

HCOOH high resolution spectroscopy in the $9.18 \mu m$ region

Franck Bielsa, Khelifa Djerroud, Andrei Goncharov, Albane Douillet, Tristan Valenzuela, Christophe Daussy, Laurent Hilico, Anne Amy-Klein

▶ To cite this version:

Franck Bielsa, Khelifa Djerroud, Andrei Goncharov, Albane Douillet, Tristan Valenzuela, et al.. HCOOH high resolution spectroscopy in the $9.18 \mu m$ region. 2007. hal-00162635

HAL Id: hal-00162635 https://hal.science/hal-00162635v1

Preprint submitted on 15 Jul 2007

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

hal-00162635, version 1 - 15 Jul 2007

HCOOH high resolution spectroscopy in the $9.18\mu m$ region

Franck Bielsa,^{1,2} Khelifa Djerroud,³ Andrei Goncharov,^{3,4} Albane Douillet,^{1,2}

Tristan Valenzuela,^{1,2} Christophe Daussy,³ Laurent Hilico,^{1,2} and Anne Amy-Klein³

¹Laboratoire Kastler Brossel UMR 8552, EVRY, UPMC, CNRS, ENS, C 74, Université Pierre et Marie Curie

4 place Jussieu, 75252 Paris, France

²Département de Physique et Modélisation, Université d'Evry Val d'Essonne

Boulevard F. Mitterrand, 91025 Evry cedex

³Laboratoire de Physique des Lasers, UMR 7538 CNRS, Université Paris 13, 99 av. J.-B. Clément, 93430 Villetaneuse, France

⁴The Institute of Laser Physics, Siberian Branch of the Russian

Academy of Science, Pr. Lavrentyeva 13/3, 630090 Novosibirsk, Russia

(Dated: today)

We report on higly accurate absolute frequency measurement against a femtosecond frequency comb of 6 saturated absorption lines of formic acid (HCOOH) with an accuracy of 1 kHz. We also report the frequency measurement of 17 other lines with an accuracy of 2 kHz. Those lines are in quasi coincidence with the 9R(36) to 9R(42) CO₂ laser emission lines and are probed either by a CO₂ or a widely tunable quantum cascade laser phase locked to a master CO₂ laser. The relative stability of two HCOOH stabilized lasers is characterized by a relative Allan deviation of 4.5 $10^{-12} \tau^{-1/2}$. They give suitable frequency references for H₂⁺ Doppler free two-photon spectroscopy.

Keywords: CO_2 laser, quantum cascade laser, formic acid, saturated absorption, frequency measurement.

I. INTRODUCTION

Electron to proton mass ratio is a fundamental constant that is determined with a relative accuracy of 4.6 10^{-10} [1] from separate electron and proton mass measurements [2, 3]. Molecular hydrogen ion (H₂⁺ or HD⁺) vibrational spectroscopy has recently been proposed as a new tool for very accurate direct optical determination of the electron to proton mass ratio with an improved accuracy [4, 5, 6]. H₂⁺ high resolution vibrational spectroscopy is feasible using Doppler free two-photon transitions in the 9.1-9.2 μ m range (1087-1099 cm⁻¹), corresponding to the 9R(34)-9R(52) CO₂ laser emission lines. There is no close coïncidence between CO₂ laser lines and H₂⁺ two-photon lines. However, we have shown that the recently developped quantum cascade laser sources (QCL) are suitable for H₂⁺ spectroscopy. The relative accuracy level of interest on the vibrational frequency determination for mass ratio metrology purposes is 2.10^{-10} , that corresponds to 2. 10^{-7} cm⁻¹ (6 kHz), hence the need for highly accurate molecular frequency references in the 9.1-9.2 μ m range.

Recently, the 9 μ m band ${}^{12}C^{16}O_2$ absolute frequencies have been measured with an accuracy better than 1 kHz up to the 9R(36) line [7] using a CO₂ saturated absorption stabilized CO₂ laser. Because the 9 μ m band is a hot band and because the molecular population dramatically decrease with the angular quantum number, CO₂ saturated absorption signal to noise ratio becomes too low for laser stabilisation purposes beyond the 9R(36) line. Among the large variety of molecules (SF₆, OsO₄, NH₃, CO₂, HCOOH) which are well known to provide frequency references for the CO₂ lasers, formic acid ν_6 band is the only one that presents intense lines in the 9.1-9.2 μ m range [8]. As an example, Fig. 1 shows the (v=0,L=2) \rightarrow (v=1,L=2) H₂⁺ two-photon excitation spectrum and formic acid lines close to the 9R(42) CO₂ emission line. Through the paper, CO₂ and HCOOH denote the standard isotopes. At 300 K, HCOOH is in the *trans* conformation, the *cis* one representing only 0.1% of the molecules [9].

Line frequency measurements in the HCOOH ν_6 band have been performed using Fourier transform, diode laser, CO₂ laser, and laser-radiofrequency double resonance and microwave spectroscopy [10, 11, 12, 13, 14, 15]. Among them, the most accurate ones were obtained by CO₂ laser spectroscopy with accuracies in the 50-250 kHz range [11] for a small number of lines. Recently, absolute line intensities of the ν_6 and ν_8 band have been determined accurately [9]. When compared to theoretical models taking into account Coriolis coupling between ν_6 and ν_8 bands, those measurements lead to the determination of effective hamiltonian parameters and the prediction of HCOOH lines with an uncertainty of 10^{-6} cm⁻¹ (30 kHz) [8]. Those predictions have recently been improved, the line strengh being corrected by a factor 2 and the line frequency given with the same uncertainty.

The aim of this work is the measurement at the kHz accuracy level of molecular reference frequencies in quasi coincidence with the 9R(36) to 9R(42) CO₂ laser lines. We first describe the CO₂/HCOOH reference laser and the widely tunable QCL we have implemented. We then describe the frequency measurements and give the absolute frequencies of 23 HCOOH lines. Finally, we analyse the relative stability of two independent lasers locked to two different HCOOH lines.



FIG. 1: (a) Full lines : HCOOH lines of intensity larger than 10^{-23} cm⁻¹/(molec cm⁻²) from HITRAN database [8]. Dotted lines : additional lines predicted by A. Perrin [16]. The labels indicate the presently measured lines of Table I. When operated on the 9R(42) CO₂ line, the CO₂/HCOOH reference laser is stabilized on line 4.(b) (v=0,L=2) \rightarrow (v=1,L=2) H₂⁺ two-photon transition spectrum in arbitrary units.

II. THE LASER SOURCES

The experimental setup consists in a HCOOH stabilised CO_2 reference laser (see Fig. 2) and in a tunable QCL (see Fig. 3).

A. The $CO_2/HCOOH$ reference laser

The CO₂ laser is a sealed-off low pressure (13 Torr) dual-discharge-arm 1m-long laser emitting in a single longitudinal mode. The laser cavity is made of a 90-95% efficiency diffraction grating in the Littrow configuration and of a high reflectivity mirror at 9.2 μ m. The main output of the laser is the zeroth order of the grating. Laser oscillation is obtained up to the 9R(48) emission line of CO₂ at 9.142 μ m with more than 1W of optical power. The CO₂ laser frequency is stabilised onto a formic acid saturated absorption signal detected in transmission of a 1m long optical cavity of finesse 100 and beam waist 3.8 mm containing a low pressure HCOOH gaz (0.5-2 mTorr, 66-266 Pa). Both the laser and the cavity mirrors are glued on piezoelectric transducers for sweeping and modulation purposes. The cavity resonance frequency is slightly modulated (frequency: $f_1=33$ kHz, depth: ~1 kHz) and is locked to the laser frequency using first harmonic phase sensitive detection at f_1 of the transmitted intensity. The intracavity optical power is set at 400 μ W. The laser frequency is modulated (frequency $f_2=5.67$ kHz, depth 100-200 kHz). A third harmonic phase-sensitive detection at $3f_2$ gives the saturated absorption signal (see Fig. 4) with a high signal to noise ratio of more than 1000 at 1s integration time. This signal is used to stabilise the CO₂ laser with a loop bandwidth of 100Hz. In the case of the 9R(42) CO₂ line, there is no coincidence with intense HCOOH lines, so the CO₂ laser frequency is up-shifted by 128 MHz using an acousto-optic modulator to reach a HCOOH line.

B. The QCL system

The QCL is a singlemode cw distributed feedback laser [17] operated in a liquid nitrogen cryostat. The threshold current is 400 mA at 77 K, the maximum current is about 1 A with a 9 V polarisation voltage. It can deliver up to 160mW optical power and is tunable from 9.166 μ m to 9.240 μ m. The temperature and current tunabilities are respectively 3 GHz/K and 150 MHz/mA. Because the QCL's emission spectrum is several MHz wide and exhibits a jitter over tens of MHz, the free running QCL is not suitable for high resolution infrared spectroscopy. The spectral

features of singlemode CO₂ lasers are much better with a linewidth in the kHz range [18]. We have shown that they can be written out on the QCL spectrum with a 300-1500 MHz tunable frequency offset using a fast phase-lock loop [19] with a bandwidth of 6 MHz. The tunability range has been extended to 2 GHz by applying a 3-5 V reverse bias to the HgCdZnTe room temperature detector that monitors the beat note between the CO₂ laser and the QCL, resulting in a better detectivity. The 6 MHz loop bandwidth implies that the CO₂ laser frequency modulation is also written out on the QCL frequency with the same depth. For HCOOH spectroscopy, the QCL beam is injected into a second Fabry-Perot cavity similar to the first one (1.5 m long, finesse 100, beam waist 4.2 mm, intra-cavity optical power 400 μ W, HCOOH pressure 0.5-2 mTorr). The cavity resonance frequency is modulated (frequency f'₁=50 kHz, depth \approx 1 kHz) and is locked to the QCL emission frequency. A third harmonic phase-sensitive detection at 3f₂ gives the HCOOH saturated absorption signal depicted in Fig. 4.



FIG. 2: $CO_2/HCOOH$ reference laser. The CO_2 laser is stabilised on a HCOOH saturated absorption line. The frequency shifter is an acousto-optic modulator only used when the CO_2 laser is operated on the 9R(42) CO_2 line to up-shift the laser frequency by 128 MHz. CD: Liquid nitrogen cooled HgCdTe detector. PZT: piezo transducer. PSD : phase sensitive detection.



FIG. 3: QCL spectrometer. The QCL and CO₂/HCOOH reference laser beams are beated on a fast room temperature HgCdZnTe detector (HD). The beatnote frequency is divided by 8 and phase compared to a RF synthesizer of frequency δ . The QCL is phase-locked to the reference laser with a tunable frequency offset 8 δ . Because the frequency modulation of the reference laser is transferred to the QCL by the fast phase-lock loop, the HCOOH saturated absorption line is phase sensitive detected at the third harmonic f₂, see Fig. 2. PZT: piezo transducer, PLL : phase-lock loop.

III. RESULTS AND DISCUSSION

We have performed absolute (A) and relative (R) frequency measurements of HCOOH saturated absorption lines. The results are given in Table I and are respectively labelled by A or R in the second column.

A. Absolute frequency mesurement

For the absolute measurements, the $CO_2/HCOOH$ reference laser (Fig. 2) has been transported to LPL laboratory where a femtosecond frequency comb infrared frequency measurement facility is implemented [20]. We have measured

the absolute frequency of 6 HCOOH lines in quasi coincidence with 9R(36) to 9R(42) CO₂ laser lines for which no accurate molecular reference were known.

The HCOOH/CO₂ reference laser frequency is measured by comparison with a very high-harmonic of the repetition rate of a femto-second (fs) laser [20]. In the frequency domain, the fs laser provides a comb of modes separated by the repetition rate [21]. A second comb is produced by a sum-frequency generation (SFG) of the fs laser comb and the CO_2 laser in a nonlinear crystal. This results in a beat between the low frequency part of the SFG comb and the high frequency part of the initial comb. The infrared frequency is thus compared to the difference between two modes of the comb. The beatnote is finally used to phase-lock the repetition rate to the CO_2 laser frequency. This scheme is independent of the comb offset and does not require any broadening of the comb. To complete the frequency measurement procedure, the repetition rate (about 1 GHz) is detected with a fast photodiode and counted against a local oscillator at 1 GHz. This local oscillator is phase-locked to a reference signal transmitted via a 43-km long optical fiber from the LNE-SYRTE laboratory, located in Paris [22]. This laboratory has developed a high stability oscillator, which is based on a combination of a cryogenic Sapphire oscillator (CSO), an H-Maser and a set of low noise microwave synthesizers [23]. Its frequency is steered by the H-Maser in the long-term, and monitored by the Cs atomic fountain for accuracy [24]. This signal shows a frequency stability slightly below $10^{-14} \tau^{-1}$ in the range 1-10 s, and 10^{-15} from 10 to 105 s. The transfer through the optical link degrades this stability by less than one order of magnitude, while the phase noise introduced by the link can be efficiently suppressed with an active correction. Finally the resolution of the whole measurement chain is better than 0.3 Hz (10^{-14} in relative value) and will definitely not be a limitation for the present measurements.

The day to day repetability of the measurements is 100 Hz for controlled experimental conditions. No significant frequency shifts are observed by varying the intracavity optical power between 400μ W and 2mW, varying the laser modulation depth by more than a factor eight and introducing significant but small error signal offsets. We have observed a positive pressure shift smaller than 2 kHz by varying the HCOOH pressure up to 4 mTorr. At the 0.5 mTorr level, formic acid pressure cannot be precisely controlled in our setup, probably because of adsorption effects on the vacuum chamber walls. We therefore give a conservative uncertainty of 1 kHz or 3.3 10^{-8} cm⁻¹ on the absolute frequency measurements.

B. HCOOH spectroscopy

We have measured the HCOOH lines that are red or blue detuned by 300-1600 MHz with respect to the closest $CO_2/HCOOH$ reference laser line using the widely tunable phase-locked QCL. The QCL frequency is swept by 1.6 kHz increments and the saturated absorption profile is recorded and fitted by the third derivative of a lorentzian shape to determine the line center and width (see Fig 4). We are easily able to detect lines with a signal to noise ratio between 3 and 1000 in a 1s integration time, the former corresponding to normalized intensities down to 1.6 10^{-22} cm⁻¹/(molec cm⁻²), (taking into account the recent corrections of the line intensities by a factor of two [9]). With optimized intracavity power and gaz pressure, the sensitivity of the spectrometer is low enough to detect the CO_2 9R(42) saturated absorption line that have a 5. 10^{-24} cm⁻¹/(molec cm⁻²) normalized intensity. Nevertheless, absolute HCOOH line intensities cannot be determined precisely from the line profile because of the large uncertainty on the actual pressure in the cavity.

The HCOOH fitted linewidths are between 160 and 220 kHz and are dominated by modulation broadening. Indeed, the transit time broadening is 26 kHz, the pressure broadening is $0.32 \text{ cm}^{-1}/\text{atm}$ [9] i.e. 12.8 kHz/mTorr and the natural width and power broadening are negligible.

Thanks to the high signal to noise ratio of the lines, the fitted line centers are determined with uncertainties better than 1 kHz. Nevertheless, taking into account the independant uncertainties on the reference line frequency (1 kHz), the line position (1 kHz) and the HCOOH pressure determination (1 kHz), we give a conservative uncertainty of 2 kHz or $6.6 \ 10^{-8} \text{ cm}^{-1}$ on the HCCOH line frequency determined with the QCL spectrometer.

C. Discussion

In Table I, the measured frequencies are compared to the HITRAN database predicted values [8], that are given with a 10^{-6} cm⁻¹ resolution. Our measurements agree with the HITRAN predictions at the 10^{-4} cm⁻¹ level for most of the lines excepted one for which a 8. 10^{-4} cm⁻¹ (24 MHz) offset is observed. The agreement with recently predicted lines by A. Perrin [16] is better at the 2. 10^{-5} cm⁻¹ level excepted for two lines. In the last column of the Table, the * labels predicted doublets.

A few doublets predicted by A. Perrin [16] at 1089.040400, 1089.998920, 1090.055270 and 1091.066960 cm⁻¹ have not been observed in a frequency interval of ± 8 MHz (2.6 10^{-4} cm⁻¹) around the predicted values, with a sensitivity



FIG. 4: Third harmonic phase sensitive saturated absorption signal of HCOOH line n° 3 of Table I versus the frequency offset 8 δ between the reference laser and the phase locked QCL (see Fig. 3). The integration time is 1 s per point. Full line: adjustement by a third lorentzian derivative. The full width at half maximum is 220 kHz.

of $10^{-22} \text{ cm}^{-1}/(\text{molec cm}^{-2})$.

In Table II, we compare the present frequency measurements with previously published ones. There is a good agreement with the measurements obtained by CO_2 laser saturation spectroscopy and reported in [11], the accuracy being improved by at least one order of magnitude.

D. Reference frequency stability

The aim of this work is to realise a reliable and stable frequency reference to probe two-photon vibrational transitions in H_2^+ with a resolution of the order of 1 kHz. We have measured the relative frequency stability of two independant systems. The first one is the CO₂/HCOOH reference laser stabilized on the HCOOH line n°4 of Table I. The second system involves the QCL and a master CO₂ laser similar to the first one. The QCL is phase locked to the CO₂ laser with a fixed frequency offset. The QCL beam probes HCOOH line n°7 of Table I. The saturated absorption signal is used to lock the master CO₂ laser frequency. To characterize the relative stability of the two laser systems, the beatnote frequency is counted with 1 s gate time. The Allan deviation shown in Fig. 5 exhibits a white frequency noise behaviour of 146 Hz $\tau^{-1/2}$, that is 4.5 10⁻¹² in relative value. The relative stability reaches the 10 Hz level for 200 s integration time. Both lasers are locked using saturated absorption signal having similar signal to noise ratios so individual stabilities are expected to be comparable and of the order of 100 Hz $\tau^{-1/2}$.

IV. CONCLUSION

We have set up a CO_2/HCOOH reference laser that can be operated from the 9R(36) to 9R(42) line and measured the absolute frequency of 6 saturated absorption lines with an accuracy of 1 kHz (3.3 10^{-8}cm^{-1}) improved by two orders of magnitude as compared to previously reported results or predictions. Using a quantum cascade laser phase (QCL) locked to a master CO₂ laser, we have determined the frequency of 17 additional formic acid lines with an accuracy of 2 kHz. The frequency stability of both the reference laser and the phase locked QCL are characterized by a relative Allan deviation of 3.1 $10^{-12} \tau^{-1/2}$. This performance demonstrates that the CO₂/HCOOH phase locked QCL is a suitable tool for H₂⁺ high resolution spectroscopy at the kHz level.

Acknowledgments

We thank O. Acef, G. Santarelli and M. Lours (LNE-SYRTE). We also thank A. Perrin (LISA, Université de Créteil) for fruitfull discussions about HCOOH. Laboratoire Kastler Brossel is UMR 8552 du CNRS. This work was

$\rm CO_2$		f	δf	σ_m	δ_H	δ_P								
line		(kHz)	$\left(\mathrm{kHz}\right)$	$({\rm cm}^{-1})$	$({\rm cm}^{-1})$	$({\rm cm}^{-1})$	J'	\mathbf{K}_{a}	K_c	' J"	\mathbf{K}_{a}	"K _c "		
9R(36)	А	32 615 874 282.1	1.0	1087.948 459 40		-0.000601	V6 26	21	5	27	21	6	*	
							V6 26	21	6	27	21	7		
9R(38)	R	32 645 920 103.8	2.0	1088.950 680 13	-0.000030	0.000040	V6 17	3	15	17	4	14		
	R	32 646 482 873.7	2.0	$1088.969 \ 452 \ 12$	0.000022	0.000072	V6 19	3	16	20	3	17		
	R	32 646 774 027.4	2.0	1088.979 163 96	-0.000216	-0.000096	V6 21	11	10	22	11	11	*	
							V6 21	11	11	22	11	12		
	A	32 647 442 530.0	1.0	1089.001 462 80	-0.000247	-0.000117	V6 30	1	29	30	2	28		
	Α	32 647 451 806.2	1.0	1089.001 772 23		-0.000408	V6 52	6	46	53	5	49		
	R	32 648 078 880.7	2.0	1089.022 689 18	-0.000031	-0.000051	V6 16	3	14	16	4	13		
	R	32 648 797 827.9	2.0	1089.046 670 68	0.000101	-0.000019	V6 20	0	20	21	0	21	L	
9R(40)	R	$32 \ 677 \ 093 \ 109.1$	2.0	1089.990 499 66	0.00007	-0.000010	$\rm V6~29$	1	28	29	2	27		
	R	$32\ 677\ 848\ 146.1$	2.0	$1090.015\ 684\ 99$	0.000105	-0.000255	$\rm V6~19$	1	19	20	0	20		
	Α	$32\ 678\ 207\ 409.6$	1.0	$1090.027\ 668\ 73$	-0.000021	-0.000021	$\rm V6~18$	3	16	19	3	17		
	Α	$32\ 678\ 247\ 320.7$	1.0	$1090.029\ 000\ 02$		-0.000070	$\rm V6~22$	0	22	22	2	21		
	R	$32\ 679\ 212\ 591.0$	2.0	$1090.061\ 197\ 97$	0.000018	0.000088	V6~4	2	3	5	3	2		
	R	$32\ 679\ 681\ 163.9$	2.0	$1090.076 \ 827 \ 88$	0.000018	-0.000012	V6~4	2	2	5	3	3		
0R(42)	в	32 706 583 453 5	2.0	1000 074 101 67	0 000338	0.000082	$\rm V6~18$	9	9	19	9	10	4	1
510(42)	10	52 100 505 455.5	2.0	1030.374 131 07	-0.000338		$\rm V6~18$	9	10	19	9	11	Ť	T
	R	$32\ 706\ 672\ 889.5$	2.0	$1090.977\ 174\ 93$	0.000025	-0.000035	$\rm V6~17$	2	16	18	2	17		2
	R	32 707 063 566 1	2.0	1000 000 206 50	0 000797	0.000077	$\rm V6~20$	15	5	21	15	6	*	3
	10	02 101 000 000.1	2.0	1000.000 200 00	0.000101		$\rm V6~20$	15	6	21	15	$\overline{7}$		Ű
	Α	$32\ 708\ 391\ 980.5$	1.0	$1091.034\ 517\ 64$	-0.000038	-0.000032	$\rm V6~21$	2	20	21	3	19		4
	R	$32\ 709\ 015\ 560.0$	2.0	$1091.055\ 318\ 01$	-0.000042	-0.000112	V6 8	4	5	7	5	2		5
	R	$32\ 709\ 017\ 250.3$	2.0	$1091.055\ 374\ 39$	-0.000036	-0.000116	V6 8	4	4	7	5	3		6
	R	$32\ 709\ 175\ 798.5$	2.0	$1091.060\ 662\ 99$	-0.000147	-0.000023	$V6\ 17$	5	12	18	5	13		7
	R	$32\ 709\ 189\ 841.6$	2.0	$1091.061\ 131\ 42$	-0.000149	0.000081	V6 17	5	13	18	5	14		8
	в	32 700 274 024 5	2.0	1001 063 060 48		0.000001	$\rm V6~22$	20	3	23	20	4		0
	n	52 109 214 924.0	2.0	1031.005 303 46		-0.000091	$\rm V6~22$	20	2	23	20	3		9

TABLE I: HCOOH line frequencies. The first column gives the closest CO₂ laser line and the second one the measurement system (A for absolute frequency measurement and R for relative measurement by QCL spectroscopy). Columns 3 and 4 give the measured frequencies f (this work) and their uncertainties δf . Column 5 is the corresponding wavenumber σ_m . Columns 6 and 7 give the differences $\delta_H = \sigma_m - \sigma_H$ and $\delta_P = \sigma_m - \sigma_P$ between our measurements and the wavenumbers σ_H and σ_P predicted respectively by the HITRAN database [8] and A. Perrin [16]. The following columns give the usual line nomenclature. (*) denotes predicted doublets. Figures in last column refers to Fig. 1.

	(a)	(b)	(a)-(b)							
	kHz	MHz	kHz	J	, K	K_c	' J"	\mathbf{K}_{a}	"K _c "	ref.
9R(38)	32 646 774 027 4 (2 0)	32 646 775 1	-1072.6	V6 2	1 11	10	22	11	11	[14]
510(00)	02 010 111 021.1 (2.0)	02 010 110.1		$V6\ 2$	1 11	11	22	11	12	
	$32 \ 647 \ 442 \ 530.0 \ (1.0)$	$32 \ 647 \ 442.52(0.1)$	10.0	V6 3	$0 \ 1$	29	30	2	28	[11] *
	$32 \ 647 \ 451 \ 806.2 \ (1.0)$	$32\ 647\ 451.80(0.1)$	6.2	V6~5	$2 \ 6$	46	53	5	49	[11] *
9R(40)	$32\ 678\ 207\ 409.6\ (1.0)$	32 678 207.36	49.6	V6 1	8 3	16	19	3	17	[11]

TABLE II: Comparison of the present frequency measurements (a) and already published measurements (b). The line assignements are those given in HITRAN or the recent reference [16]. *: the line assignement in ref. [11] is different.



FIG. 5: Square root of the Allan deviation of the two HCOOH stabilized laser beatnote as a function of integration time τ , obtained from a record of 633 1s gate time measurements. The dashed curve is a fit giving a 146 Hz $\tau^{-1/2}$ behaviour. Dead time between measurements is 0.25 s.

supported by an ACI jeune 03-2-379, BNM grants 02-3-008 and 04-03-009 and Evry University.

- [1] Codata 2002 : P. J. Mohr, B. N. Taylor, Review of Modern Physics 77, 1 (2005).
- [2] T. Beier, H. Häffner, N. Hermanspahn, S. G. Karshenboim, H.-J. Kluge, W. Quint, S. Stahl, J. Verdú, G. Werth, Phys. Rev. Lett. 88, 011603 (2002).
- [3] D. L. Farnham, R. S. Van Dyck, P. B. Schwinberg, Phys. Rev. Lett. 75, 3598 (1995).
- [4] L. Hilico, N. Billy, B. Grémaud, D. Delande, J. Phys. B 34, 491 (2001).
- [5] J-Ph. Karr, S. Kilic, L. Hilico, J. Phys. B 38, 853 (2005).
- [6] B. Roth, J. C. J. Koelemeij, H. Daerr, S. Schiller, Phys. Rev. A 74, 040501(R) (2006).
- [7] A. Amy-Klein, H. Vigué, C. Chardonnet, J. Mol. Spec. 228, 206-212 (2004).
- [8] HITRAN database : L.S. Rothman, D. Jacquemart, A. Barbe, D. Chris Benner, M. Birk, L.R. Brown, M.R. Carleer, C. Chackerian Jr., K. Chance, L.H. Coudert, V. Dana, V.M. Devi, J.-M. Flaud, R.R. Gamache, A. Goldman, J.-M. Hartmann, K.W. Jucks, A.G. Maki, J.-Y. Mandin, S.T. Massie, J. Orphal, A. Perrin, C.P. Rinsland, M.A.H. Smith, J. Tennyson, R.N. Tolchenov, R.A. Toth, J. Wander Auwera, P. Varanasi, G. Wagner, J. Quant. Spec. Rad. Tr 96, 139 (2005).
- [9] J. Vander Auwera, K. Didriche, A. Perrin, F. Keller, J. Chem. Phys. 126, 1 (2007).
- [10] E. Willemot, D. Dangoisse, J. Bellet, J. Mol. Spec 77, 161-168 (1979).
- [11] B.M. Landsberg, D. Crocker, R. B. Butcher, J. Mol. Spec 92, 67-76 (1982).
- [12] H.-T. Ma, R. J. Butcher, J. Mol. Spec. 107, 284-291 (1984).
- [13] R. E. Bumgarner, J.-In Choe, S. G. Kukolich, J. Mol. Spec 132261-276 (1988).
- [14] T.L. Tan, K.L. Goh, P.P. Ong, H.H. Teo, J. Mol. Spec. 202, 194-206 (2000).
- [15] O.I. Baskakov, J. Demaison, Journal of Molecular spectroscopy 211, 262 (2002).
- [16] A. Perrin, J. Vander Auwera, Journal of Quantitative Spectroscopy and Radiative Transfert JQRST 2007, doi:10.1016/j.jqrst.2007.05.002.
- [17] model SBCW496DN from AlpesLasers.
- [18] B. Frech, L. F. Constantin, A. Amy-Klein, O. Phavorin, C. Daussy, Ch. Chardonnet, M. Mürtz, Appl. Phys. B 67, 217 (1998).
- [19] F. Bielsa, A. Douillet, T. Valenzuela, J.-Ph. Karr, L. Hilico, Optics Letters 32, 1641 (2007).
- [20] A. Amy-Klein, A. Goncharov, M. Guinet, C. Daussy, O. Lopez, A. Shelkovnikov, C. Chardonnet, Optics Letters 30, 3320 (2005).
- [21] See for instance the review paper : S. T. Cundiff, J. Ye and J. L. Hall, Review of Scientific Instruments 72, 3749 (2001).
- [22] C. Daussy, O. Lopez, A. Amy-Klein, A. Goncharov, M. Guinet, C. Chardonnet, F. Narbonneau, M. Lours, D. Chambon, S. Bize, A. Clairon, G. Santarelli, M. E. Tobar and A. N. Luiten, Physical Review Letters 94, 203904 (2005).
- [23] D. Chambon, S. Bize, M. Lours, F. Narbonneau, H. Marion, A. Clairon, G. Santarelli, A. Luiten and M. Tobar, Review of Scientific Instruments 76, 094704 (2005).
- [24] C. Vian, P. Rosenbusch, H. Marion, S. Bize, L. Cacciapuoti, S. Zhang, M. Abgrall, D. Chambon, I. Maksimovic, P. Laurent, G. Santarelli, A. Clairon, A. Luiten, M. Tobar, and C. Salomon, IEEE Trans. Instrum. Meas. 54, (2005).