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Synthesis and structure of a heterocyclic ansa pyrrole amino acid

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Abstract

We report a synthetic route to ansa pyrrole amino acids via olefin ring-closing metathesis of diene precursors in the presence of Grubbs I catalyst. The dienes were prepared by Grignard addition to pyrrole sulfinyl imines. The success of the macrocyclic ring closure depends on the dienes structure and only in the case of the 13-membered compound **28** sufficient material could be isolated by preparative HPLC separation to investigate its structure spectroscopically. As also rationalized by our computations at the B3LYP/6-311+G(d,p)//B3LYP/6-31G(d) level of theory, **28** is configurationally stable.

1. Introduction

Planar chirality arises from stereogenicity resulting from the arrangement of out-of-plane groups with respect to a reference plane.¹ Structural studies of molecules possessing elements of planar chirality started in the 1940s with the synthesis and resolution of 1,12-dioxo[12]paracyclophane as the first chiral ‘ansa’ compound.² Lüttringhaus was the first who realized and correctly predicted the existence of enantiomeric ansa compounds.³

Compounds displaying planar chirality are found among η^n -olefin metal⁴ and η^n -arene metal complexes,⁵ cyclophanes⁶ and ansa compounds.⁷ Although many of such planar-chiral molecules have been synthesized, only one example of an amino acid with planar chirality exists, at least to the best of our knowledge. Pelter et al. prepared homochiral amino acids, where the chirality depends solely on the chirality of the 4-amino-13-carboxy[2.2]paracyclophane.⁸ However, MOPAC computations of simple peptides derived from this amino

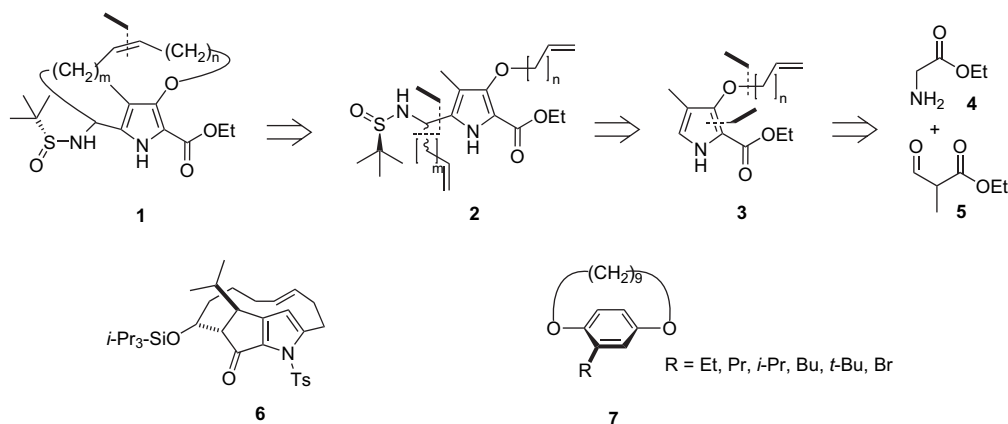
acid showed that the chain distorts due to the close proximity of the amino and carboxyl groups, so that there was no hydrogen bonding between the second carbonyl and the amide NH groups.⁹

We report here the synthesis of a planar-chiral amino acid in which the ansa-bridge is formed by an olefin ring-closing metathesis. [Scheme 1](#) shows the general structure of the target heterocyclic amino acids **1** with planar chirality, and the synthetic strategy consisting of ring-closing metathesis (RCM) of suitable dienes **2** in the presence of Grubbs’ catalyst. The diene precursors were prepared from sulfinyl imines. RCM is a good method for the synthesis of macrocycles¹⁰ and is also employed as a key step for the synthesis of cyclophane derivatives.¹¹ The successful RCM to a 13-membered ring was reported in the synthesis of the macrotricyclic core of roseophilin **6**.¹² The formation of the rather strained ansa-chain in this target molecule was supported by conformational control in the precursor bringing the alkene moieties closer together and lowering the enthalpic barrier during ring formation.

Preliminary computations (for a full investigation see below) indicated that such an ansa amino acid would be configurationally stable owing to a rather large barrier for internal

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Scheme 1. Top: Retrosynthetic analysis of the target heterocyclic amino acids **1** with planar chirality. Bottom: Roseophilin **6** and planar-chiral cyclophanes **7**.

rotation.¹³ Studies by König¹⁴ on the substituent effects on the rotational energy barriers of structurally related planar-chiral cyclophanes **7** using dynamic enantioselective gas chromatography and computer simulation support this reasoning for compound **1**: all substituents of compounds **7** were found to be too bulky to permit a rotation of the arene at the experimental conditions (133–145 °C).

2. Results and discussion

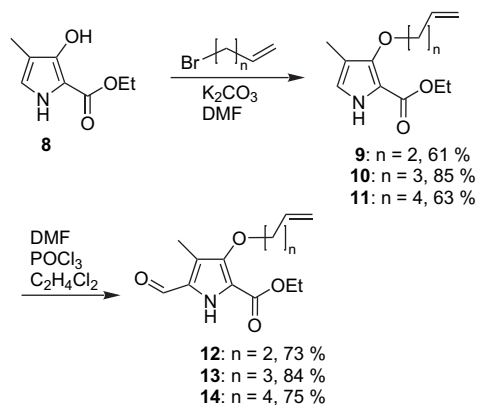
In earlier studies, we have prepared hydroxypyrrole amino acids (HOPAS)¹⁵ and incorporated them into small peptides with hairpin structures.¹⁶ The heterocyclic amino acid is a dipeptide mimic and shows a peptide β -sheet hydrogen bonding pattern. The chemistry of the HOPAS dipeptide mimic was extended by a palladium catalyzed allylation reaction, which allowed the introduction of a quaternary stereogenic centre.¹⁷ We now use the dipeptide mimic skeleton in the synthesis of a suitable precursor molecule for RCM.

Ethyl 3-hydroxy-4-methyl-pyrrole-2-carboxylate **8** was prepared as starting material in three steps according to a literature procedure.¹⁸ The first alkene functionality was introduced by alkylation of the hydroxyl group (Scheme 2). Williamson ether synthesis gave alkenyloxy pyrrole derivatives **9–11** in 61–85% yield, using potassium carbonate as base and DMF as solvent. Next, pyrrole aldehydes **12–14**

were prepared by Vilsmeier–Haack formylation of **9–11** and obtained crystalline in 73–84% yield.

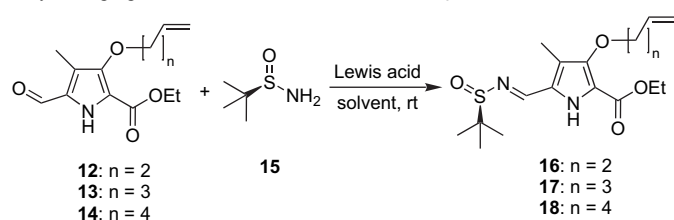
N-Sulfinyl imines are versatile intermediates in the asymmetric synthesis of chiral amines. Ellman et al. have employed Lewis acidic dehydrating agents MgSO₄, CuSO₄ and Ti(OEt)₄ for the condensation of (*R*)-*tert*-butanesulfinamide (**15**) with aldehydes.¹⁹ Recently, the formation of sulfinimines by the catalytic action of Yb(OTf)₃ was achieved and reaction conditions were extended to Ellman's sulfinyl imines.²⁰ We have tested various conditions to convert pyrrole aldehyde **13** into (*R*)-*N*-*tert*-butanesulfinyl aldimines **16–18**. CuSO₄ was not effective as Lewis acidic dehydrating agent and Yb(OTf)₃ did not induce any conversion. With Ti(OEt)₄ in dichloromethane the condensation proceeds in excellent yields for all three aldehydes **12–14** (Table 1). The results show the importance to select the right Lewis acid for the conversion of the heterocyclic aldehydes into the corresponding butanesulfinyl aldimines. The structure of compound **17** was confirmed by an X-ray structure analysis (Fig. 1).

A series of dienes with various chain lengths was prepared via nucleophilic addition of Grignard reagents to imines **16–18**. The nucleophilic additions of organomagnesium reagents to sulfinyl imines have been explored in detail by Ellman,



Scheme 2. Synthesis of pyrrole-aldehydes **12–14**.

Table 1
Synthesis of *N*-*tert*-butanesulfinyl aldimines **16–18** exploring Lewis acidic dehydrating agents CuSO₄, Ti(OEt)₄ and Yb(OTf)₃



Entry	Aldehyde	Lewis acid	Solvent	Product	Yield (%)
1	13	CuSO ₄	CH ₂ Cl ₂	17	36
2	13	Yb(OTf) ₃	THF	17	0
3	13	Ti(OEt) ₄	CH ₂ Cl ₂	17	100
4	12	Ti(OEt) ₄	CH ₂ Cl ₂	16	92
5	14	Ti(OEt) ₄	CH ₂ Cl ₂	18	91

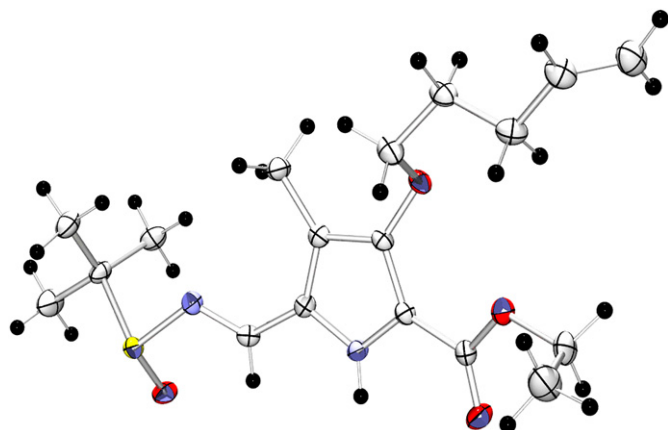
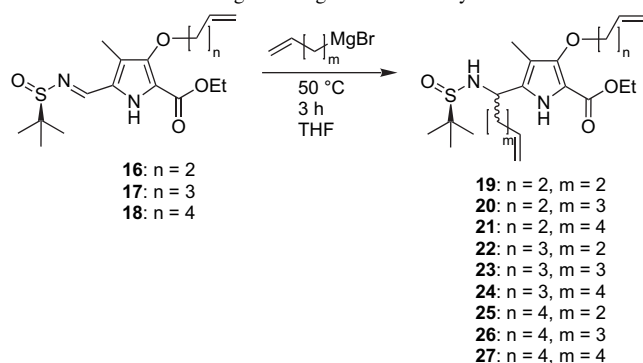


Figure 1. Structure of compound **17** in the solid state.

Tang and others.²¹ General protocols are described for the addition of alkyl and aryl Grignard reagents to *N*-sulfinyl aldimines with high diastereoselectivity and good yields. Typically, about 2 equiv of Grignard reagents are added at $-78\text{ }^{\circ}\text{C}$ or $-48\text{ }^{\circ}\text{C}$ to the *N*-sulfinyl imines to achieve addition, but in our case no nucleophilic addition of the organomagnesium reagents to the sulfinyl imine occurred in various solvents (THF, toluene, diethyl ether, and dichloromethane), even if the reaction temperature was allowed to reach room temperature; the starting material was recovered in all cases. Only if the reaction mixture was heated to $50\text{ }^{\circ}\text{C}$ nucleophilic addition occurred. The addition products were isolated in moderate to good chemical yields and *syn/anti* ratios of 9:1 (Table 2).

Table 2
Addition of unsaturated Grignard reagents to *N*-sulfinyl imines **16–18**



Entry	Product	Imine	Yield ^a (%)	Diastereomeric ratio ^b (<i>syn:anti</i>)
1	19	16	61	91:9
2	20	16	63	93:7
3	21	16	64	91:9
4	22	17	68	90:10
5	23	17	71	90:10
6	24	17	65	93:7
7	25	18	68	94:6
8	26	18	50	91:9
9	27	18	72	94:6

^a Yields were determined by mass balance of purified material.

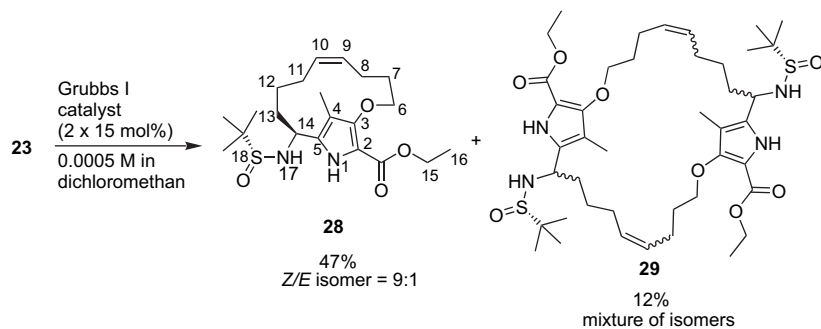
^b Ratios of diastereomers were determined by ^1H NMR or HPLC analysis.

Ring-closing diene metathesis has been applied in the synthesis of medium sized rings (9–21-membered ring macrocycles), but yields and optimal conditions (choice of catalyst; solvent) vary and are still difficult to predict.²² Therefore, the series of dienes **19–27**²³ with different side chain lengths (Table 2) were reacted with various catalysts. High dilution was used to avoid polymerization via acyclic diene metathesis. Commercially available Grubbs I, Grubbs II and second generation Hoveyda–Grubbs²⁴ catalysts were tested in dry and degassed dichloromethane and toluene.

The outcome of the RCM reactions was monitored by TLC, mass spectrometry and HPLC–MS analyses. Only a small fraction of all tested dienes gave macrocyclic products and the best conditions were found using Grubbs I catalyst ($2 \times 15\text{ mol}\%$) in high dilution conditions (0.0005 M) in dichloromethane. The formation of the expected macrocyclic structure, although in minor amounts, was indicated by HPLC and mass spectrometry for dienes **23–Boc**, **24**, **25**, **26** and **27**, but only for **23** the HPLC–MS analysis confirmed amounts of product sufficient for isolation. In the case of diene **23**, besides the 13-membered ansa-compound **28**, a 26-membered macrocycle **29** was obtained as the minor product (Scheme 3).²⁵ Both products were isolated by preparative HPLC and investigated in detail by NMR spectroscopy. Isomeric products resulting from the minor *anti* diastereomer of **23** were detected analytically, but the amount was too small for isolation. The intramolecular cyclization of compound **23** to macrocycle **28** (yield: 47%) resulted in a 9:1 ratio of *Z/E* double bond isomers, inseparable by HPLC, favouring the *Z*-configuration.²⁶ The observed coupling constant ($J=10.8\text{ Hz}$) of the olefinic protons in the ^1H NMR spectrum confirms this assignment (see Supplementary data). The proton resonance in the open-chain precursor **23** has a chemical shift of $\delta=5.86$, while in the cyclic form this resonance is shifted downfield by $\Delta\delta=0.8\text{--}0.6$. The larger macrocyclic structure **29** (yield: 12%) was isolated as a mixture of stereoisomers (*cis/trans* double bonds; head to tail and head to head orientation). Under the reaction conditions and on standing the products lose their *tert*-butanesulfinamide group.

The stereochemical analysis of the RCM reaction is hindered due to *syn/anti*-diastereomers of the starting material, diene **23**, and the formation of double bond *E/Z* isomers in the product. An attempt to reduce the number of isomers by hydrogenation of the double bond was unfortunately not successful.

Density functional theory (DFT) computations at the B3LYP/6-31G(d) level were performed to examine the conformational stability of **28**, more accurate final energies were evaluated using a larger 6-311+G(d,p) basis set. This approach has shown to provide reasonable results for the conformational energy differences of macrocycles.²⁷ The structures **K1** and **K2** were constructed as simplified models for the two possible atropisomers of **28**. The rotation of the pyrrole ring in the macrocycle **K1** leads to conversion into the other atropisomer **K2**. This path encompasses transition structure **TS1** that is accompanied by an activation enthalpy of 68.1 kcal/mol at 0 K (Fig. 2).



Scheme 3. Synthesis of 13-membered ansa-heterocyclic amino acid **28** by ring-closing olefin metathesis reaction of compound **23**.

This very high energy barrier emphasizes the conformational stability of **K1** and **K2**; it originates entirely from steric effects. During the rotation through the macrocycle, the methyl group demands significant elongation of the C–C bonds that nears bond breaking between C(33)–C(30) (Fig. 2).

For comparison, we also studied the conversion of the structure **K3**, in which the methyl group is replaced by a hydrogen atom. The internal rotation of **K3** to its atropisomer **K4** through transition state **TS2** requires an activation enthalpy (via **TS2**) of merely 6.1 kcal/mol (Fig. 3). Hence, the **K3/K4** couple would not be configurationally stable.

With the reasonable assumption that stable, non-interconverting atropisomers formed, eight product stereoisomers are possible in total. However, the 9:1 ratio of the *syn/anti* stereoisomers and the *Z/E* double bond isomers led to significantly different amounts of the product isomers, even in the absence

of any stereoinduction in the RCM. Therefore the detection of all minor compounds is difficult or even impossible.

Detailed HPLC–MS analysis of compound **28** revealed four isomeric compounds: one major isomer, two minor isomers and one isomer in traces (see [Supplementary data](#)). The major isomer has 18-*R*, 14-*S* *syn* stereochemistry and *Z*-configuration of the 9,10-double bond as determined by NMR (see [Supplementary data](#)). Using a chiral HPLC, the major isomer peak splits into two peaks in a ratio of about 3:1. This may indicate stereochemical induction of the sulfinylamine and C-14 stereocenters in the ring-closing process.²⁸ However, the overall large number of possible isomers and the small amount of product available did not allow the elucidation and assignment of the absolute configuration of the product isomers.

In conclusion, we report a synthetic route to ansa pyrrole amino acids. The ansa-bridge forms by an olefin ring-closing

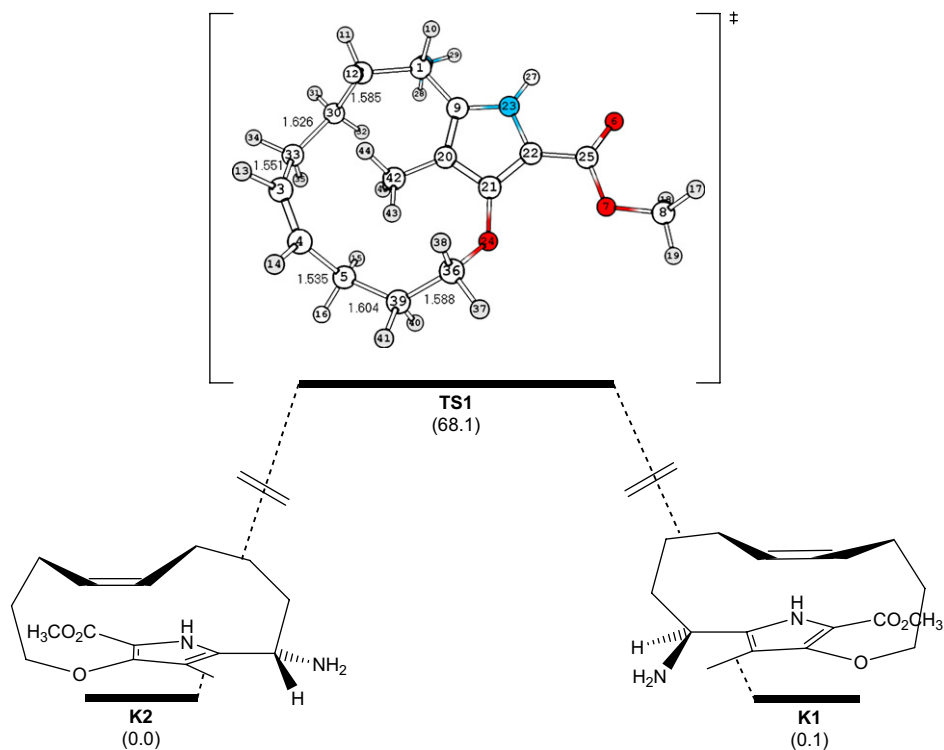


Figure 2. Computed relative enthalpies (ΔH_0) of structures **K1**, **K2**, and **TS1** in kcal/mol and optimized structure of **TS1** with selected bond lengths in Å at B3LYP/6-311+G(d,p)//B3LYP/6-31G(d) level of the theory.

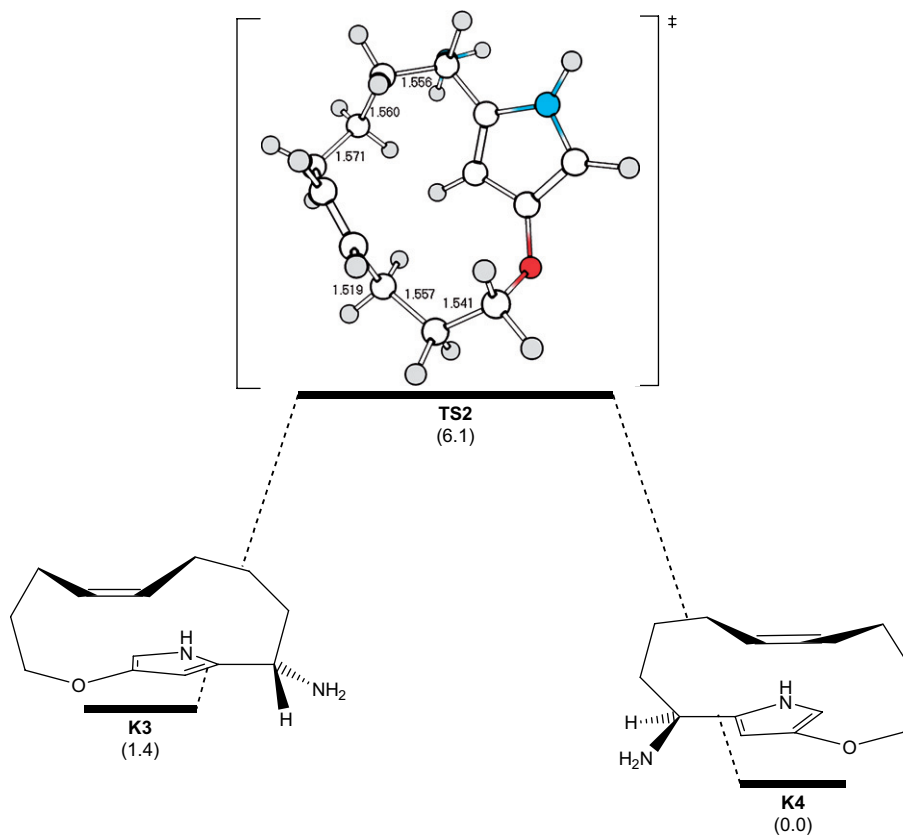


Figure 3. Computed relative enthalpies (ΔH_0) of structures **K3**, **K4**, and **TS2** in kcal/mol and optimized structure of **TS2** with selected bond lengths in Å at B3LYP/6-311+G(d,p)//B3LYP/6-31G(d) level of the theory.

metathesis reaction of diene precursors in the presence of Grubbs I catalyst. Diene precursors were prepared by Grignard addition to pyrrole sulfinyl imines. Only some of the dienes gave macrocyclic ring-closing products, as identified by HPLC–MS analyses; but yields and selectivities of the ring-closing reactions were low. Only in the case of the 13-membered compound **28** sufficient material could be obtained by preparative HPLC separation to investigate its structure spectroscopically. As also rationalized by our computations at the B3LYP/6-311+G(d,p)//B3LYP/6-31G(d) level of theory, **28** is configurationally stable. Overall, the reported synthetic approach to ansa pyrrole amino acids is feasible, but suffers from poor efficiency of the ring-closing metathesis reaction.

3. Experimental

3.1. Computational methods

All DFT computations were performed with the Gaussian 03 software package.²⁹ Structures were optimized at the DFT level of theory using B3LYP³⁰ and a 6-31G(d) basis set.³¹ Analytical vibrational frequencies were computed at B3LYP/6-31G(d) for all stationary structures to identify minima and transition structures, and to obtain zero-point vibrational energies (ZPVE). The transition structures were confirmed in all cases by the presence of one imaginary frequency. The associated vibration of this frequency was

confirmed to correspond to movements in the direction of the reaction coordinate. Additional single-point energies were evaluated at the same level of theory but with a larger basis set [6-311+G(d,p)] for all structures utilizing the B3LYP/6-31(d) geometries.²⁷

3.1.1. Ethyl 3-(but-3-enyloxy-1H-pyrrole-2-carboxylate) (**9**)

To a suspension of K_2CO_3 (364 mg, 2.63 mmol) in dry DMF (25 mL) was added ethyl 3-hydroxy-4-methyl-pyrrole-2-carboxylate **8** (420 mg, 2.48 mmol). After the reaction mixture was stirred for 10 min at room temperature, 4-bromo-1-butene (282 μ L, 2.63 mmol) was added dropwise and the reaction was stirred for 40 h at 80 °C. The reaction mixture was quenched with H_2O (100 mL) and extracted seven times with each 15 mL of CH_2Cl_2 . The collected organic layer²⁷ was washed with each 25 mL of 0.5 M NaOH, H_2O and saturated $KHSO_4$ and dried over $MgSO_4$. The solvent was evaporated and the crude product was purified using column chromatography on silica gel (PE/EtOAc=8:2, R_f =0.37) yielding **9** (339 mg, 1.52 mmol, 61%) as a colourless oil. 1H NMR (300 MHz, $CDCl_3$): δ =1.35 (t, 3J =7.1, 3H, H-12), 1.96 (d, 4J =0.8, 3H, CH_3), 2.49 (ddt, 3J =13.6, 3J =6.8, 2J =1.4, 2H, H-7), 4.04 (t, 3J =6.8, 2H, H-6), 4.29 (q, 3J =7.1, 2H, H-11), 5.00–5.19 (m, 2H, H-9), 5.91 (ddt, 3J =17.2, 3J =10.3, 3J =6.8, 1H, H-8), 6.54 (dd, 3J =3.4, 4J =0.7, 1H, H-5), 9.03 (br s, 1H, NH); ^{13}C NMR (75 MHz, $CDCl_3$): δ =8.4 (+, CH_3), 14.5 (+, C-12), 34.6 (–, C-7),

60.0 (–, C-11), 74.0 (–, C-6), 111.4 (C_{quat}, C-2), 113.2 (C_{quat}, C-4), 116.5 (–, C-9), 119.7 (+, C-5), 134.9 (+, C-8), 149.8 (C_{quat}, C-3), 160.7 (C_{quat}, C-10); IR (neat): $\tilde{\nu}$ [cm⁻¹]=3318, 3078, 2981, 2933, 2873, 2744, 1668, 1285, 1028; MS (CIMS, NH₃): *m/z* (%)=224.2 (100) MH⁺, 241.2 (37) [MNH₄]⁺; HRMS calcd for C₁₂H₁₇NO₃ [M⁺]: 223.1208; found: 223.1206±0.6 ppm—C₁₂H₁₇NO₃ (223.27).

3.1.2. Ethyl 4-methyl-3-(pent-4-enyloxy)-1H-pyrrole-2-carboxylate (**10**)

To a suspension of K₂CO₃ (1.62 g, 11.7 mmol) in dry DMF (75 mL) was added ethyl 3-hydroxy-4-methyl-pyrrole-2-carboxylate **8** (1.80 g, 10.6 mmol). After the reaction mixture was stirred for 10 min at room temperature, 5-bromo-1-pentene (1.39 mL, 11.7 mmol) was added dropwise and the reaction was stirred for 2 d at 80 °C. The reaction mixture was quenched with H₂O (400 mL) and extracted seven times with each 50 mL of CH₂Cl₂. The collected organic layer was washed with each 100 mL of 0.5 M NaOH, H₂O, saturated KHSO₄ and dried over MgSO₄. The solvent was evaporated and the crude product was purified using column chromatography on silica gel (PE/EtOAc=7:3, R_f=0.45) yielding **10** (2.14 g, 9.02 mmol, 85%) as a colourless oil. ¹H NMR (600 MHz, CDCl₃): δ =1.35 (t, ³J=7.1, 3H, H-13), 1.86 (m, 2H, H-7), 1.99 (d, ⁴J=0.8, 3H, CH₃), 2.28 (m, 2H, H-8), 4.02 (t, ³J=6.5, 2H, H-6), 4.31 (q, ³J=7.4, 2H, H-12), 4.98 (ddt, ³J=10.2, ³J=2.0, ⁴J=1.3, 1H, H-10_{cis}), 5.05 (ddt, ³J=17.2, ³J=2.0, ⁴J=1.6, 1H, H-10_{trans}), 5.86 (ddt, ³J=17.2, ³J=10.2, ³J=6.6, 1H, H-9), 6.57 (dq, ³J=3.4, ⁴J=0.8, 1H, H-5), 8.58 (d, ³J=3.4, 1H, NH); ¹³C NMR (150 MHz, CDCl₃): δ =8.5 (+, CH₃), 14.5 (+, C-13), 29.4 (–, C-7), 30.2 (–, C-8), 60.0 (–, C-12), 74.2 (–, C-6), 111.5 (C_{quat}, C-2), 113.3 (C_{quat}, C-4), 114.8 (–, C-10), 119.6 (+, C-5), 138.2 (+, C-9), 149.9 (C_{quat}, C-3), 160.5 (C_{quat}, C-11); IR (neat): $\tilde{\nu}$ [cm⁻¹]=3318, 3076, 2979, 2938, 2872, 2743, 1668, 1285, 1029 MS (EI, 70 eV): *m/z* (%)=237.1 (23) [M⁺], 169.0 (26) [M–C₄H₈]⁺, 123.0 (100) [M–C₄H₈–C₂H₆O]⁺; HRMS calcd for C₁₃H₁₉NO₃ [M⁺]: 237.1369; found: 237.1363±2.0 ppm—C₁₃H₁₉NO₃ (237.30).

3.1.3. Ethyl 3-(hex-5-enyloxy)-4-methyl-1H-pyrrole-2-carboxylate (**11**)

To a suspension of K₂CO₃ (491 mg, 3.55 mmol) in dry DMF (15 mL) was added ethyl 3-hydroxy-4-methyl-pyrrole-2-carboxylate **8** (600 mg, 3.55 mmol). After the reaction mixture was stirred for 10 min at room temperature, 6-bromo-1-hexene (460 μ L, 3.55 mmol) was added dropwise and the reaction was stirred for 40 h at 80 °C. The reaction mixture was quenched with H₂O (75 mL) and extracted seven times with each 10 mL of CH₂Cl₂. The collected organic layer was washed with each 20 mL of 0.5 M NaOH, H₂O and saturated KHSO₄ and dried over MgSO₄. The solvent was evaporated and the crude product was purified using column chromatography on silica gel (PE/EtOAc=8:2, R_f=0.40) yielding **6** (565 mg, 2.25 mmol, 63%) as a colourless oil. ¹H NMR (300 MHz, CDCl₃): δ =1.28 (t, ³J=7.1, 3H, H-14), 1.46–1.58 (m, 2H, H-8), 1.64–1.76 (m, 2H, H-7), 1.92 (d,

⁴J=0.7, 3H, CH₃), 2.00–2.11 (m, 2H, H-9), 3.94 (t, ³J=6.5, 2H, H-6), 4.25 (q, ³J=7.1, 2H, H-13), 4.83–5.03 (m, 2H, H-11), 5.76 (ddt, ³J=17.0, ³J=10.3, ³J=6.7, 1H, H-10), 6.50 (dd, ³J=3.4, ⁴J=0.7, 1H, H-5), 8.82 (br s, 1H, NH); ¹³C NMR (75 MHz, CDCl₃): δ =7.5 (+, CH₃), 13.5 (+, C-14), 24.3 (–, C-8), 28.7 (–, C-7), 32.6 (–, C-9), 58.9 (–, C-13), 73.7 (–, C-6), 110.4 (C_{quat}, C-2), 112.2 (C_{quat}, C-4), 113.5 (–, C-11), 118.8 (+, C-5), 137.7 (+, C-10), 149.0 (C_{quat}, C-3), 150.7 (C_{quat}, C-12); IR (neat): $\tilde{\nu}$ [cm⁻¹]=3315, 2978, 2936, 2868, 1666, 1285, 1028; MS (EI, 70 eV): *m/z* (%)=251.3 (11) [M⁺], 169.2 (36) [M–C₆H₁₀]⁺, 123.0 (100) [M–C₆H₁₀–C₂H₆O]⁺; HRMS calcd for C₁₄H₂₁NO₃ [M⁺]: 251.1521; found: 251.15±0.9 ppm—C₁₄H₂₁NO₃ (251.32).

3.1.4. Ethyl 3-(but-enyloxy)-5-formyl-4-methyl-1H-pyrrole-2-carboxylate (**12**)

Compound **9** (220 mg, 0.99 mmol) in C₂H₄Cl₂ (10 mL) was added dropwise to an ice cooled solution of DMF (85.5 μ L, 1.10 mmol) and POCl₃ (101 μ L, 1.10 mmol) in C₂H₄Cl₂ (10 mL), the mixture was stirred for 30 min, another 30 min at room temperature and was refluxed for 24 h. H₂O (40 mL) and EtOAc (20 mL) were added to the cooled mixture and the aqueous layer was extracted three times with each 20 mL of EtOAc. The combined organic layers were washed three times with 10% solution of Na₂CO₃ and dried over MgSO₄. The solvent was evaporated and the crude product was purified using column chromatography on flash-silica gel (PE/EtOAc=8:2, R_f=0.21) yielding **12** (182 mg, 0.72 mmol, 73%), as white crystals, mp=62 °C; ¹H NMR (300 MHz, CDCl₃): δ =1.34 (t, ³J=7.1, 3H, H-12), 2.23 (s, 3H, CH₃), 2.37–2.56 (m, 2H, H-7), 4.03 (t, ³J=6.7, 2H, H-6), 4.32 (q, ³J=7.1, 2H, H-11), 4.92–5.24 (m, 2H, H-9), 5.72–6.06 (m, 1H, H-8), 9.20 (br s, 1H, NH), 9.69 (s, 1H, CHO); ¹³C NMR (300 MHz, CDCl₃): δ =6.9 (+, CH₃), 14.3 (+, C-12), 34.5 (–, C-7), 61.1 (–, C-11), 74.5 (–, C-6), 117.0 (–, C-11), 117.6 (C_{quat}, C-2), 122.6 (C_{quat}, C-4), 127.8 (C_{quat}, C-5), 134.5 (+, C-8), 149.1 (C_{quat}, C-3), 159.6 (C_{quat}, C-10), 179.1 (+, CHO); IR (KBr): $\tilde{\nu}$ [cm⁻¹]=3442, 3261, 2982, 2928, 2861, 2263, 1676, 1280, 1023; MS (EI, 70 eV): *m/z* (%)=251.2 (26) [M⁺], 151.1 (100) [M⁺–NH–(S=O)(CH₃)₃]; HRMS calcd for C₁₃H₁₇NO₄ [M⁺]: 251.1158; found: 251.1162±1.6 ppm.

3.1.5. Ethyl 5-formyl-4-methyl-3-(pent-4-enyloxy)-1H-pyrrole-2-carboxylate (**13**)

Compound **10** (1.94 g, 8.15 mmol) in C₂H₄Cl₂ (10 mL) was added dropwise to an ice cooled solution of DMF (698 μ L, 8.97 mmol) and POCl₃ (821 μ L, 8.97 mmol) in C₂H₄Cl₂ (20 mL), the mixture was stirred for 30 min, another 30 min at room temperature and was refluxed for 24 h. H₂O (80 mL) and EtOAc (40 mL) were added to the cooled mixture and the aqueous layer was extracted three times with each 70 mL EtOAc. The combined organic layers were washed three times with 10% solution of Na₂CO₃ and dried over MgSO₄. The solvent was evaporated and the crude product was purified using column chromatography on flash-silica gel (PE/EtOAc=9:1, R_f=0.40) yielding **13** (1.82 g, 6.87 mmol,

84%) as white crystals, mp=49 °C; ¹H NMR (400 MHz, CDCl₃): δ=1.34 (t, ³J=7.2, 3H, H-13), 1.83 (m, 2H, H-7), 2.22 (m, 2H, H-8), 2.24 (s, 3H, CH₃), 3.98 (t, ³J=6.5, 2H, H-6), 4.34 (q, ³J=7.2, 2H, H-12), 4.96 (ddt, ³J=10.2, ³J=2.0, ⁴J=1.3, 1H, H-10cis), 5.02 (ddt, ³J=17.1, ³J=2.0, ⁴J=1.6, 1H, H-10trans), 5.82 (ddt, ³J=17.1, ³J=10.2, ³J=6.6, 1H, H-9), 9.55 (br s, 1H, NH), 9.71 (s, 1H, CHO); ¹³C NMR (100 MHz, CDCl₃): δ=7.0 (+, CH₃), 14.3 (+, C-13), 29.3 (−, C-7), 30.1 (−, C-8), 61.1 (−, C-12), 74.7 (−, C-6), 115.0 (+, C-10), 117.7 (C_{quat}, C-2), 122.4 (C_{quat}, C-4), 128.0 (C_{quat}, C-5), 137.9 (+, C-9), 149.2 (C_{quat}, C-3), 160.5 (C_{quat}, C-11), 179.4 (+, CHO); MS (CI, NH₃): *m/z* (%)=283.3 (100) [MNH₄]⁺, 266.2 (57) MH⁺. Elemental analysis calcd (%) for C₁₄H₁₉NO₄ (265.31): C 63.36, H 7.22, N 5.28; found: C 63.16, H 7.57, N 5.21.

3.1.6. Ethyl 5-formyl-3-(hex-5-enyloxy)-4-methyl-1H-pyrrole-2-carboxylate (**14**)

Compound **11** (315 mg, 1.25 mmol) in C₂H₄Cl₂ (10 mL) was added dropwise to an ice cooled solution of DMF (107 μL, 1.38 mmol) and POCl₃ (125 μL, 8.97 mmol) in C₂H₄Cl₂ (10 mL), the mixture was stirred for 30 min, another 30 min at room temperature and was refluxed for 24 h. H₂O (40 mL) and EtOAc (20 mL) were added to the cooled mixture and the aqueous layer was extracted three times with 20 mL of EtOAc. The combined organic layers were washed three times with 10% solution of Na₂CO₃ and dried over MgSO₄. The solvent was evaporated and the crude product was purified using column chromatography on flash-silica gel (PE/EtOAc=9:1, *R_f*=0.41) yielding **9** (265 mg, 0.95 mmol, 75%) as white crystals, mp=48 °C; ¹H NMR (300 MHz, CDCl₃): δ=1.38 (t, ³J=7.1, 3H, H-14), 1.48–1.87 (m, 4H, H-7 and H-8), 2.05–2.20 (m, 2H, H-9), 2.27 (s, 3H, CH₃), 4.01 (t, ³J=6.6, 2H, H-6), 4.36 (q, ³J=7.1, 2H, H-13), 4.90–5.10 (m, 2H, H-11), 5.72–5.93 (m, 1H, H-10), 9.24 (br s, 1H, NH), 9.73 (s, 1H, CHO); ¹³C NMR (75 MHz, CDCl₃): δ=6.9 (+, CH₃), 14.4 (+, C-14), 25.3 (−, C-8), 29.5 (−, C-7), 33.5 (−, C-9), 61.1 (−, C-13), 75.3 (−, C-6), 114.7 (−, C-11), 117.6 (C_{quat}, C-2), 122.5 (C_{quat}, C-4), 127.8 (C_{quat}, C-5), 138.5 (+, C-10), 149.3 (C_{quat}, C-3), 159.7 (C_{quat}, C-12), 179.1 (+, CHO); IR (KBr): $\tilde{\nu}$ [cm^{−1}]=3447, 3268, 2979, 2940, 2867, 2362, 1672, 1277, 1027; MS (EI, 70 eV): *m/z* (%)=279.3 (5) [M⁺], 197.1 (48) [M−C₆H₁₀]⁺, 151.1 (100) [M−C₆H₁₀−C₂H₆O]⁺; HRMS calcd for C₁₅H₂₁NO₄ [M⁺]: 279.1471; found: 279.1475±1.9 ppm—C₁₅H₂₁NO₄ (279.34).

3.2. General procedure (GP1) for the synthesis of tert-butanesulfinyl imines

To a solution of pyrrole aldehyde (1.0 equiv) and Ti(OEt)₄ (2.0 equiv) in dry dichloromethane (5 mL) was added (*R*)-tert-butanesulfinamide (1.2 equiv) under dinitrogen, and the mixture was stirred at the given temperature and time. The reaction mixture was quenched with a mixture of saturated NH₄Cl solution (10 mL) and brine (10 mL) while vigorously stirred. The resulting suspension was filtered through a plug of Celite and the filter cake was washed well with EtOAc (30 mL). The filtrate was transferred to a separatory funnel,

where the organic layer was washed three times with each 10 mL of brine. The brine layer was extracted once with a small volume of EtOAc and the combined organic portions were dried over MgSO₄, filtered and concentrated under vacuum. The sulfinyl imines were purified by silica gel chromatography if no other method is given.

3.2.1. (*R*)-Ethyl 3-(but-3-enyloxy)-4-methyl-5-((2-methylpropan-2-ylsulfinamido)methyl)-1H-pyrrole-2-carboxylate (**16**)

Compound **12** (251 mg, 0.64 mmol), (*R*)-tert-butanesulfinamide (**15**, 93 mg, 0.77 mmol) and Ti(OEt)₄ (292 mg, 1.28 mmol) in dry dichloromethane (5 mL) were allowed to react according to the GP1 at room temperature for 36 h yielding 152 mg (0.43 mmol, 92%, conversion corrected yield, 43 mg of starting material regained) of **16** (Et₂O/hexanes=1:1; *R_f*=0.22) as colourless crystals, mp=81 °C; ¹H NMR (300 MHz, CDCl₃): δ=1.25 (s, 9H, *t*-Bu), 1.40 (t, ³J=7.1, 3H, H-12), 2.21 (s, 3H, CH₃), 2.52–2.55 (m, 2H, H-7), 4.06 (t, ³J=6.8, 2H, H-6), 4.39 (q, ³J=7.1, H-11), 5.09–5.21 (m, 2H, H-9), 5.92–5.94 (m, 1H, H-8), 8.47 (s, 1H, CHN), 9.28 (br s, 1H, NH); ¹³C NMR (75 MHz, CDCl₃): δ=7.5 (+, CH₃), 14.4 (+, C-12), 22.5 (+, *t*-Bu), 34.5 (−, C-7), 57.8 (+, *t*-Bu), 60.9 (−, C-11), 74.4 (−, C-6), 116.1 (C_{quat}, C-2), 116.9 (−, C-9), 120.2 (C_{quat}, C-4), 126.2 (C_{quat}, C-5), 134.5 (+, C-8), 149.5 (C_{quat}, C-3), 150.5 (+, CHN), 160.1 (C_{quat}, C-10); IR (KBr): $\tilde{\nu}$ [cm^{−1}]=3447, 3256, 2980, 2959, 2926, 2868, 1701, 1593, 1272, 1059, 744; MS (FAB+): *m/z* (%)=355 (100) [MH]⁺, 289 (43) [M−C₄H₈]⁺; HRMS calcd for C₁₇H₂₇N₂O₄S⁺: 355.1697; found: 355.1692+1.5 ppm—C₁₇H₂₆N₂O₄S (354.46).

3.2.2. (*R*)-Ethyl 4-methyl-5-[(2-methylpropan-2-ylsulfinamido)methyl]-3-pent-4-enyloxy-1H-pyrrole-2-carboxylate (**17**)

Compound **13** (450 mg, 1.70 mmol), (*R*)-tert-butanesulfinamide (**15**, 247 mg, 2.04 mmol) and Ti(OEt)₄ (776 mg, 3.4 mmol) in dry dichloromethane were allowed to react according to the GP1 at 35 °C for 48 h yielding 625 mg (quantitative) of **17** (Et₂O/hexanes=1:1; *R_f*=0.32) as colourless crystals, mp=79 °C; ¹H NMR (400 MHz, CDCl₃): δ=1.24 (s, 9H, *t*-Bu), 1.39 (t, ³J=7.1, 3H, H-13), 1.80–1.93 (m, 2H, H-7), 2.20 (s, 3H, CH₃), 2.22–2.31 (m, 2H, H-8), 4.01 (t, ³J=6.5, 2H, H-6), 4.37 (dq, ³J=7.1, ²J=1.1, 2H, H-12), 4.93–5.15 (m, 2H, H-10), 5.86 (ddt, ³J=17.0, ³J=10.3, ³J=6.7, 1H, H-9), 8.44 (s, 1H, CHN), 9.11 (br s, 1H, NH); ¹³C NMR (100 MHz, CDCl₃): δ=7.4 (+, CH₃), 14.4 (+, C-13), 22.6 (+, *t*-Bu), 29.3 (−, C-7), 30.1 (−, C-8), 57.9 (+, *t*-Bu), 60.9 (−, C-12), 74.7 (−, C-6), 115.0 (−, C-10), 116.2 (C_{quat}, C-2), 120.3 (C_{quat}, C-4), 126.2 (C_{quat}, C-5), 138.0 (+, C-9), 149.6 (C_{quat}, C-3), 150.3 (+, CHN), 160.1 (C_{quat}, C-11); IR (KBr): $\tilde{\nu}$ [cm^{−1}]=3437, 3227, 2980, 2935, 2869, 1693, 1265, 1057, 1025, 746; MS (EI, 70 eV): *m/z* (%)=368.2 (6) [M⁺], 312.0 (100) [M−C₄H₈]⁺. Elemental analysis calcd (%) for C₁₈H₂₈N₂O₄S (368.49): C 58.67, H 7.66, N 7.60; found: C 58.61, H 7.68, N 7.31; Crystal data: C₁₈H₂₈N₂O₄S, *M_r*=368.49, colourless flat prism, triclinic, space group *P*1, *a*=8.9068(9) Å, *b*=10.5174(12) Å,

$c=11.9303(12)\text{ \AA}$, $\alpha=71.989(12)^\circ$, $\beta=84.932(12)^\circ$, $\gamma=70.971(13)^\circ$, $Z=2$, $V=1004.7(2)\text{ \AA}^3$, $D_x=1.218\text{ mg/m}^3$, $\mu=0.184\text{ mm}^{-1}$, $F(000)=396$, crystal size $0.44\times 0.30\times 0.12\text{ mm}$, θ -range for data collections $2.35^\circ\text{--}26.82^\circ$, index ranges $-11\leq h\leq 11$, $-13\leq k\leq 13$, $-15\leq l\leq 15$, reflections collected/unique $10,549/7803$ [$R_{\text{int}}=0.0218$], data/restraints/parameters $7803/3/461$, goodness-of-fit on F^2 0.993 , final R indices [$I>2\sigma(I)$] $R_1=0.0278$, $wR_2=0.0654$, R indices (all data) $R_1=0.0307$, $wR_2=0.0663$, largest diff. peak and hole 0.328 and -0.136 e \AA^{-3} .

3.2.3. (*R*)-Ethyl 3-(hex-5-enyloxy)-4-methyl-5-((2-methylpropan-2-ylsulfonamido)methyl)-1*H*-pyrrole-2-carboxylate (**18**)

Compound **14** (240 mg, 0.86 mmol), (*R*)-*tert*-butanesulfonamide (**15**, 125 mg, 1.03 mmol) and $\text{Ti}(\text{OEt})_4$ (392 mg, 1.72 mmol) in dry dichloromethane were allowed to react according to the GP1 at room temperature for 36 h yielding 240 mg (92%, conversion corrected yield, 48 mg of **14** regained) of **18** ($\text{Et}_2\text{O}/\text{hexanes}=1:1$; $R_f=0.22$) as white crystals, $\text{mp}=75^\circ\text{C}$; $^1\text{H NMR}$ (300 MHz, CDCl_3): $\delta=1.26$ (s, 9H, *t*-Bu), 1.40 (t, $^3J=7.1$, 3H, H-14), 1.57–1.62 (m, 2H, H-8), 1.77–1.83 (m, 2H, H-7), 2.10–2.15 (m, 2H, H-9), 2.21 (s, 3H, CH_3), 4.01 (t, $^3J=6.5$, 2H, H-6), 4.39 (q, $^3J=7.1$, 2H, H-13), 4.96–5.07 (m, 2H, H-11), 5.83–5.85 (m, 1H, H-10), 8.46 (s, 1H, CHN), 9.25 (br s, 1H, NH); $^{13}\text{C NMR}$ (75 MHz, CDCl_3): $\delta=6.7$ (+, CH_3), 13.7 (+, C-14), 21.8 (+, *t*-Bu), 24.6 (–, C-8), 28.9 (–, C-7), 32.8 (–, C-9), 57.1 (+, *t*-Bu), 60.2 (–, C-13), 74.5 (–, C-6), 114.2 (–, C-11), 115.5 (C_{quat} , C-2), 119.6 (C_{quat} , C-4), 125.5 (C_{quat} , C-5), 137.8 (+, C-10), 149.0 (C_{quat} , C-3), 149.7 (+, CHN), 159.4 (C_{quat} , C-12); IR (KBr): $\tilde{\nu} [\text{cm}^{-1}]=3445, 3120, 2989, 2939, 2866, 2701, 1693, 1504, 1267, 1056, 747$; MS (FAB+): m/z (%)=383 (100) [MH] $^+$, 326 (40) [$\text{M}^+-\text{C}_4\text{H}_8$] $^+$; HRMS calcd for $\text{C}_{19}\text{H}_{31}\text{N}_2\text{O}_4\text{S}^+$: 382.2015; found: 383.2005+2.8 ppm.

3.3. General procedure (GP2) for the Grignard addition to *N*-*tert*-butanesulfinyl imines

To a solution of the sulfinyl imine (1 equiv) in THF, the appropriate Grignard reagent (2.5–3.3 equiv) in dry Et_2O (5 mL) was added dropwise and the conversion was monitored by TLC. The reaction was stirred at $50\text{--}60^\circ\text{C}$ for approximately 4 h. Upon reaction completion, the excess organometallic reagent was destroyed with saturated aqueous NH_4Cl (10 mL) and the resulting suspension was diluted with brine (10 mL). The suspension was filtered through a plug of Celite and the filter cake was washed with EtOAc ($2\times 10\text{ mL}$). The filtrate was transferred into a separatory funnel, the aqueous layer was washed with EtOAc ($3\times 10\text{ mL}$), organic layers were combined, dried over MgSO_4 and concentrated to afford the crude product. Diastereomeric ratios were determined by HPLC analyses or by NMR of the crude product. Purification was performed by flash-silica chromatography using Biotage SP4 chromatography system. Yields correspond to the mass balance of purified material.

3.3.1. (*R*)-Ethyl 3-(but-3-enyloxy)-5-(1-(1,1-dimethylethylsulfonamido)pent-4-enyl)-4-methyl-1*H*-pyrrole-2-carboxylate (**19**)

Sulfinyl imine **16** (44 mg, 0.12 mmol) and but-3-enylmagnesium bromide (0.24 mL, 1.3 M in Et_2O , 0.3 mmol) were reacted according to the GP2. Flash-silica chromatography (Biotage SP4 chromatography system, $\text{EtOAc}/\text{cyclohexane}=7:3$) gave 30 mg (61%) of **19** in a diastereomeric *syn/anti* ratio of 91:9. The diastereomeric ratio was determined by $^1\text{H NMR}$. $^1\text{H NMR}$ (300 MHz, CDCl_3): $\delta=1.18$ (s, 9H, $(\text{CH}_3)_3$), 1.31 (t, $^3J=7.1$, 3H, H-12), 1.80–2.06 (m, 4H, H-14, H-15), 1.90 (s, 3H, CH_3), 2.40–2.52 (m, 2H, H-7), 3.95–4.06 (m, 2H, H-6), 4.20–4.32 (m, 2H, H-11), 4.41–4.50 (m, 1H, H-13), 4.90–5.16 (m, 4H, H-9, H-17), 5.71 (dddd, $^3J=16.9$, $^3J=10.3$, $^3J=6.5$, $^3J=6.1$, 1H, H-8), 5.80–5.96 (m, 1H, H-16), 9.04 (br s, 1H, N-H).

3.3.2. (*R*)-Ethyl 3-(but-3-enyloxy)-5-(1-(1,1-dimethylethylsulfonamido)hex-5-enyl)-4-methyl-1*H*-pyrrole-2-carboxylate (**20**)

Sulfinyl imine **16** (32 mg, 0.09 mmol) and pent-4-enylmagnesium bromide (0.25 mL, 0.94 M in Et_2O , 0.23 mmol) were reacted according to the GP2. Flash-silica chromatography (Biotage SP4 chromatography system, $\text{EtOAc}/\text{cyclohexane}=7:3$) gave 24 mg (63%) of **20** in a diastereomeric *syn/anti* ratio of 93:7. The diastereomeric ratio was determined by $^1\text{H NMR}$. $^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta=1.14$ (s, 9H, $(\text{CH}_3)_3$), 1.29 (t, $^3J=7.1$, 3H, H-12), 1.32–1.38 (m, 2H, H-15), 1.88 (s, 3H, CH_3), 1.92–2.01 (m, 2H, H-16), 2.40–2.48 (m, 2H, H-7), 3.99 (dt, $^3J=6.7$, $^2J=2.3$, 2H, H-6), 4.16–4.29 (m, 2H, H-11), 4.40 (ddd, $^3J=7.3$, $^3J=7.3$, $J=1.6$, 1H, H-13), 4.83–5.24 (m, 4H, H-9, H-18), 5.65 (dddd, $^3J=17.0$, $^3J=10.1$, $^3J=6.8$, $^3J=6.7$, 1H, H-8), 5.85 (dddd, $^3J=17.1$, $^3J=10.3$, $^3J=6.8$, $^3J=6.7$, 1H, H-17), 9.01 (br s, 1H, N-H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): $\delta=7.5$ (+, CH_3), 14.5 (+, C-12), 22.6 (+, $(\text{CH}_3)_3$), 25.2 (–, C-15), 33.2 (–, C-14), 34.6 (–, C-16), 35.5 (–, C-7), 52.1 (+, C-13), 55.7 (C_{quat} , C- $(\text{CH}_3)_3$), 60.0 (–, C-11), 74.1 (–, C-6), 110.5 (C_{quat} , C-2), 111.0 (C_{quat} , C-4), 115.2 (–, C-18), 116.6 (–, C-9), 131.5 (C_{quat} , C-5), 134.9 (+, C-8), 137.9 (+, C-17), 149.8 (C_{quat} , C-3), 160.4 (C_{quat} , C-11); IR (neat): $\tilde{\nu} [\text{cm}^{-1}]=3448, 3255, 3077, 2979, 2929, 2866, 1665, 1468, 1032, 911$; MS (FAB+): m/z (%)=425 (23) [$\text{M}+\text{H}$] $^+$, 304 (100) [$\text{M}^+-\text{NH}-(\text{S}=\text{O})(\text{CH}_3)_3$] $^+$; HRMS calcd for $\text{C}_{22}\text{H}_{37}\text{O}_4\text{N}_2\text{S}$: 425.2574; found: 425.2487+3.1 ppm.

3.3.3. (*R*)-Ethyl 3-(but-3-enyloxy)-5-(1-(1,1-dimethylethylsulfonamido)hept-6-enyl)-4-methyl-1*H*-pyrrole-2-carboxylate (**21**)

Sulfinyl imine **16** (40 mg, 0.11 mmol) and hex-5-enylmagnesium bromide (0.34 mL, 0.82 M in Et_2O , 0.28 mmol) were reacted according to the GP2. Flash-silica chromatography (Biotage SP4 chromatography system, $\text{EtOAc}/\text{cyclohexane}=7:3$) gave 31 mg (64%) of **21** in a diastereomeric *syn/anti* ratio of 91:9. The diastereomeric ratio was determined by $^1\text{H NMR}$. $^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta=1.11\text{--}1.36$ (m, 4H, H-15, H-16), 1.15 (s, 9H, $(\text{CH}_3)_3$), 1.29 (t, $^3J=7.1$, 3H, H-12), 1.67–1.83 (m, 2H, H-14), 1.88 (s, 3H, CH_3), 1.90–1.99 (m, 2H, H-17), 2.44 (ddt, $^3J=6.8$, $^3J=6.8$, $^2J=1.4$, 2H, H-7), 3.50 (d, $^3J=2.2$, 1H, N-H),

3.94–4.05 (m, 2H, H-69), 4.17–4.30 (m, 2H, H-11), 4.35–4.44 (m, 1H, H-13), 4.83–4.93 (m, 2H, H-9), 4.98–5.12 (m, 2H, H-19), 5.67 (dddd, $^3J=17.0$, $^3J=10.2$, $^3J=6.7$, $^3J=6.7$, 1H, H-8), 5.86 (dddd, $^3J=17.1$, $^3J=10.3$, $^3J=6.7$, $^3J=6.7$, 1H, H-18), 9.00 (br s, 1H, H-pyrrole); ^{13}C NMR (100 MHz, CDCl_3): $\delta=7.5$ (+, CH_3), 14.5 (+, C-12), 22.6 (+, $(\text{CH}_3)_3$), 25.4 (–, C-16), 28.4 (–, C-15), 33.4 (–, C-14), 34.6 (–, C-17), 35.9 (–, C-7), 52.1 (+, C-13), 55.7 (C_{quat} , C- $(\text{CH}_3)_3$), 59.9 (–, C-11), 74.1 (–, C-6), 110.5 (C_{quat} , C-2), 111.0 (C_{quat} , C-4), 114.7 (–, C-19), 116.6 (–, C-9), 131.6 (C_{quat} , C-5), 135.0 (+, C-8), 138.4 (+, C-18), 149.8 (C_{quat} , C-3), 160.4 (C_{quat} , C-11); IR (neat): $\tilde{\nu}$ [cm^{-1}]=3422, 2979, 2929, 2861, 1647, 1465, 1280, 1032; MS (EI, 70 eV): m/z (%)=438.2 (6) [M^+], 318.2 (100) [$\text{M}^+-\text{NBoc}-(\text{S}=\text{O})(\text{CH}_3)_3$]; HRMS calcd for $\text{C}_{23}\text{H}_{38}\text{N}_2\text{O}_4\text{S}$ [M^+]: 438.2548; found: 438.2548 \pm 1.9 ppm.

3.3.4. (R)-Ethyl 5-(1-(1,1-dimethylethylsulfonamido)pent-4-enyl)-4-methyl-3-(pent-4-enyloxy)-1H-pyrrole-2-carboxylate (22)

Sulfinyl imine **17** (50 mg, 0.14 mmol) and but-3-enylmagnesium bromide (0.28 mL, 1.3 M in Et_2O , 0.35 mmol) were reacted according to the GP2. Flash-silica chromatography (Biotage SP4 chromatography system, $\text{EtOAc}/\text{cyclohexane}=7:3$) gave 40 mg (68%) of **22** in a diastereomeric *syn/anti* ratio of 90:7. The diastereomeric ratio was determined by ^1H NMR and HPLC. ^1H NMR (600 MHz, CDCl_3): $\delta=1.24$ (s, 9H, $(\text{CH}_3)_3$), 1.37 (t, $^3J=7.1$, 3H, H-13), 1.84–1.89 (m, 2H, H-7), 1.97 (s, 3H, CH_3), 1.09–2.03 (m, 2H, H-15), 2.04–2.12 (m, 2H, H-16), 2.24–2.29 (m, 2H, H-8), 3.55 (d, $^3J=1.5$, 1H, N-H), 4.04 (dt, $^2J=6.4$, $^3J=2.7$, 2H, H-6), 4.30–4.37 (m, 2H, H-12), 4.53 (dt, $^3J=7.1$, $^3J=1.9$, 1H, H-14), 5.00 (ddt, $^3J=10.3$, $^2J^\dagger$ and $^4J^\dagger$, 1H, H-10cis), 5.03 (ddt, $^3J=10.3$, $^2J^\dagger$ and $^4J^\dagger$, 1H, H-18cis), 5.05 (ddt, $^3J=16.9$, $^2J^\dagger$ and $^4J^\dagger$, 1H, H-18trans), 5.07 (ddt, $^3J=17.0$, 2J and $^4J^\dagger$, 1H, H-10trans), 5.78 (ddt, $^3J=16.9$, $^3J=10.3$, $^3J=6.6$, 1H, H-17), 5.88 (ddt, $^3J=17.0$, $^3J=10.3$, $^3J=6.7$, 1H, H-19), 8.98 (br s, 1H, pyrrole-H); ^{13}C NMR (150 MHz, CDCl_3): $\delta=7.5$ (+, CH_3), 14.5 (+, C-13), 22.6 (+, $(\text{CH}_3)_3$), 29.4 (–, C-7), 30.1 (–, C-16), 30.2 (–, C-8), 35.0 (–, C-15), 51.6 (+, C-14), 55.7 (C_{quat} , C- $(\text{CH}_3)_3$), 59.9 (–, C-12), 74.3 (–, C-6), 110.6 (C_{quat} , C-2), 111.1 (C_{quat} , C-4), 114.8 (–, C-10), 115.9 (–, C-18), 131.1 (C_{quat} , C-5), 137.0 (+, C-17), 138.2 (+, C-9), 150.0 (C_{quat} , C-3), 160.3 (C_{quat} , C-11); IR (neat): $\tilde{\nu}$ [cm^{-1}]=3460, 3252, 3077, 2978, 2927, 2868, 1665, 1468, 1032, 911; MS (EI, 70 eV): m/z (%)=424.2 (6) [M^+], 304.2 (100) [$\text{M}^+-\text{NH}-(\text{S}=\text{O})(\text{CH}_3)_3$], 367.8 [$\text{M}^+-\text{C}_4\text{H}_9$]; HRMS calcd for $\text{C}_{22}\text{H}_{37}\text{N}_2\text{O}_4\text{S}$ [M^+]: 425.2474; found: 425.2475 \pm 0.3 ppm.

3.3.5. (R)-Ethyl 5-(1-(1,1-dimethylethylsulfonamido)hex-5-enyl)-4-methyl-3-(pent-4-enyloxy)-1H-pyrrole-2-carboxylate (23)

Sulfinyl imine **17** (100 mg, 0.27 mmol) and pent-4-enylmagnesium bromide (0.73 mL, 1.3 M in Et_2O , 0.90 mmol)

were reacted according to the GP2. Flash-silica chromatography (Biotage SP4 chromatography system, $\text{EtOAc}/\text{cyclohexane}=7:3$) gave 40 mg (71%) of **23** in a diastereomeric *syn/anti* ratio of 9:1. The diastereomeric ratio was determined by ^1H NMR and HPLC. ^1H NMR (600 MHz, CDCl_3): $\delta=1.20$ (s, 9H, $(\text{CH}_3)_3$), 1.35 (t, $^3J=7.0$, 3H, H-13), 1.22–1.46 (m, 2H, H-16), 1.77–1.93 (m, 4H, H-7, H-15), 1.95 (s, 3H, CH_3), 1.98–2.07 (m, 2H, H-17), 2.22–2.29 (m, 2H, H-8), 4.03 (m, 3H, H-6, N-H), 4.37 (m, 2H, H-12), 4.43–4.48 (m, 1H, H-14), 4.90–5.10 (m, 4H, H-10, H-19), 5.67–5.75 (m, 1H, H-18), 5.80–5.90 (m, 1H, H-9), 9.30 (s, 1H, N-H); ^{13}C NMR (150 MHz, CDCl_3): $\delta=7.5$ (+, CH_3), 14.5 (+, C-13), 22.6 (+, $(\text{CH}_3)_3$), 25.2 (–, C-16), 29.4 (–, C-7), 30.2 (–, C-8), 33.2 (–, C-17), 35.7 (–, C-15), 52.4 (+, C-14), 55.8 (C_{quat} , C- $(\text{CH}_3)_3$), 60.0 (–, C-12), 74.3 (–, C-6), 110.4 (C_{quat} , C-2), 110.7 (C_{quat} , C-4), 114.8 (–, C-10), 115.1 (–, C-19), 132.0 (C_{quat} , C-5), 137.9 (+, C-18), 138.2 (+, C-9), 149.9 (C_{quat} , C-3), 160.6 (C_{quat} , C-11); IR (neat): $\tilde{\nu}$ [cm^{-1}]=3458, 3252, 3075, 2976, 2928, 2866, 1665, 1468, 1273, 1032; MS (EI, 70 eV): m/z (%)=438.2 (6) [M^+], 318.1 (100) [$\text{M}^+-\text{NH}-(\text{S}=\text{O})(\text{CH}_3)_3$]; HRMS calcd for $\text{C}_{23}\text{H}_{38}\text{N}_2\text{O}_4\text{S}$ [M^+]: 438.2555; found: 438.2550 \pm 0.7 ppm.

3.3.6. (R)-Ethyl 5-(1-(1,1-dimethylethylsulfonamido)hept-6-enyl)-4-methyl-3-(pent-4-enyloxy)-1H-pyrrole-2-carboxylate (24)

Sulfinyl imine **17** (200 mg, 0.54 mmol) and hex-5-enylmagnesium bromide (1.27 mL, 0.94 M in Et_2O , 1.19 mmol) were reacted according to the GP2. Flash-silica chromatography (Biotage SP4 chromatography system, $\text{EtOAc}/\text{cyclohexane}=7:3$) gave 159 mg (65%) of **24** in a diastereomeric *syn/anti* ratio of 93:7. The diastereomeric ratio was determined by ^1H NMR. ^1H NMR (600 MHz, CDCl_3): $\delta=1.21$ (s, 9H, $(\text{CH}_3)_3$), 1.22–1.41 (m, 7H, H-13, H-16, H-17), 1.77–1.91 (m, 4H, H-7, H-15), 1.95 (s, 3H, CH_3), 1.98–2.04 (m, 2H, H-18), 2.21–2.32 (m, 2H, H-8), 3.77–3.88 (m, 1H, N-H), 3.94–4.05 (m, 2H, H-6), 4.23–4.38 (m, 2H, H-12), 4.41–4.51 (m, 1H, H-14), 4.88–5.09 (m, 4H, H-10, H-20), 5.69–5.78 (m, 1H, H-19), 5.82–5.92 (m, 1H, H-9), 9.23 (br s, 1H, N-H); ^{13}C NMR (150 MHz, CDCl_3): $\delta=7.5$ (+, CH_3), 14.5 (+, C-13), 22.6 (+, $(\text{CH}_3)_3$), 25.4 (–, C-17), 28.4 (–, C-16), 29.4 (–, C-7), 30.2 (–, C-8), 33.4 (–, C-18), 36.1 (–, C-15), 52.4 (+, C-14), 55.8 (C_{quat} , C- $(\text{CH}_3)_3$), 60.0 (–, C-12), 74.3 (–, C-6), 110.4 (C_{quat} , C-2), 110.8 (C_{quat} , C-4), 114.6 (–, C-10), 114.8 (–, C-20), 132.0 (C_{quat} , C-5), 138.2 (+, C-19), 138.4 (+, C-9), 149.9 (C_{quat} , C-3), 160.6 (C_{quat} , C-11); IR (neat): $\tilde{\nu}$ [cm^{-1}]=3458, 3252, 3075, 2976, 2928, 2866, 1665, 1469, 1280, 1033, 910, 733; MS (FAB+): m/z (%)=453 (20) [MH^+], 332 (100) [$\text{M}-\text{NH}-(\text{S}=\text{O})(\text{CH}_3)_3^+$]; HRMS calcd for $\text{C}_{24}\text{H}_{41}\text{N}_2\text{O}_4\text{S}$ [M^+]: 453.2787; found: 453.2813+10 ppm.

3.3.7. (R)-Ethyl 5-(1-(1,1-dimethylethylsulfonamido)pent-4-enyl)-3-(hex-5-enyloxy)-4-methyl-1H-pyrrole-2-carboxylate (25)

Sulfinyl imine **18** (63 mg, 0.14 mmol) and but-3-enylmagnesium bromide (0.33 mL, 1.3 M in Et_2O , 0.41 mmol) were

[†] Coupling constants smaller than 2 Hz are not precise and are therefore not documented.

reacted according to the GP2. Flash-silica chromatography (Biotage SP4 chromatography system, EtOAc/cyclohexane=7:3) gave 48 mg (68%) of **25** in a diastereomeric *syn/anti* ratio of 95:5. The diastereomeric ratio was determined by ¹H NMR. ¹H NMR (400 MHz, CDCl₃): δ=1.23 (s, 9H, (CH₃)₃), 1.28 (t, ³J=7.1, 3H, H-14), 1.45–1.58 (m, 2H, H-8), 1.62–1.75 (m, 2H, H-7), 1.78–2.01 (m, 4H, H-16 and H-17), 1.88 (s, 3H, CH₃), 2.01–2.10 (m, 2H, H-9), 3.85 (d, ³J=2.9, 1H, N-H), 3.93 (t, ³J=6.8 2H, H-6), 4.13–4.29 (m, 2H, H-13), 4.36–4.48 (m, 2H, H-15), 4.75–5.07 (m, 4H, H-11 and H-19), 5.54–5.91 (m, 2H, H-10 and H-18), 8.15 (br s, 1H, pyrrole-H); ¹³C NMR (100 Hz CDCl₃): δ=7.5 (+, CH₃), 14.5 (+, C-14), 22.6 (+, (CH₃)₃), 25.3 (–, C-8), 29.7 (–, C-7), 30.1 (–, C-17), 33.6 (–, C-9), 35.2 (–, C-16), 52.0 (+, C-15), 55.8 (C_{quat}, C-(CH₃)₃), 60.0 (–, C-13), 74.8 (–, C-6), 110.9 (C_{quat}, C-2), 111.1 (C_{quat}, C-4), 114.9 (–, C-11), 115.8 (–, C-19), 131.6 (C_{quat}, C-5), 137.1 (+, C-18), 138.7 (+, C-10), 150.0 (C_{quat}, C-3), 160.6 (C_{quat}, C-12); IR (neat): $\tilde{\nu}$ [cm⁻¹]=3461, 3300, 2978, 2933, 2867, 1663, 1272, 1031, 909; MS (FAB+): *m/z* (%)=439 (26) [MH]⁺, 318 (100) [M–NH–(S=O)(CH₃)₃]⁺; HRMS calcd for C₂₃H₃₉O₄N₂S: 439.2631; found: 439.2614+3.9 ppm.

3.3.8. (R)-Ethyl 5-(1-(1,1-dimethylethylsulfonamido)hex-5-enyl)-3-(hex-5-enyloxy)-4-methyl-1H-pyrrole-2-carboxylate (**26**)

Sulfinyl imine **18** (63 mg, 0.14 mmol) and pent-4-enylmagnesium bromide (0.44 mL, 0.94 M in Et₂O, 0.41 mmol) were reacted according to the GP2. Flash-silica chromatography (Biotage SP4 chromatography system, EtOAc/cyclohexane=7:3) gave 36 mg (50%) of **26** in a diastereomeric *syn/anti* ratio of 91:9. The diastereomeric ratio was determined by ¹H NMR. ¹H NMR (400 MHz, CDCl₃): δ=1.14 (s, 9H, (CH₃)₃), 1.18–1.41 (m, 2H, H-17), 1.28 (t, ³J=7.1, 3H, H-14), 1.46–1.57 (m, 2H, H-8), 1.62–1.84 (m, 4H, H-7 and H-16), 1.88 (s, 3H, CH₃), 1.91–2.00 (m, 2H, H-18), 2.01–2.10 (m, 2H, H-9), 3.62 (d, ³J=2.5, 1H, N-H), 3.88–3.98 (m, 2H, H-6), 4.15–4.29 (m, 2H, H-13), 4.36–4.45 (m, 1H, H-15), 4.85–5.99 (m, 4H, H-11 and H-20), 5.59–5.82 (m, 2H, H-10 and H-19), 9.03 (br s, 1H, pyrrole-H); ¹³C NMR (100 MHz, CDCl₃): δ=7.5 (+, CH₃), 14.5 (+, C-14), 22.6 (+, (CH₃)₃), 25.2 (–, C-17), 25.3 (–, C-8), 29.7 (–, C-7), 33.2 (–, C-18), 33.6 (–, C-9), 35.6 (–, C-16), 52.2 (+, C-15), 55.8 (C_{quat}, C-(CH₃)₃), 60.0 (–, C-13), 74.8 (–, C-6), 110.5 (C_{quat}, C-2), 110.9 (C_{quat}, C-4), 114.6 (–, C-11), 115.2 (–, C-20), 131.6 (C_{quat}, C-5), 137.9 (+, C-19), 138.7 (+, C-10), 150.0 (C_{quat}, C-3), 160.5 (C_{quat}, C-12); IR (neat): $\tilde{\nu}$ [cm⁻¹]=3420, 3257, 2978, 2931, 2864, 1664, 1468, 1273, 1032; MS (FAB+): *m/z* (%)=453 (20) [MH]⁺, 332 (100) [M–NH–(S=O)(CH₃)₃]⁺; HRMS calcd for C₂₄H₄₁O₄N₂S: 453.2787; found: 453.2807+4.5 ppm.

3.3.9. (R)-Ethyl 5-(1-(1,1-dimethylethylsulfonamido)hept-6-enyl)-3-(hex-5-enyloxy)-4-methyl-1H-pyrrole-2-carboxylate (**27**)

Sulfinyl imine **18** (63 mg, 0.16 mmol) and hex-5-enylmagnesium bromide (0.44 mL, 0.94 M in Et₂O, 0.41 mmol) were

reacted according to the GP2. Flash-silica chromatography (Biotage SP4 chromatography system, EtOAc/cyclohexane=7:3) gave 36 mg (72%) of **21** in a diastereomeric *syn/anti* ratio of 94:6. The diastereomeric ratio was determined by proton NMR. ¹H NMR (400 MHz, CDCl₃): δ=1.13 (s, 9H, (CH₃)₃), 1.19–1.45 (m, 4H, H-17 and 18), 1.28 (t, ³J=7.1, 3H, H-14), 1.46–1.58 (m, 2H, H-8), 1.60–1.84 (m, 4H, H-7 and H-16), 1.88 (s, 3H, CH₃), 1.90–2.11 (m, 4H, H-9 and H-19), 3.72 (d, ³J=2.9, 1H, N-H), 3.86–3.98 (m, 2H, H-6), 4.16–4.31 (m, 2H, H-13), 4.33–4.43 (m, 1H, H-15), 4.80–5.01 (m, 4H, H-11 and H-21), 5.59–5.83 (m, 2H, H-10 and H-20), 9.17 (br s, 1H, pyrrole-H); ¹³C NMR (100 MHz, CDCl₃): δ=7.5 (+, CH₃), 14.5 (+, C-14), 22.6 (+, (CH₃)₃), 25.3 (–, C-8), 25.4 (–, C-18), 28.4 (–, C-17), 29.7 (–, C-7), 33.4 (–, C-19), 33.7 (–, C-9), 36.1 (–, C-16), 52.3 (+, C-15), 55.8 (C_{quat}, C-(CH₃)₃), 59.9 (–, C-13), 74.8 (–, C-6), 110.4 (C_{quat}, C-2), 110.8 (C_{quat}, C-4), 114.2 (–, C-11), 114.6 (–, C-21), 131.9 (C_{quat}, C-5), 138.4 (+, C-20), 138.7 (+, C-10), 150.0 (C_{quat}, C-3), 160.6 (C_{quat}, C-12); IR (neat): $\tilde{\nu}$ [cm⁻¹]=3460, 3256, 3076, 2977, 2930, 2861, 1666, 1469, 1280, 1032, 994; MS (FAB+): *m/z* (%)=467 (18) [MH]⁺, 346 (100) [M–NH–(S=O)(CH₃)₃]⁺; HRMS calcd for C₂₅H₄₂O₄N₂S: 466.2865; found: 466.2849+3.5 ppm.

3.4. (R)-1-tert-Butyl 2-ethyl 5-(1-(1,1-dimethylethylsulfonamido)hex-5-enyl)-4-methyl-3-(pent-4-enyloxy)-1H-pyrrole-1,2-dicarboxylate **23-Boc** and (R)-ethyl 5-(1-(N-(tert-butoxycarbonyl)-2-methylpropan-2-ylsulfonamido)hex-5-enyl)-4-methyl-3-(pent-4-enyloxy)-1H-pyrrole-2-carboxylate **23a-Boc**

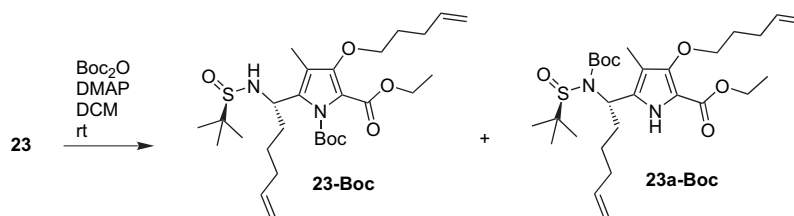
To a solution of **23** (130 mg, 0.30 mmol) in dry CH₂Cl₂ (20 mL) at room temperature DMAP (40.3 mg, 0.33 mmol) was added and the mixture was stirred for 10 min under dinitrogen atmosphere. (Boc)₂O (131 mg, 0.60 mmol) was added and the mixture was stirred overnight (Scheme 4). The solution was washed with water (4×10 mL) and dried over MgSO₄. The solvent was evaporated in vacuum and the two products were separated by column chromatography on silica gel (cyclohexane/EtOAc=8:2, R_f=0.13 and 0.47) to give 44 mg of **23-Boc** (29%) as a colourless oil and 99 mg of **23a-Boc** (61%) as a colourless oil.

3.4.1. Compound **23-Boc**

IR: $\tilde{\nu}$ [cm⁻¹]=3450, 3302, 3077, 2979, 2934, 2870, 1713; MS (EI, 70 eV): *m/z* (%)=538.1 (0.2) [M⁺], 318.2 (100) [M⁺–NBoc–(S=O)(CH₃)₃]; HRMS calcd for C₂₈H₄₆N₂O₆S [M⁺]: 538.3077; found: 538.3075±1.2 ppm.

3.4.2. Compound **23a-Boc**

¹H NMR (600 MHz, CDCl₃): δ=1.15–1.23 (m, 2H, H-16), 1.27 (s, 9H, TBS), 1.34 (t, ³J=7.1, 3H, H-13), 1.52 (s, 9H, Boc), 1.69–1.79 (m, 1H, H-15A), 1.80–1.87 (m, 2H, H-7), 1.95–2.05 (m, 2H, H-17), 2.02 (s, 3H, CH₃), 2.19–2.29 (m, 2H, H-8), 2.67–2.80 (m, 1H, H-15B), 3.95–4.00 (m, 1H, H-6A), 4.01–4.07 (m, 1H, H-6B), 4.23–4.36 (m, 2H, H-12), 4.57–4.65 (m, 1H, H-14), 4.89–5.11 (m, 4H, H-10, H-19),



Scheme 4. Preparation of the Boc-protected diene **23**.

5.71 (dddd, $^3J=17.0$, $^3J=10.3$, $^3J=6.7$, $^3J=6.7$, 1H, H-18), 5.86 (dddd, $^3J=17.0$, $^3J=10.3$, $^3J=6.7$, $^3J=6.7$, 1H, H-9), 9.70 (br s, 1H, H-pyrrole); ^{13}C NMR (75 MHz, CDCl_3): $\delta=7.4$ (+, CH_3), 14.5 (+, C-13), 22.8 (+, TBS), 26.3 (–, C-16), 28.3 (+, Boc), 29.5 (–, C-7), 30.2 (–, C-8), 33.2 (–, C-17), 33.5 (–, C-15), 45.5 (+, C-14), 59.6 (–, C-12), 60.4 (C_{quat} , TBS), 74.2 (–, C-6), 83.7 (C_{quat} , Boc), 110.3 (C_{quat} , C-2), 112.5 (C_{quat} , C-4), 114.7 (–, C-10), 114.9 (–, C-19), 130.8 (C_{quat} , C-5), 137.9 (+, C-18), 138.4 (+, C-9), 149.2 (C_{quart} , C-3), 154.5 (C_{quart} , C-Boc amide), 160.6 (C_{quat} , C-11); IR: $\tilde{\nu}$ [cm^{-1}]=3450, 3397, 3077, 2978, 2932, 1694; MS (EI, 70 eV): m/z (%)=538.1 (7) [M^+], 482.1 (13) [$\text{M}-\text{C}_4\text{H}_8$] $^+$, 318.2 (94) [$\text{M}^+-\text{NH}-(\text{S}=\text{O})(\text{CH}_3)_3$], 276.1 (100) [$\text{M}^+-\text{NH}-(\text{S}=\text{O})(\text{CH}_3)_3-\text{C}_2\text{H}_2\text{O}$]; HRMS calcd for $\text{C}_{28}\text{H}_{46}\text{N}_2\text{O}_6\text{S}$ [M^+]: 538.3077; found: 538.3079 \pm 0.4 ppm.

3.5. General procedure (GP3) for ring-closing metathesis of sulfinyl imine dialkenes

Sulfinyl imine dienes were dissolved as a 0.0005 M solution in anhydrous and degassed CH_2Cl_2 under an argon atmosphere. Bis(tricyclohexylphosphine) benzylidene ruthenium(IV) dichloride (Grubbs I catalyst) (15 mol %), dissolved in anhydrous and degassed CH_2Cl_2 (50 mL), was added slowly over a period of 2 h by a syringe pump to the reaction mixture at 40 °C, and stirred for additional 24 h. Additional catalyst (15 mol %) was added and the mixture was refluxed for 1 d. After 50 equiv of DMSO was added, the mixture was stirred at room temperature for 1 d. Concentration under reduced pressure and filtration through a pad of silica afforded the crude product, which was purified by HPLC.

3.5.1. Ethyl 11-((R)-1,1-dimethylethylsulfonamido)-15-methyl-2-oxa-13-azabicyclo[10.2.1]pentadecan-1(14),6,12(15)-triene-14-carboxylate (**28**)

Compound **23** (20 mg, 0.05 mmol) was reacted following the GP3. Preparative HPLC afforded 7 mg of the ansa-bridged **28** (47%, conversion corrected yield, 5 mg of **23** regained) as an inseparable 9:1 *E/Z* mixture and 3.5 mg of the larger macrocycle **29** (12%) as a mixture of stereoisomers.

3.5.2. Compound **28**

^1H NMR (600 MHz, CDCl_3 , DQF-COSY, HSQC, HMBC): $\delta=0.84$ – 0.92 (m, 1H, H-8A), 0.92 – 1.02 (m, 1H, H-11A), 1.26 (s, 9H, $(\text{CH}_3)_3$), 1.38 (t, 3H, $^3J=7.13$, H-16), 1.43–1.53 (m, 2H, H-7A, H-8B), 1.55–1.63 (m, 1H, H-13A), 1.65–1.75 (m, 2H, H-11B, H-12A), 2.00–2.13 (m, 4H, H-7B,

CH_3), 2.15–2.21 (m, 2H, H-12B, H-13B), 3.62 (br s, 1H, N-H), 4.13–4.20 (m, 1H, H-6A), 4.24–4.34 (m, 2H, H-6B, H-15A), 4.43–4.49 (m, 1H, H-15B), 4.54 (dd, 1H, $^2J=11.96$, $^3J=5.26$, H-14), 5.13 (ddt, 1H, $^3J=10.8$, $^3J=3.3$, $^4J=2.0$, H-10), 5.21 (ddt, 1H, $^3J=10.8$, $^3J=2.2$, $^4J=2.2$, H-9), 9.01 (br s, 1H, H-pyrrole); ^{13}C NMR (assignment by HSQC, HMBC, 600 MHz, CDCl_3): $\delta=9.1$ (+, CH_3), 14.5 (–, C-16), 22.6 (+, $(\text{CH}_3)_3$), 24.9 (–, C-11), 25.5 (–, C-12), 26.5 (–, C-8), 31.1 (–, C-7), 35.2 (–, C-13), 51.8 (+, C-14), 55.6 (C_{quat} , $\text{C}(\text{CH}_3)_3$), 59.9 (–, C-15), 70.9 (–, C-6), 111.8 (C_{quat} , C-2), 120.0 (C_{quat} , C-4), 128.5 (+, C-10), 130.2 (+, C-9), 131.0 (C_{quat} , C-5), 150.0 (C_{quat} , C-3), 160.6 (C_{quat} , C-ester); MS (ES, $\text{DCM}/\text{MeOH}+10$ mmol/L NH_4Ac): m/z (%)=411.2 (100) [$\text{M}+\text{H}$] $^+$, 821.6 (13) [$2\text{M}+\text{H}$] $^+$; HRMS calcd for $\text{C}_{21}\text{H}_{34}\text{N}_2\text{O}_4\text{S}$ [M^+]: 410.2232; found: 410.2233 \pm 1.5 ppm.

3.5.3. Compound **29**

MS (ES, $\text{DCM}/\text{MeOH}+10$ mmol/L NH_4Ac): m/z (%)=819.7 (100) [$\text{M}-\text{H}$] $^-$, 933.6 (10) [$\text{M}+\text{TFA}$] $^-$, 879.7 (9) [$\text{M}+\text{CH}_3\text{COO}$] $^-$, 855.6 (8) [$\text{M}+\text{Cl}$] $^-$.

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References and notes

1. Eliel, E.; Wilen, S. H. *Stereochemistry of Organic Compounds*; Wiley-Interscience: New York, NY, 1994.
2. Lüttringhaus, A.; Gralheer, H. *Liebigs Ann.* **1947**, 557, 112–120;
3. Lüttringhaus, A.; Gralheer, H. *Liebigs Ann.* **1941**, 550, 67–98.
4. For an example, see: Paley, R. S. *Chem. Rev.* **2002**, 102, 1493–1523; Paley, R. S.; Estroff, L. A.; Gauguier, J.-M.; Hunt, D. K.; Newlin, R. C. *Org. Lett.* **2000**, 2, 365–368; Comely, A. C.; Gibson, S. E. *J. Chem. Soc., Perkin Trans. 1* **1999**, 223–238.
5. For an example, see: Liu, D.; Xie, F.; Zhang, W. *Tetrahedron Lett.* **2007**, 48, 585–588; Djukic, J.-P.; Michon, C.; Maise-Francois, A.; Allagapen, R.; Pfeffer, M.; Dotz, K. H.; De Cian, A.; Fischer, J. *Chem.—Eur. J.* **2000**, 6, 1064–1077.
6. Tanaka, K.; Sagae, H.; Toyoda, K.; Noguchi, K.; Hirano, M. *J. Am. Chem. Soc.* **2007**, 129, 1522–1523; Aly, A. A.; Hopf, H.; Jones, P. G.; Dix, I.

- Tetrahedron* **2006**, *62*, 4498–4505; Rajakumar, P.; Swaroop, M. G. *Tetrahedron Lett.* **2006**, *47*, 3019–3022; Ueda, T.; Kanomata, N.; Machida, H. *Org. Lett.* **2005**, *7*, 2365–2368.
- For an example, see: Fürstner, A.; Szillat, H.; Gabor, B.; Mynott, R. *J. Am. Chem. Soc.* **1998**, *120*, 8305–8314.
 - Marchand, A.; Maxwell, A.; Mootoo, B.; Pelter, A.; Reid, A. *Tetrahedron* **2000**, *56*, 7331–7338; Pelter, A.; Kidwell, H.; Crump, R. A. N. C. *J. Chem. Soc., Perkin Trans. 1* **1997**, 3137–3140; Pelter, A.; Crump, R. A. N. C.; Kidwell, H. *Tetrahedron Lett.* **1996**, *37*, 1273–1276.
 - Pelter, A.; Crump, R. A. N. C.; Kidwell, H. *Tetrahedron: Asymmetry* **1997**, *8*, 3873–3880.
 - Handbook of Metathesis*; Grubbs, R. H., Ed.; Wiley-VCH: Weinheim, 2003; *Alkene Metathesis in Organic Synthesis*; Fürstner, A., Ed.; Springer: Berlin, 1998; Connon, S. J.; Blechert, S. *Angew. Chem., Int. Ed.* **2003**, *42*, 1900–1923; Grubbs, R. H.; Miller, S. J.; Fu, G. C. *Acc. Chem. Res.* **1995**, *28*, 446–452; Fürstner, A. *Angew. Chem., Int. Ed.* **2000**, *39*, 3012–3043.
 - For examples, see: El-azizi, Y.; Schmitzer, A.; Collins, S. K. *Angew. Chem., Int. Ed.* **2006**, *45*, 968–973; Watson, M. D.; Jaekel, F.; Severin, N.; Rabe, J. P.; Muellen, K. *J. Am. Chem. Soc.* **2004**, *126*, 1402–1407; Martinez, V.; Blais, J.-C.; Bravic, G.; Astruc, D. *Organometallics* **2004**, *23*, 861–874; Fürstner, A.; Stelzer, F.; Rumbo, A.; Krause, H. *Chem.—Eur. J.* **2002**, *8*, 1856–1871; Smith, A. B., III; Adams, C. M.; Kozmin, S. A. *J. Am. Chem. Soc.* **2001**, *123*, 990–991.
 - Fürstner, A.; Gastner, T.; Weintritt, H. *J. Org. Chem.* **1999**, *64*, 2361–2366.
 - The high temperature required for racemization does not permit determining the energy of the barrier experimentally, e.g., by variable temperature NMR measurements.
 - Hochmuth, D. H.; Koenig, W. A. *Liebigs Ann.* **1996**, 947–951.
 - Bonauer, C.; Walenzyk, T.; Koenig, B. *Synthesis* **2006**, 1–20; Bonauer, C.; Koenig, B. *Synthesis* **2005**, 2367–2372.
 - Bonauer, C.; Zabel, M.; Koenig, B. *Org. Lett.* **2004**, *6*, 1349–1352.
 - Maity, P.; Koenig, B. *Synthesis* **2006**, 2719–2724.
 - Chong, R.; Clezy, P. S. *Aust. J. Chem.* **1967**, *20*, 935–950.
 - Liu, G.; Cogan, D. A.; Owens, T. D.; Tang, T. P.; Ellman, J. A. *J. Org. Chem.* **1999**, *64*, 1278–1284.
 - Jiang, Z.-Y.; Chan, W. H.; Lee, A. W. M. *J. Org. Chem.* **2005**, *70*, 1081–1083.
 - McMahon, J. P.; Ellman, J. A. *Org. Lett.* **2004**, *6*, 1645–1647; Mukade, T.; Dragoli, D. R.; Ellman, J. A. *J. Comb. Chem.* **2003**, *5*, 590–596; Evans, J. W.; Ellman, J. A. *J. Org. Chem.* **2003**, *68*, 9948–9957; Plobeck, N.; Powell, D. *Tetrahedron: Asymmetry* **2002**, *13*, 303–310; Tang, T. P.; Volkman, S. K.; Ellman, J. A. *J. Org. Chem.* **2001**, *66*, 8772–8778; Bailey, P. D.; Clayson, J.; Boa, A. N. *Contemp. Org. Synth.* **1995**, *2*, 173–187.
 - Armstrong, S. K. *J. Chem. Soc., Perkin Trans. 1* **1998**, 371–388; Grubbs, R. H.; Chang, S. *Tetrahedron* **1998**, *54*, 4413–4450.
 - The *N*-Boc-protected derivative **23-Boc** was tested to explore the effect of the pyrrole amine function on the RCM reaction. However, reactions with and without *N*-Boc protection of the pyrrole nitrogen atom gave equal results.
 - This catalyst was tested to prevent the possible formation of a chelate intermediate between the *tert*-butyldimethylsilyl (TBS) protecting group and the ruthenium ion trapping the catalyst in an inactive form. This chelation should be avoided by the chelating effect of the isopropoxy-group of the second generation Hoveyda–Grubb's catalyst: Zaja, M.; Connon, S. J.; Dunne, A. M.; Rivard, M.; Buschmann, N.; Jiricek, J.; Blechert, S. *Tetrahedron* **2003**, *59*, 6545–6558.
 - Cross metathesis processes generally lead to the thermodynamically most stable members of a set of structurally related isomers: Smith, A. B., III; Adams, C. M.; Kozmin, S. A.; Paone, D. V. *J. Am. Chem. Soc.* **2001**, *123*, 5925–5937.
 - The prediction of alkene geometry in ring-closing ene–ene metathesis is still difficult: Arisawa, M.; Nishida, A.; Nakagawa, M. *J. Organomet. Chem.* **2006**, *691*, 5109–5121; Fürstner, A.; Dierkes, T.; Thiel, O. R.; Blanda, G. *Chem.—Eur. J.* **2001**, *7*, 5286–5298; Fürstner, A.; Thiel, O. R.; Ackermann, L. *Org. Lett.* **2001**, *3*, 449–451; Lee, C. W.; Grubbs, R. H. *Org. Lett.* **2000**, *2*, 2145–2147; Fürstner, A.; Thiel, O. R.; Kindler, N.; Bartkowska, B. *J. Org. Chem.* **2000**, *65*, 7990–7995; Fürstner, A.; Thiel, O. R.; Blanda, G. *Org. Lett.* **2000**, *2*, 3731–3734.
 - (a) Lam, P. C.-H.; Carlier, P. R. *J. Org. Chem.* **2005**, *70*, 1530–1538; (b) Mitchell, R. H.; Blunden, R.; Hollett, G.; Bandyopadhyay, S.; Vaughan Williams, R.; Twamley, B. *J. Org. Chem.* **2005**, *70*, 675–680.
 - HPLC analysis of the RCM product from diene **23-Boc** on a chiral column showed four isomeric compounds (one major, three minor). This again indicates a stereochemical bias of the ring-closing reaction.
 - Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Montgomery, J. A., Jr.; Vreven, T.; Kudin, K. N.; Burant, J. C.; Millam, J. M.; Iyengar, S. S.; Tomasi, J.; Barone, V.; Mennucci, B.; Cossi, M.; Scalmani, G.; Rega, N.; Petersson, G. A.; Nakatsuji, H.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Klene, M.; Li, X.; Knox, J. E.; Hratchian, H. P.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Ayala, P. Y.; Morokuma, K.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Zakrzewski, V. G.; Dapprich, S.; Daniels, A. D.; Strain, M. C.; Farkas, O.; Malick, D. K.; Rabuck, A. D.; Raghavachari, K.; Foresman, J. B.; Ortiz, J. V.; Cui, Q.; Baboul, A. G.; Clifford, S.; Cioslowski, J.; Stefanov, B. B.; Liu, G.; Liashenko, A.; Piskorz, P.; Komaromi, I.; Martin, R. L.; Fox, D. J.; Keith, T.; Al-Laham, M. A.; Peng, C. Y.; Nanayakkara, A.; Challacombe, M.; Gill, P. M. W.; Johnson, B.; Chen, W.; Wong, M. W.; Gonzalez, C.; Pople, J. A. *Gaussian 03, Revision D.02*; Gaussian: Wallingford, CT, 2004.
 - (a) Becke, A. D. *Phys. Rev. A* **1988**, *38*, 3098–3100; (b) Becke, A. D. *J. Chem. Phys.* **1993**, *98*, 5648–5652. The Lee–Yang–Parr nonlocal correlation functional: (c) Lee, C.; Yang, W.; Parr, R. G. *Phys. Rev. B* **1988**, *37*, 785–789.
 - Hehre, W. J.; Ditchfield, R.; Pople, J. A. *J. Chem. Phys.* **1972**, *56*, 2257–2261.