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# Modeling the Voltage Drop Across the Cathode Sheath in HPS

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Abstract-An electrical cathode model (ECM) of a highpressure sodium (HPS) lamp based on physical laws has been developed. The proposed ECM calculates the instantaneous voltage drop in a cathode sheath and the temperature distribution inside the cathode using as input parameter the cathode geometry and the positive column current. The model is based on the electrode heat transport equation, which is solved using the finite-element method. So, since it stems from the physics describing the cathode behavior, it is found that the ECM predicts in a satisfactory way the cathode voltage drop over a wide range of work conditions. The obtained results were compared with those reported in the current literature. It can be concluded that the ECM is a useful tool in understanding the interaction between the positive column and the cathode in order to improve, for instance, HPS lamp ballast designs, as measuring the potential drop across the cathode sheath using a commercial lamp is particularly difficult. The model takes into account the temperature dependence of the thermal conductivity, the electrical cathode resistivity, and the total emissivity.

*Index Terms*—Cathode temperature distribution, electrical cathode model (ECM), finite-element method, high-pressure sodium (HPS) lamps.

#### I. INTRODUCTION

THE ELECTRODES of a gas discharge lamp act as an interface between the positive column and the electronic circuit. An important fraction of the total lamp input power is lost by dissipation through the electrodes. Therefore, it is essential to understand the interaction between electrodes and positive column in order to improve performance by reducing the electrode losses.

Since high-pressure sodium (HPS) discharges normally work under an alternating current, each electrode acts as both anode and cathode during each period. From an electrical point of view, the voltage drop at the anode is negligible with respect

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to that at the cathode [1]. Therefore, by using only a cathode model, it is possible to obtain a reasonable representation of the electrical interaction between electrodes and positive column, with a considerable reduction of complexity.

The experimental study of the relationship between the cathode and positive column is not particularly easy because it necessitates a sophisticated experimental setup to take the measurements. Consequently, numerical modeling of the plasma–cathode interaction represents an option to carry out this study. In this paper, an electrical cathode model (ECM) is proposed to predict the voltage drop across the cathode sheath under different work conditions. Using the ECM, it is possible to study the cathode operation. The ECM is based on the equation of the electrode power at the front of the rod that was proposed by de Groot and Van Vliet [2]. This equation relates the temperature gradient at the front of the electrode (active surface) with the electrical power losses in the electrode.

The aim of this paper is to present a numerical cathode model for an HPS discharge that is able to predict the cathode voltage drop at different frequencies and supply current waveforms. The model was developed in Matlab language applying the finite-element method. Since this model is based on the physical equations that decree the cathode behavior, it is valid over a wide range of operation conditions. The ECM user only needs to enter the amplitude, frequency, and waveform of the positive column current, and for a given cathode geometry, the model predicts the instantaneous voltage drop across the cathode sheath. Besides, the ECM enables the determination of the temperature distribution inside the cathode. This is of great interest because the lifetime of the HPS lamps is strongly related to it [3].

In order to determine the electrical losses at the cathode, it is necessary to calculate the temperature gradient on the active surface (Section III-A) and the temperature distribution through the cathode (Section III-B). The temperature distribution in the electrode at a given time can be calculated by means of the electrode heat transport equation (Section III-B). Using the cathode temperature distribution, it has been possible to obtain the temperature gradient in the active surface and, consequently, the electrode power losses (Section IV-C).

#### II. CATHODE GEOMETRY

The cathode of an HPS discharge lamp is usually manufactured from a refractory material having a high melting point that is meant to support high temperatures of operation. The cathode geometry of an HPS lamp discharge is shown in Fig. 1.

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Fig. 1. Longitudinal cross section of a typical electrode used in HPS lamps.

With an aim to increase the emission surface, the HPS lamp cathodes are composed of a tungsten rod wrapped in one or two coils of the same material. A mixture that is composed of oxides (generally, BaO or CaO) and carbonates (emitter materials such as BaCO<sub>3</sub> or CaCO<sub>3</sub>) is deposited between the coils in order to increase the electron emission. The reaction of the oxides with the tungsten, i.e.,  $6BaO + W \rightarrow Ba_3WO_6 +$ 3Ba [2], generates the required free barium for the formation of a monolayer that is capable of reducing the electrode work function in such a way that the operation of the lamp is improved.

#### **III. CATHODE MODELING**

After the definition of Groot and van Vlient [2], the electrical losses  $P_{el}$  at the front of the cathode (y = 0) are defined by

$$P_{el} = \frac{1}{4} \pi (d_e)^{3/2} k \left(\frac{dT}{dy}\right)_{y=0}$$
(1)

where  $d_e$  symbolizes diameter of the rode-type electrode, k is the tungsten thermal conductivity, T is the electrode temperature, and  $y = z(d_e^{-1/2})$  is the reduced axial coordinate, with z equal to the axial coordinate. From (1), it is possible to appreciate that the voltage drop at the front of the cathode is related to the temperature gradient at its active surface. So, in order to calculate the electric losses at the cathode sheath, it is necessary to obtain the temperature distribution through the cathode, based on the cathode boundary conditions. Once the cathode temperature distribution is known, then it is possible to specify the temperature gradient at the active surface.

#### A. Temperature Calculation at the Cathode Active Surface

To remove an electron from a metallic surface, it is necessary that the electron absorbs enough energy from an external source to escape from the electric potential barrier that originally confined it. This energy is called the work function. The Fermi level is the top of all possible electron energy levels at absolute zero temperature. If a metal is warmed up, its electrons exceed the Fermi level, and the energy is distributed according to the Fermi–Dirac statistic. With a higher temperature of the metal, the number of electrons with sufficient energy to scale the potential barrier and leave the metal is increased.

The process briefly described here is called thermionic emission. In the cathode of an HPS discharge, the thermionic emission is enhanced by the electric field applied and by the presence of an activator, which lowers the work function of the electrode surface.

It has been assumed in this paper that the local discharge current density in the cathode active surface j is  $j = j_i + j_e$ , where  $j_i$  represents the ionic current density and  $j_e$  is the electron current density composed by the thermoionic emission  $j_e^{th}$  [described by the equation of Richardson–Dushman (4)] corrected by the electric field and the secondary electron emission  $j_e^{sec}$  related to the  $\gamma$ -Townsend process  $j_e = j_e^{th} + j_e^{sec}$ . According to [4], the emission of secondary electrons due to the bombardment of sodium ions over a tungsten cathode is equal to 1% of the ions that reach the cathode. Using the  $\beta$ -Waymouth coefficient, which relates the ionic  $j_i$  and electronic  $j_e$  currents at the cathode sheath of a HPS discharge [5], [6], we have

$$\beta = \frac{j_i}{j_e} = \frac{j_i}{j_e^{th} + j_e^{\text{sec}}} \tag{2}$$

$$j_e^{\text{sec}} = \gamma j_i \tag{3}$$

$$j_e^{th} = AT^2 \exp\left(\frac{e\varphi(E_k)}{k_B T}\right) \tag{4}$$

where  $A = (4\pi e m_e k_B^2)/h^3 = 1.2 \times 10^6 \text{ Am}^{-2} \text{ K}^{-2}$  represents the Richardson constant [6]–[8], e is the electron charge,  $k_B$  is the Boltzmann constant, and  $E_k$  corresponds to the electric field on the cathode sheath, which is related to the potential drop at the cathode sheath  $V_k$  in the MacKeown equation [9]  $E_k^2 \approx (4j_i/\varepsilon_0)\sqrt{m_iV_k/2e}$ , where, in turn,  $m_i$  represents the sodium ionic mass,  $\varepsilon_0$  is equal to the vacuum permittivity, and

$$\varphi(E_k) = \varphi_0 - \Delta \varphi(E_k), \qquad \Delta \varphi(E_k) = \sqrt{\frac{eE_k}{4\pi\varepsilon_0}}.$$
 (5)

Here,  $\varphi_0$  is the tungsten work function, and  $\Delta \varphi$  symbolizes the Schottky correction for the tungsten work function.

In HID lamps, two modes of cathodic arc attachment have been observed mainly: 1) the diffuse mode and 2) the spot mode. The major difference between the diffuse and spot modes is the current density, which is low in the case of the diffuse mode and high in the spot mode and the area of the arc attachment. With the aim of calculate the temperature at the cathode active surface for an HPS discharge, it can be assumed that the mode of cathodic arc attachment is the diffuse mode [10]. When the cathode operates in diffuse mode, the plasma covers the active surface of the electrode and their sides. In this paper, only the active surface is considered in the electronic emission. The operation of the cathode in diffuse mode is favored by the presence of an electric field lower than ( $E < 10^7$  V/m), a current density ( $j < 10^8$  A/m<sup>2</sup>), a weak cooling of the electrode and a low discharge pressure.

Taking into account the previous arguments and substituting (2) and (3) into (4), it is possible to derive the following equation, which relate the total density of current to the temperature of the active surface of the cathode  $T_{act}$ :

$$j = \left(\frac{1+\beta}{1-\gamma\beta}\right) AT_{\rm act}^2 \exp\left(\frac{e\varphi(E_k)}{k_B T_{\rm act}}\right).$$
 (6)

Symbol	Value	Units	
ρ	19300	[kg m <sup>-3</sup> ]	
С	133	[J kg <sup>-1</sup> K <sup>-1</sup> ]	
k	776/(T <sup>0.256</sup> )	[W m <sup>-1</sup> K <sup>-1</sup> ]	
$ ho_0$	5.28×10 <sup>-8</sup>	[W m]	
α	$6.76 \times 10^{-3}$	[K <sup>-1</sup> ]	

TABLE I TUNGSTEN COEFFICIENTS

By solving the previous equation, it is possible to calculate the temperature in the cathode active surface. According to Waymouth [5], the value of  $\beta$  was considered as the 20% of the electrons current density. Using (6), we have confirmed that, changing  $\beta$  by 100%, the temperature at the active surface changes only 5%.

#### B. Heat Transport Equation of the Cathode

The temperature distribution inside the cathode body and at the surface may be found by means of solving the heat flux equation in the cathode body (7), with the prevailing boundary conditions (see Section III-C). The following equation describes the variation in the quantity of heat per unit of volume of the electrode [2], [6]:

$$\rho c \frac{\partial T}{\partial t} = S + \nabla (k \nabla T) \tag{7}$$

where  $\rho$  symbolizes the mass density, c is the specific heat capacity, T is the electrode temperature, t denotes the time dependent of the equation, and k represents the thermal conductivity of the tungsten. The term S from (7) represents the power that is supplied by the electric field to the electrode and is given by the Joule effect [6], i.e.,

$$S = \rho_e j^2. \tag{8}$$

Here, j represents the discharge current density, and  $\rho_e$  is the electrical conductivity of the electrode. The heat produced in the electrode material by the discharge current causes the rise in the electrode temperature and consequently  $\rho_e$  change; this change is given by the relation

$$\rho_e(T) = \rho_0 \left\{ 1 + \alpha [T - T_0] \right\}.$$
(9)

Now, the constants  $\rho_0$  and  $\alpha$  depend on the cathode material. The values for tungsten are shown in Table I.  $T_0$  is the active surface temperature. The following condition for the total discharge current must be verified:

$$2\pi \int_{0}^{r_{0}} jr \, dr = I \tag{10}$$

where  $r_0 = d_e/2$  is the radius of the cathode, r is the radial coordinate, and I is the discharge current.



Fig. 2. Structured solution to compute the cathode voltage drop.

#### C. Boundary Conditions

The front (active surface) and end of the cathode were characterized by the Dirichlet boundary condition. The temperature at the end of the cathode was experimentally measured [6] and taken as  $T_{end} = 600$  K, and active surface temperature  $T_{act}$  is obtained by solving (6). The cathode boundary conditions on the lateral borders can be classified as the Neumann condition; the physical meaning of this condition is a continuous energy flow (radiation, conduction, and convection) between the cathode and the adjacent medium. The equation that defines the boundary conditions is [11]

$$k(T_b) \left(\frac{\partial T}{\partial \overline{n}}\right)_b = \overline{n} \left[\zeta(T_b - T_{\rm amb}) + \varepsilon_w(T_b)\sigma\left(T_b^4 - T_{\rm amb}^4\right)\right]$$
(11)

where  $T_b$  represents the temperature on the cathode border,  $T_{\rm amb}$  is the ambient temperature,  $\zeta$  is the convection transport coefficient,  $\varepsilon_w$  is the tungsten emissivity, and  $\sigma$  symbolizes the Stefan Boltzmann constant. The term  $(\partial T/\partial \bar{n})_b$  means the temperature derivative in the direction  $\bar{n}$  normal to the cathode surface.

#### **IV. RESULT AND DISCUSSION**

The flowchart represented in Fig. 2 shows a summary of the procedure used to calculate the cathode voltage drop. In this chart, the input and output parameters are represented by rounded rectangles, so as to differentiate them from the processing steps.

As we can see, in order to obtain the temperature at the electrode active surface, it is necessary to know the initial cathode voltage drop and the discharge current. With the obtained temperature, it is possible to calculate the temperature



Fig. 3. Graphic solution for the temperature at the cathode active surface using a discharge current density of  $j = 1.2732 \times 10^{6}$  Am<sup>-2</sup>.



Fig. 4. Temperature distribution (gray scale) and heat flow (arrows) in an HPS discharge cathode.

distribution through the electrode, and following the next steps, a new cathode voltage drop can be obtained. The process is repeated until the program output converges.

Equation (7) was solved by the Matlab finite-element method [12]. The characteristics of the employed cathode were the following (see Fig. 1):  $d_e = 1 \text{ mm}, r_2 = 1.5 \text{ mm}, r_3 = 1.0 \text{ mm}, d_1 = 1.5 \text{ mm}, d_2 = 3 \text{ mm}, \text{ and } d_3 = 5 \text{ mm}.$  The resulting values are presented here.

#### A. Temperature Calculation at the Active Surface of the Cathode

This temperature is obtained by solving (6). The solution was obtained by means of an iterative process that changes the active surface temperature until converging with the discharge current density. In other words, the solution of (6) is a temperature that produces an electric current density equal to the discharge. The results for a discharge current density of  $1.2732\times 10^6~\text{A/m}^2$  in an HPS lamp (400 W) are shown in Fig. 3. The temperature obtained at the active surface reaches 1650 K.

#### B. Temperature Distribution Through the Cathode

The temperature distribution through the cathode of an HPS discharge for an active surface temperature  $T_{\rm act} = 1650$  K is exhibited in Fig. 4, which corresponds to a 4-A discharge current. It can be observed that the temperature distribution has axial symmetry. Considering that the emissive layer consumption rate in diffuse mode is the same in all points, the cathode will work appropriately until the depletion of the emissive layer takes place. The temperature distribution inside the cathode is of great interest because the lifetime of the HPS lamps is strongly related to it [7].



Fig. 5. Temperature gradient at the front of the electrode using a sinusoidal current of 4 A.



Fig. 6. Diagram of the proposed ECM.

#### C. Temperature Gradient at the Rod Front

By knowing the temperature distribution in the cathode, the temperature gradient at the front of the rod can be deduced. The temperature distribution profile through the cathode center is displayed in Fig. 5. At the front of the electrode, the temperature is 1650 K; nevertheless, when we approached the back of the cathode, the temperature decreases until 600 K. The temperature decrease is due to the cathode conduction [(7) and (11)] and radiation losses defined by (11). The temperature gradient at the front of the electrode is also shown in Fig. 5. The temperature profile was compared with references [6], [13], and [14], proving to be in close agreement.

#### D. Cathode Voltage Drop for Different Frequencies

By using the temperature gradient obtained in the previous section, the electrical power (1) can be established, and consequently, the cathode voltage drop can be determined. The procedure adopted in this paper for assessing the influence of the frequency and waveform of the discharge current supply in the cathode operation is summarized in Fig. 6. We apply a current source I with variable frequency f and waveform wf to the ECM. The XY cathode V-I characteristics of an HPS (400 W) discharge using a sinusoidal current at different frequencies are shown in Fig. 7. Notice that the hysteresis phenomenon appears mainly at low frequencies because the cathode temperature modulation is greater and an extra energy is necessary in each cycle to reach the work temperature.

According to Fig. 7, increasing the sinusoidal current frequency, the hysteresis at the electric cathode characteristics is reduced. This can be translated into a modulation decrease of the discharge impedance and better electronic control. To validate the model, some typical results obtained for tungsten cathodes in HPS discharges are reported in Table II.

The peak cathode voltage drop for an HPS lamp (400 W) is around 20 V when the discharge current is 4 A [5]. The differences appearing are probably due to the simplifying assumptions considered in the present model.

#### V. CONCLUSION

An electrical model for the cathode of an HPS discharge intended for lamp ballast simulation has been developed in this paper. The model was based on the heat transfer equation for the electrode and implemented in Matlab using the finiteelement method. Since the model is built on the physical equations that define the cathode behavior over a wide range of operation conditions, the ECM is able to predict the electrical behavior of the cathode for different frequencies and supply current waveforms. The experimental study of the interaction between the cathode and positive column is a difficult task to carry out; because it necessitates a sophisticated experimental setup to make the measurements, consequently, the ECM model represents an excellent alternative. The ECM user only needs to supply the amplitude, frequency, and waveform of the positive column current, and for a given cathode geometry, the model predicts the instantaneous drop of voltage at the cathode sheath. Furthermore, the ECM enables to know the temperature distribution inside the cathode. This is of a great interest provided that the lifetime of HPS lamps is strongly related to it.

The obtained results using the model have been compared with the current literature, finding a good agreement, even at different work conditions. Thus, we concluded that the ECM is a useful tool in understanding the interaction between the positive column and the cathode in order to improve ballast design.



Fig. 7. Current (in amperes)-voltage (in volts) characteristics on the cathode of an HPS discharge, using a sinusoidal current (4 A): (a) 60 Hz, (b) 1 kHz, (c) 10 kHz, and (d) 40 kHz.

 TABLE II

 Result Comparisons Obtained for Tungsten Cathodes in HPS Discharges

					cathode	cathode
Cathode Gas	Current [A]	Waveform	Frequency	voltage drop	voltage drop	
			[Hz.]	(peak)	(peak)	
					[V]	[V]
Tungsten	sodium	4.45	sinusoidal	60	17 [2]	16.4 ECM
Tungsten	sodium	4	sinusoidal	60	20 [5]	19.5 ECM

In addition, the cathode model here presented can be used to assess the influence of the frequency and waveform of the discharge current supply in the cathode operation. From the results shown in Fig. 7, we can conclude that, by increasing the frequency of the sinusoidal discharge current supply, the hysteresis at the electric cathode characteristics is reduced. This can be translated into a modulation decrease of the discharge impedance and better electronic control.

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