving medians.](www.atmos-chem-phys.net/8/1403/2008/)
Fig. 2. Seasonally averaged diurnal cycles with standard deviation of means at TF (a) and PM (b).
the downward trend was much less pronounced, which may be attributable to the presence of a stronger natural source strength or less efficient loss of Hg°, assuming a similar regional anthropogenic contribution at the two sites.

Seasonally averaged diurnal cycles of Hg° at TF exhibited noticeable daily patterns in summer and fall with amplitudes of 20–25 fmol/mol and 15–20 fmol/mol respectively (Fig. 2a). The fluctuations were insignificant in winter and spring with amplitudes <10 fmol/mol. These seasonal characteristics are likely attributable to the presence and frequency of the occurrence of the nocturnal inversion at altitudes below ~300 m in New England (Talbot et al., 2005). During summer and fall the inversion can be present on as many as 50% of the nights, and like O₃, it inhibits near-surface replenishment of Hg° by limiting exchange with remnant boundary layer air aloft. Beneath the inversion, Hg° can be effectively removed from the atmosphere by nighttime sinks (e.g. oxidation and dry deposition), causing the observed nighttime dips in Hg° mixing ratios at TF (Fig. 2a). The daily minimum in Hg° mixing ratios occurred between 10:00–11:00 in summer and ~11:00 in fall, which correspond to sunrise and initiation of vertical mixing driven by diurnal heating (Talbot et al., 2005). We discuss this phenomenon in more detail in Sect. 5.

In contrast, predominant features at PM included the absence of an Hg° diurnal cycle and relatively small seasonal variation in its mixing ratio (160–175 fmol/mol) except for a moderate decrease to ~150 fmol/mol in falls 2005 and 2006 and a more significant one to ~140 fmol/mol in summer 2006. Comparison of Hg° seasonality at PM and TF indicates that mixing ratios at the two sites were similar in winter and spring, but lower at TF in summer and fall, with the largest differences of 20–50 fmol/mol occurring near 12:00 in fall (Fig. 2a, b).

It should be pointed out that at TF there was considerable interannual variability (20–30 fmol/mol) in the seasonally averaged diurnal cycle of Hg° for all seasons but summer (Fig. 2a); this difference was most pronounced during the fall season. Furthermore, the patterns and magnitude of the interannual variability for spring, winter, and fall were different, which may reflect the dominance of varying controls on Hg° levels that potentially differ by season and year.

### 4 Relationship of Hg° with key trace gases

Both Hg° and CO have significant anthropogenic sources and exhibit similar annual cycles (Fig. 3). Thus, it can be insightful to examine the relationship between Hg° and CO. Annually, CO reached its lowest mixing ratios at TF (Fig. 3a) in June–July and remained low over the period July–October, followed by a maximum in mid-January to February. This is consistent with the annual cycle of OH, as variation in CO is driven primarily via oxidation by OH. In comparison, the timing of maxima and minima in Hg° lagged behind that of CO by 2–3 months. This implies that other equally or more important oxidative processes beside reaction with OH and/or possible sources influence the annual temporal variation of Hg°. It should be noted that annual cycles of Hg° and CO at PM resembled those at TF with less variability (Fig. 3b) due to its more rural and higher elevation location.

Identification of a positive correlation between a trace gas of interest and CO, a general tracer of combustion emissions, is an effective way to demonstrate the important contribution of anthropogenic sources to the atmospheric mixing ratio of

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**Table 1. Summary statistics (N – number of data points, 10th – the 10th percentile value, Med – the median value, 90th – the 90th percentile value, Mean – the mean value, σ – standard deviation. All measures are in unit of fmol/mol, except N) of Hg° mixing ratios at TF and PM for all seasons.**

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**Atmos. Chem. Phys., 8, 1403–1421, 2008**
a trace gas (Parrish et al., 1993). To minimize confounding issues with natural sources of Hg\(^{\circ}\), we examined the wintertime relationship between Hg\(^{\circ}\) and CO at TF and PM and found good linear correlations, indicating that the primary source of Hg\(^{\circ}\) in winter at our sites was anthropogenic emissions. The 2006 Hg\(^{\circ}\)-CO relationship at PM exhibited a relatively tight linear correlation with \(r^2=0.53\) (\(p=0.01\)) and a slope of 0.23±0.003 fmol/nmol (\(p=0.05\)), while the correlation in winter 2007 yielded \(r^2=0.34\) (\(p=0.01\)) and a slope of 0.21±0.006 fmol/nmol (\(p=0.05\)) (Fig. 4). In atmospheric mercury budgets, it is commonly assumed that the principal anthropogenic source of mercury is emissions from coal-fired power plants and waste incineration (Pacyna and Pacyna, 2002; Seigneur et al., 2004). A linear Hg\(^{\circ}\)-CO relationship indicates that the two trace gases have long enough lifetimes to become well mixed in air masses on a regional scale. More specifically, their mixing ratios represent contributions of all anthropogenic emissions impacting the regional air shed superimposed on the background Hg\(^{\circ}\) level. As mentioned in Sect. 2, the PM site is particularly well suited to determine the regional source characteristics of Hg\(^{\circ}\). Nonetheless, because PM is situated 1–3 transport days downwind of the polluted northeast corridor and the Mid-West, it is challenging to delineate the contribution from different sources to the ambient level of Hg\(^{\circ}\) in New England.

Applying the Hg\(^{\circ}\)-CO relationship, we deduced the regional background level of Hg\(^{\circ}\) in air masses that affect New England. First, the lowest 10th percentile of CO mixing ratios were 164 nmol/mol and 154 nmol/mol at PM in winters 2006 and 2007 respectively, which we defined as the regional background level. The background levels of Hg\(^{\circ}\) were then estimated to be 154 fmol/mol and 165 fmol/mol in winters 2006 and 2007 respectively, which are close to the seasonally averaged Hg\(^{\circ}\) mixing ratios of 163 fmol/mol and 172 fmol/mol in winters 2006 and 2007 (Table 1).

A striking difference between the two sites was the factor of 2 lower slope values for Hg\(^{\circ}\)-CO and Hg\(^{\circ}\)-NO\(_2\) at TF compared to PM. Interpretation of these relationships at TF is not straightforward owing to local sources and more variable transport to the site (Mao and Talbot, 2004b), with \(r^2\) values for Hg\(^{\circ}\)-CO ranging from 0.15 in 2005 to 0.31 in 2004 during the four winters (Fig. 5) (all with \(p=0.05\) except 2005). Note that the slope values exhibited strong interannual variability, and increased by more than a factor of 2 from 0.06±0.001 fmol/nmol (\(p=0.05\)) in winter 2004 to 0.14±0.004 fmol/nmol (\(p=0.05\)) in winter 2006. Furthermore, the maximum mixing ratio of CO in winters 2006 and 2007 did not exceed 600 nmol/mol, whereas in winter 2003 CO peaked at close to 1000 nmol/mol. Lower maximum levels were also observed in CO\(_2\) and NO\(_x\) at TF in winters 2006 and 2007 compared to previous years, indicating that less polluted air masses arrived at the site. The Hg\(^{\circ}\)-NO\(_y\) relationship observed at TF in winter had large interannual variability in the slope values (Fig. 6). Similar to the Hg\(^{\circ}\)-CO correlation, the slope value increased by more than a factor of 2 from 0.50±0.01 fmol/nmol (\(p=0.05\)) in winter 2004 to 1.22±0.02 fmol/nmol (\(p=0.05\)) in winter 2006. The mechanisms driving this phenomenon are explored in Sect. 7.2.

The relationships of Hg\(^{\circ}\) with C\(_2\)Cl\(_4\), C\(_2\)H\(_3\)Cl, O\(_3\), CO\(_2\), SO\(_2\), and CH\(_4\) were investigated at TF. Our analysis showed that Hg\(^{\circ}\) was not correlated with C\(_2\)Cl\(_4\), C\(_2\)H\(_3\)Cl, O\(_3\), and SO\(_2\) (not shown). It is reasonable not to observe relationships between Hg\(^{\circ}\) and other gaseous species due to their different sources and lifetimes. Gaseous elemental mercury is primarily emitted from combustion sources and has a lifetime of 0.5–2 years (Schroeder and Munthe, 1998), while C\(_2\)Cl\(_4\) and C\(_2\)H\(_3\)Cl, used as degreasing agents in industrial applications, have lifetimes of a couple of months (Kindler et al., 1995; Olaguer, 2002; NOAA GMD, 2000) and 4–7 days (Kindler et al., 1995), respectively.

Because the main source of SO\(_2\) is coal-fired power plants, whose emissions may also contribute significantly to atmospheric mercury, it has been used as a mercury tracer in previous studies. For example, Friedli et al. (2004) found that Hg\(^{\circ}\) was well correlated with CO and SO\(_2\) in plumes sampled over the western North Pacific during the ACE-Asia airborne field campaign. However, our long-term data set at TF shows no obvious correlation between Hg\(^{\circ}\) and SO\(_2\)
in all seasons except winter when there was a diffuse positive correlation ($r^2 < 0.2$) in all four winters (2004–2007, not shown). This result most likely reflects the increased lifetime of SO$_2$ in winter, which persists over extended transport times and retains the source signature longer than in the warm season. In all other seasons, the largest mixing ratios of SO$_2$ corresponded to low-to-moderate levels of Hg$^\circ$ (100–200 fmol/mol) with no identifiable relationship. Overall, TF
may be too distant from significant coal-fired combustion sources in the eastern and midwestern US to define a consistent relationship between Hg° and SO2.

There is an excellent positive linear correlation between greenhouse gases and combustion tracers in New England, indicating a significant anthropogenic impact on regional air quality (Shipham et al., 1998a, b). Since Hg° was correlated reasonably well with CO in winter, we examined its relationships with CO2 and CH4 during winter when natural source/sink strengths are minimal. As expected, the correlation between Hg° and CO2 (not shown; r2=0.3–0.5, slope = 1.6–1.8 fmol/mmol) was similar to that of Hg°-CO. The correlation between Hg° and CH4 was poor, likely related to weaker CH4 emissions compared to CO2 from combustion sources.

5 Nighttime depletion and dry depositional losses

Time series of O3 and Hg° mixing ratios for August 2004, 2005, and 2006 show synchronized depletion in the early morning hours (Fig. 7). The nighttime dips in Hg° were most pronounced when the O3 mixing ratio approached zero, presumably nights with strong nocturnal inversions. The occurrence of O3 depletion results from titration by NO and dry deposition, combined with negligible vertical exchange with residual boundary layer air aloft (Talbot et al., 2005). In contrast, a continuous enhancement in CO2 mixing ratios to 500–600 mmol/mol occurs until sunrise (not shown), driven by emissions from ecosystem respiration and local combustion processes (Talbot et al., 2005). The coincident depletion of Hg° and O3, and enhancement in CO2 provides strong evidence for a net surface sink of Hg°.

Sinks of Hg° are in general considered to be chemical oxidation and dry deposition. Total chemical loss of Hg° through reaction with O3, OH, and NO3 was estimated using typical mixing ratios at TF and published rate constants. These estimates were in very close agreement for the 2004 and 2005 warm seasons in spite of year-to-year differences in mixing ratios of Hg°. The reaction of Hg° with NO3 can be a significant nighttime loss pathway for Hg° (Sommar et al., 1997). Using kNO3=4×10−15 cm3 molecule−1 s−1 (Sommar et al., 1997), a NO3 mixing ratio = 10 pmol/mol (Brown et al., 2004; Ambrose et al., 2007) and Hg°=100 fmol/mol, the loss rate of Hg° would be 0.35 fmol/mol h−1 or 3.5 fmol/mol d−1 for 10 h of darkness.

Fig. 6. Five-minute averaged Hg° vs. NOy mixing ratios at TF during winters 2004 (a), 2005 (b), 2006 (c), and 2007 (d).
Published rate constants for reaction of Hg\(°\) with O\(_3\) vary greatly, yielding significantly different estimates of Hg\(°\) loss. The lower limit of the rate constant is \(3 \times 10^{-20}\) molecule\(^{-1}\) cm\(^3\) s\(^{-1}\) (Hall, 1995), while the upper limit is \(7.5 \times 10^{-19}\) molecule\(^{-1}\) cm\(^3\) s\(^{-1}\) (Calvert and Lindberg, 2005). The seasonally averaged loss rates from 2004, 2005, and 2006 were very similar, differing only by a few percent, and thus the averaged loss rates were listed in Table 2. Using the upper limit, total daily loss of Hg\(°\) during the 2004 warm season at TF varied from \(\sim 12\) fmol/mol d\(^{-1}\) in April to \(\sim 5\) fmol/mol d\(^{-1}\) in fall (Fig. 8a). Results are comparable for 2005 and 2006; to avoid redundancy only values for 2005 are shown in Fig. 8b. The estimated Hg\(°\) loss resulting from oxidation by OH (OH concentration adopted from Spivakovsky et al., 2000; rate constant from Sommar et al., 2001) was negligible compared to that of O\(_3\). Note that on average, day- and nighttime total loss (O\(_3\)+OH+NO\(_3\)) (4.7 and 6.1 fmol/mol d\(^{-1}\), respectively) were of equal importance, while loss of 2.6 fmol/mol d\(^{-1}\) by reaction with O\(_3\) at night was comparable to the 3.5 fmol/mol d\(^{-1}\) by NO\(_3\) (Table 2).

Calvert and Lindberg (2005) cautioned that significant loss of Hg\(°\) by the Hg\(°\)-O\(_3\) reaction is not likely to occur in the

![Fig. 7. Time series of 5-min averaged Hg\(°\) (black) and O\(_3\) (red) during the time period of 1 August–30 September for 2004 (a), 2005 (b), and 2006 (c).](image-url)
atmosphere. Using the lower limit of the rate constant, we found that the Hg° loss via oxidation by O₃ was smaller than that via OH oxidation (0.3–0.7 fmol/mol d⁻¹), decreasing from 0.4 fmol/mol d⁻¹ in April to 0.1 fmol/mol d⁻¹ in fall (Fig. 8c–d). Overall, the total chemical loss of Hg° via oxidation by OH and O₃ ranged from ~1 fmol/mol d⁻¹ in April to 0.4 fmol/mol d⁻¹ in fall. Table 2 shows that on average, the lower limit of daily total chemical loss (O₃+OH+NO₃) was 4.3 fmol/mol d⁻¹, and the nighttime loss via reaction with NO₃ comprised more than 80% of its removal.

The seasonally averaged diurnal cycles of Hg° in summer and fall showed ~20 fmol/mol of nighttime Hg° loss (Fig. 2a). Dry deposition of Hg° has been treated as a minor loss mechanism in previous studies, but our observations and analysis suggest otherwise. To quantify the lower limit of dry depositional loss, we used the upper limit attributed to chemical loss using the fastest rate constant for the Hg°-O₃ reaction. Hence, nighttime chemical loss of Hg°, ~6 fmol/mol d⁻¹, comprised 30% of the 20 fmol/mol seasonal average nighttime decrease (Fig. 2a).

We propose that removal by dry deposition was the main contributor to the remaining 70% of the nighttime loss of Hg°. The loss of Hg° via dry deposition can be better inferred from its relationship with NOₓ, in particular, by inference to HNO₃ and NOₓ. A scatter plot of Hg° versus NOₓ (Fig. 9a), using all 3.5 years of data from TF, exhibited a well-formed correlation envelope along the lower boundary of the relationship. This boundary was comprised of the lowest Hg° values and NOₓ ranging from near zero to 140 nmol/mol (i.e. data colored red in Fig. 9a, denoted as lower Hg°-NOₓ boundary hereinafter). The data points on this boundary for NOₓ mixing ratios >40 nmol/mol were mostly collected in winter (December–February) and resided on the high end of NOₓ distribution at TF (Fig. 6a and b). The data points on this boundary for NOₓ <40 nmol/mol mostly occurred before sunrise in fall to early December (Fig. 9b), and exhibited a linear function that was represented by

\[ \text{Hg}^\circ = 48 + 3.2 \times \text{NO}_x. \]  

Diurnal cycles of Hg°, NOₓ, and O₃ were averaged over the 17, 44, and 24 days in 2004, 2005, and 2006, respectively, that included the data points for NOₓ <40 nmol/mol on the lower Hg°-NOₓ boundary (Fig. 10). On these days, O₃ mixing ratios decreased quickly from 15–20 nmol/mol at 00:00 to single digits in <4 h and subsequently decreased at a much slower rate to ≤5 nmol/mol in the hours preceding sunrise. NOₓ and Hg° fell at more constant rates over the same time period, implying a common loss mechanism. The differing nighttime depletion rates of NOₓ and Hg° from that of O₃ may stem from coexisting sources for NOₓ and Hg°.
as opposed to no nighttime O$_3$ production but loss from NO titration and dry deposition. Therefore, NO$_y$ appears to be a better proxy than O$_3$ to understand nighttime loss mechanisms of Hg$^\circ$.

Nighttime loss of NO$_y$ is mainly via the dry deposition of HNO$_3$ and NO$_x$. Nighttime loss of HNO$_3$ is on average $\sim$1 nmol/mol in this area based on ICARTT measurements during summer 2004 (Fischer et al., 2006). An additional 5 nmol/mol of NO$_y$ is lost at night through interconversion of NO$_y$-NO$_3$-N$_2$O$_5$ and ultimate removal of N$_2$O$_5$ based on measurement and model findings for the ICARTT period (Ambrose et al., 2007). As a result, dry depositional loss of (HNO$_3$+NO$_x$) is fast at night, and thus the close resemblance of decreasing nighttime trends of NO$_y$ and Hg$^\circ$ equates to more rapid dry deposition of Hg$^\circ$ than suggested in previous studies. Using the slope of 3.2 fmol/nmol derived in Eq. (1) and 6 nmol/mol loss of NO$_y$ we obtained $\sim$20 fmol/mol seasonal average value suggested in Fig. 2a, which represents the combined result of anthropogenic sources and removal processes (oxidation and dry deposition) at night.

![Fig. 9](image)

Fig. 9. (a) All data points of 5-min averaged Hg$^\circ$ vs. NO$_y$ over 1 November 2003–31 May 2007 with the lower boundary of the relationship highlighted in red; (b) The time of occurrence of the data points that comprised the lower boundary in (a).

![Fig. 10](image)

Fig. 10. Averaged diurnal cycles of Hg$^\circ$ (black), NO$_y$ (red), and O$_3$ (blue) for the days (17, 44, and 24 days in 2004, 2005, and 2006, respectively) that include the data points with NO$_y$<40 nmol/mol on the lower boundary of Hg$^\circ$ vs. NO$_y$ presented in Fig. 9a for 2004 (a), 2005 (b), and 2006 (c).

However, we estimate a total average Hg$^\circ$ daily loss (oxidation + deposition) of $\sim$26 fmol/mol, suggesting a source of $\sim$6 fmol/mol, likely from local anthropogenic emissions. The total annual emission of mercury in Strafford County, NH (1000 km$^2$), where TF is located, is $\sim$6850 g yr$^{-1}$ (NESCAUM, 2005a, b), which yields $\sim$9 fmol/mol assuming a nocturnal inversion layer thickness of 125 m (Talbot et al., 2005). This is of similar magnitude as the $\sim$6 fmol/mol source needed to close the budget.

To minimize the source and sink terms in the budget calculation, we consider nights with the occurrence of a nocturnal inversion layer, where the nighttime decrease in Hg$^\circ$ is determined by the net effect of chemical loss, dry deposition, and anthropogenic sources. Using a surface depletion zone of 125 m (Talbot et al., 2005), we found that the dry deposition velocity of Hg$^\circ$ was 0.17 cm s$^{-1}$ in 2004, 0.20 cm s$^{-1}$ in
2005, and 0.18 cm s⁻¹ in 2006. These values represent nighttime deposition rates, and, similar to O₃ at a forested site in central Massachusetts (Munger et al., 1998), are likely increased during daytime due to open plant stomata and higher wind speeds.

An estimate of the warm season Hg° lifetime was made using the formula from Jacob (1999). Based on ν_d=0.2 cm s⁻¹, an average Hg° mixing ratio of 150 fmol/mol, a seasonally averaged chemical loss rate of 0.5 fmol/mol d⁻¹ (using the lower limit), and a planetary boundary layer (PBL) height=2000 m, we obtain an atmospheric lifetime of ~16 days. The lifetime estimated here is an order of magnitude smaller compared to 160 days for ν_d=0.01 cm s⁻¹, a commonly used value in regional and global models. Our analysis indicates that deposition processes for Hg° are in fact very significant, and the lifetime of Hg° in the local planetary boundary layer at TF is short.

6 Oceanic influence

In Sect. 3 we showed that the rate of decrease in Hg° mixing ratios during the warm season was typically 0.5–0.6 fmol/mol d⁻¹ at TF compared to 0.2–0.3 fmol/mol d⁻¹ at PM, which resulted in a cumulative difference of 55 fmol/mol between the two sites over the period 1 April to 30 September. Global budget estimates for Hg° indicate that the ocean is its largest natural source (Mason and Sheu, 2002) and is also a large reservoir of organo-halogen trace gases. This raises the possibility of an important oceanic influence on the ambient level of Hg° at TF, both as a direct source of Hg° and enhancement of Hg° lost from oxidation by halogen atoms and oxides.

Instantaneous variability in the ambient level of Hg° can be obscured if the magnitude of the change is small compared to the ambient mixing ratio and the impact is short-lived (i.e. the measurement resolution of tracer compounds is not sufficient to capture the events). Moreover, TF is surrounded by extensive vegetation, located near the ocean, as well as being downwind of anthropogenic sources in the favorable southwest flow regime, resulting in complex and well-mixed contributions from a multitude of natural and anthropogenic sources. Consequently, clear correlations between the mixing ratios of Hg° and marine tracers (e.g. CHBr₃, CH₂Br₂, and CH₂ClI) were rarely observed in the overall dataset.

To accentuate the instantaneous variation, we defined “anomalies”, which were calculated for Hg° and three marine tracers, CHBr₃, CH₂Br₂, and CH₂ClI by subtracting the diurnal and seasonal means from the measured mixing ratios over the entire ICARTT period (1 July–16 August 2004). The means can be considered to be the baseline for the time period of interest that is determined by processes operating on time scales of days or longer. The anomaly, in essence, represents the effects of instantaneous events such as in situ chemical oxidation, emissions, and dry deposition.

![Fig. 11. Anomalies of Hg° versus CHBr₃ at TF (a) during summers 2004 and 2005 and at AI (b) in summer 2005. Anomalies were calculated by subtracting the hourly averages for the entire campaign period from the measured high resolution mixing ratios.](Image)
were concurrent hourly VOC data. The data from the time window of 02:00–09:00 was selected when oxidation of Hg° by halogen compounds should be minimal, and consequently the effect of oceanic emissions should be apparent. Again, the criterion of CO < 200 nmol/mol was applied to minimize the impact of anthropogenic sources. A positive correlation between the anomalies of Hg° and CHBr₃ was found to be better defined (r²=0.21) (Fig. 11b) than at TF (r²=0.11). This finding illustrates the oceanic source of Hg° in the marine boundary layer (MBL) and supports its potential influence on adjacent coastal areas such as at TF.

Oceanic emissions of organo-halogen species can cause loss of Hg° through oxidation reactions, such as with bromine (Br) atoms as described by Holmes et al. (2006). These mechanisms facilitate conversion of Hg° to RGM with subsequent rapid depositional loss as observed in the Arctic (Ariya et al., 2004). In the MBL with [Br]=0.5 pmol/mol (Dickerson et al., 1999), [Hg°]=100 pmol/mol, and k=3.2×10⁻¹² cm³ molecule⁻¹ s⁻¹ (Ariya et al., 2002), the production rate of RGM should be ∼14 fmol/mol h⁻¹. However, the reported concentrations of RGM in coastal and oceanic regions range from 6±6 pg m⁻³ (0.8 fmol/mol) in the North Pacific (Laurier et al., 2003) to 50±43 pg m⁻³ (6 fmol/mol) in Bermuda (Sheu, 2001). The disparity between the estimated production rate and measured concentrations suggests that highly efficient RGM loss pathways exist in the MBL.

TF, being only 25 km away from the ocean, is under marine influence year-round, and is impacted by air masses directly from the ocean 30% of the time in summer (Chen et al., 2007). Thus, the potential loss of Hg° via halogen chemistry cannot be ignored in the regional budget of Hg°. We hypothesize that the overall significantly lower Hg° levels and steeper decreasing trend in Hg° mixing ratios during the warm season at TF compared to those at PM reflects an important impact of halogen chemistry on Hg°.

7 Contrasting interannual variability in Hg° between warm and cold seasons

7.1 Warm season

As shown in Sect. 3, Hg° mixing ratios at TF decreased progressively during the warm season. The magnitude of this decrease was similar during the warm season in 2005 and 2006 at a rate of 0.6 fmol/mol d⁻¹, steeper than the rate of 0.3 fmol/mol d⁻¹ in 2004 (Fig. 12). Therefore, in this section, we focus on the change from 2004 to 2005 to identify possible causal mechanisms.

We proposed in Sect. 5 that dry deposition was the principal mechanism leading to decreased nighttime mixing ratios of Hg°. However, our calculations indicated that the downward seasonal trend cannot be attributed to dry depositional loss of Hg°, based on the little difference in depositional loss between the two years in spite of more frequent occurrences of nighttime Hg° depletion in 2005. To identify possible causes, we conducted a spectral analysis of the Hg° data over the time period April–September in 2004 and 2005. In both years the dominant periodicity was the diurnal cycle. However, >70% of the total energy was distributed over time scales of 2–30 days. Furthermore, we applied the Kolmogorov-Zurbenko (KZ) filter (Zurbenko, 1986; Hogrefe et al., 2000) to separate the diurnal (<1 day), synoptic (weekly), and long-term (>weekly) time scales. This revealed that in 2005, processes on time scales >weekly contributed 71% to the total variance compared to 32% in 2004. Thus, long-term processes might account for the more pronounced decreasing trend in 2005.

Processes on time scales >weekly can include chemical loss of Hg° through reactions with O₃ and OH, which are slower than ones with organo-halogen compounds, and variations in natural emissions which are influenced by climate. Our calculations revealed no distinct difference in seasonal trends in chemical oxidation by O₃ and OH during the two warm seasons (Fig. 10). Hydrocarbon measurements at TF from the two years showed similar levels of marine-derived halogenated trace gases such as CHBr₃ (Zhou et al., 2008), seemingly removing differential influences from halogen chemistry as the cause of year-to-year variability in Hg°.

Biomass burning can reportedly contribute to the ambient level of Hg° (e.g. Brunke et al., 2001; Friedli et al., 2001; Sigler and Lee, 2003). Data from Canada (http://nfdp.ccfm.org) shows that total area burned in Quebec and Ontario were much higher in 2005 than in 2004. In 2004, there was more precipitation and fewer hectares burned in Québec and Ontario, where we know there are fires that can influence the northeast US (Sigler and Lee, 2003; DeBell
et al., 2004). Hence, should emissions of biomass burning be the dominant effect, one would expect lower levels of Hg\textsuperscript{g}, not higher, during the warm season in 2004 than in 2005. However, the steeper decreasing trend and the much lower annual minimum value in the 2005 warming season contradicts this piece of information. The US national data (http://www.nifc.gov/fire_info/fires_acres.htm) shows no significant difference in total burned area between 2004 and 2005 nationwide (and most of the burning takes place out west and, especially, in Alaska). Therefore, the impact of emissions from biomass burning was ruled out on the variation in Hg\textsuperscript{g} levels from the warm season of 2004 to that of 2005.

Finally, a suite of climate variables was examined for the two year period, and no significant difference in atmospheric circulation patterns (e.g. sea level pressure, SLP), temperature, moisture, radiation flux, or wind speed/direction was identified. However, precipitation during the period of 1 April–30 September in 2004 was 173 mm more than over the same period in 2005. Documented effects of precipitation on Hg\textsuperscript{g} levels are threefold: (1) a short-lived (~hours) burst in emissions from soils usually occurs (Lindberg et al., 1999), (2) the dry deposition flux to a wet canopy can be enhanced up to 3-fold compared to a dry one (Lindberg et al., 1998) and, (3) reduced uptake/emission as forest stomatal exchange decreases strongly under dark conditions and drought stress (Lindberg et al., 1998). Recently, Bash and Miller (2008) found, based on flux measurements, that mercury evasion was greatest during periods when the canopy was wet from either dew or rainfall. This result suggests that depositional losses are more than offset by release of Hg\textsuperscript{g} from the canopy, possibly due to lack of RGM which is more soluble than Hg\textsuperscript{g}, or enhanced mobility and reduction of divalent mercury bound in soils or the canopy under wet conditions. It is possible that the dry conditions in summer 2005 may have contributed to the stronger decreasing trend that year. Measurements of RGM mixing ratios and both soil and canopy Hg\textsuperscript{g} flux information are needed to conduct a robust assessment of these processes.

7.2 Wintertime Hg\textsuperscript{g}

In contrast to the large year-to-year variability in the warm season Hg\textsuperscript{g} levels, relatively small changes were observed in maximum Hg\textsuperscript{g} levels from winters 2004 and 2005 to winters 2006 and 2007, when significant decreases in the maximum mixing ratios of CO, CO\textsubscript{2}, NO\textsubscript{y}, and SO\textsubscript{2} were captured by our measurements. The Hg\textsuperscript{g} level varied slightly, while the maximum NO\textsubscript{y} mixing ratio was lowered from 140 nmol/mol in 2004 and 2005 to 40 nmol/mol in 2006 and 2007 (Fig. 6) and CO from near 1000 nmol/mol to almost 400 nmol/mol (Fig. 5). This downward shift changed the relationship of Hg\textsuperscript{g} with other trace gases via variation in the slope values of their linear correlations. Contributing factors to such changes can include fluctuations in anthropogenic emissions, climate, and/or the chemical environment. The chemical processes affecting the five trace gases are different and complex, and it is difficult to obtain year-to-year mercury emission inventories. Therefore, we examined the dominant climate pattern of the four winters as the first step to understanding interannual variability in Hg\textsuperscript{g}.

We found that the 2004–2007 winters in the northeastern US were characterized by a transition from cold to near normal conditions in 2004 and 2005 to significantly warmer than normal conditions in 2006 and 2007. Normal conditions are defined by the Northeast Regional Climate Center as the 30-year means over 1971–2000 (http://www.nrcc.cornell.edu/climate/Climate_summary.html). The average temperature in the northeastern US in 2004 was reportedly 1.1\degree C colder than normal while in 2005 it was near normal at just 0.2\degree C above the 30-year mean. In contrast, 2006 was 1.5\degree C warmer than normal and ranked as the 15th warmest winter on record since 1896, while 2007 was 1.0\degree C warmer than normal.

The spatial distribution of SLP in the northeastern US indicated that the cooler years of 2004 and 2005 were characterized by greater meridional flow, in stark contrast to more zonal flow in the warmer winters of 2006 and 2007 (Fig. 13). The greater meridional flow in 2004 and 2005 facilitated transport of colder air from the Polar Regions to the Northeast, which explains the colder climate in winters 2004 and 2005.

To understand the potential impact of such climate interannual variability on the redistribution of air pollutants, we examined the potential source regions corresponding to all the data points with NO\textsubscript{y}>40 nmol/mol, which indicate heavily polluted air masses, over the entire study time period at TF (1 December 2003–31 May 2007). There were 21 episodes during the four winters from the entire dataset (i.e. an episode is defined as a time period with continuous NO\textsubscript{y}>40 nmol/mol). Five-day backward trajectories were simulated using HYSPLIT (Draxler, 1999) every 5–6 h during each episode starting at altitudes of 500 m and 1000 m from TF, which yield 55 trajectories in total. Fifteen of the 55 trajectories stayed over the North American continent with the rest (40) originating from as far north as 60\degree N which followed the paths of either direct northerly flow, or northerly flow which traversed midwestern and southern states before veering northward along the eastern US and arriving at TF (Fig. 14). Thirteen of the 21 episodes occurred in winters 2004 and 2005 and only 8 in winters 2006 and 2007. This is consistent with the circulation patterns shown in Fig. 13, where the northerly flow on the front of the Canadian high and the back of the Labrador Low brought air masses from higher latitudes to the southeastern US before turning northerward. Such flow regimes facilitate the transport Arctic air masses directly southward to midlatitudes accompanied by entrainment of pollutants from the densely populated eastern US.

Numerous studies indicate that wintertime Arctic air can be heavily polluted by long-range transport from Eurasia and
Fig. 13. Seasonally averaged sea level pressure for winters 2004 (a), 2005 (b), 2006 (c) and 2007 (d).

Fig. 14. Five-day backward trajectories using HYSPLIT for the 21 episodes of NO$_y$ > 40 nmol/mol in winters 2004–2007 starting at two heights, 500 m (a) an 1000 m (b), at TF.
North America (Wang et al., 2003; Stohl, 2006). Lamarque and Hess (2003) found that during the winter, CO in the lower troposphere at high latitudes was strongly influenced by fresh emissions from Europe, and also that aged emissions appeared to build up due to minimal photooxidation in the dark. Hence, it is reasonable to expect that Arctic air masses transported via meridional flow contribute to increased air pollutants at midlatitudes (Mao and Talbot, 2004c). However, such enhancements can easily be obscured by entrainment of fresh pollutants emitted in the eastern US. For instance, at TF there was a well-defined positive correlation between the NO$_2$ level >40 nmol/mol and NO mixing ratios. For the majority data with NO$_2$ levels >40 nmol/mol, NO comprised 40%–90% of NO$_y$ (Fig. 15). The lifetime of NO in the winter troposphere is ~1.5 days (Munger et al., 1998), and thus such high levels of NO most likely came from sources in the eastern US.

Surprisingly, mixing ratios of Hg$^\circ$ at TF rose only slightly during the 21 wintertime episodes. In the wintertime Arctic, Hg$^\circ$ concentration was found to be 1.84 ng m$^{-3}$ (~200 fmol/mol) by Schroeder et al. (1998), or more than 20 fmol/mol higher than the seasonal median values of 160–180 fmol/mol at TF (Table 1). This suggests that Hg$^\circ$ at TF would be elevated, albeit perhaps not significantly, under the direct influence of Arctic air. Furthermore, it would be anticipated that the Arctic air mass in winters 2004 and 2005 should have entrained additional mercury during transport over the US. The observed small increases in Hg$^\circ$ under such circumstances implies: 1.) a homogeneous distribution of surface Hg$^\circ$ levels over the North American continent in winter, and/or 2.) a relatively smaller impact of the eastern US anthropogenic mercury emissions due to RGM comprising a significant portion of the total mercury emissions (e.g. ~34% of the total mercury emission from coal fire plants averaged in the US, Sullivan et al., 2006).

The second implication can also be used to explain the lower slope values in the two heavily polluted winters of 2004 and 2005, when changes in Hg$^\circ$ and CO should be dominated by their anthropogenic emissions. Emitted CO is mostly retained in the atmosphere during winter due to its long lifetime, while RGM is likely removed quickly after entering the atmosphere. Consequently, the ultimate anthropogenic contribution of mercury is likely smaller than that of CO, leading to smaller Hg$^\circ$-CO slope values in those winters. For the winters 2006 and 2007, the larger slope value in the Arctic air mass transported to TF directly from the north was most likely caused by two factors. First, there was minimal entrainment of fresh anthropogenic emissions. Second, decreases in Hg$^\circ$ due to loss of dry deposition and dilution with ambient air probably exceeded reductions in CO that would have occurred by dilution and minimal chemical loss. Therefore, we suggest that the wintertime Hg$^\circ$-CO relationship should be used cautiously to estimate the anthropogenic contribution to ambient Hg$^\circ$.

8 Summary

A comprehensive analysis was conducted using multiple years of Hg$^\circ$ measurements at two inland sites, TF and PM, and one summer of measurements at a marine site, AI, from the UNH AIRMAP observing network in southern New Hampshire. A large suite of other trace gases, e.g. CO, CO$_2$, CH$_4$, NO$_y$, NO, SO$_2$, and VOCs, were employed to identify possible sources of Hg$^\circ$ in New England. The measurements of Hg$^\circ$ at TF showed distinct seasonality with annual maxima in late winter – early spring and minima in early fall, with large day-to-day variation. A decreasing trend in the mixing ratio of Hg$^\circ$ over the time period of March–September occurred at a rate of 0.5–0.6 fmol/mol d$^{-1}$ for all years except 2004 (0.3 fmol/mol d$^{-1}$). Measurements of Hg$^\circ$ at PM exhibited much smaller daily and annual variation, particularly reflected in the slower warm season decline (relative to TF) of 0.2 and 0.3 fmol/mol d$^{-1}$ in 2005 and 2006, respectively. The AI data appeared to track the variation observed at TF albeit with much higher minima.

Hg$^\circ$ was correlated most strongly with CO and NO$_y$ in winter. The Hg$^\circ$-CO relationship at PM suggested that the primary source of Hg$^\circ$ in winter is anthropogenic emissions. Applying the Hg$^\circ$-CO relationship, we found that the seasonally averaged Hg$^\circ$ mixing ratio of ~160 fmol/mol at PM can be considered the regional background level. The positive Hg$^\circ$-NO$_y$ correlation along the lower boundary of all data points indicated dry deposition as a stronger sink for Hg$^\circ$ than suggested by previous studies. We estimated a dry deposition velocity of 0.17–0.20 cm s$^{-1}$ for Hg$^\circ$, and a lifetime of ~16 days in the local boundary layer at TF.

It should be noted that a weak but positive correlation was found between Hg$^\circ$ and CHBr$_3$ at both TF and AI, which suggests a role of the oceanic source influencing the ambient
levels of $\text{Hg}^\circ$ in the marine and coastal environments. It was also hypothesized that the overall significantly lower $\text{Hg}^\circ$ levels and steeper decreasing trend during the warm season at TF compared to those at PM may reflect the impact of marine halogen chemistry.

Contrasting interannual variability in $\text{Hg}^\circ$ levels was identified between the warm and cold seasons. The stronger decline in warm season $\text{Hg}^\circ$ during 2005 and 2006 compared to 2004 may indicate that changes in precipitation played a role in mitigating evasion from the surface. In contrast, the $\text{Hg}^\circ$ level showed relative small changes from winters 2004 and 2005 to winters 2006 and 2007 while the mixing ratios of CO, CO$_2$, NO$_x$, and SO$_2$ decreased dramatically. Our measurements suggested that the predominant meridional flows in colder winters transported the polluted Arctic air mass across the US and then veered northward to TF while entraining fresh emissions, which resulted in high levels of anthropogenic tracers. However, little variation in $\text{Hg}^\circ$ possibly indicates a homogeneous distribution of surface $\text{Hg}^\circ$ in winter and/or rapid removal of RGM released from anthropogenic sources. During warmer winters the $\text{Hg}^\circ$-CO slope value possibly reflects the ratio of $\text{Hg}^\circ$ loss relative to changes in CO more than their emission ratios.

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