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## Elaboration and Characterisation of two new Iron Titanium Phosphates Fe<sub>0.5</sub>TiO(PO<sub>4</sub>) and Fe<sub>0.5</sub>Ti<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>

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#### Abstract

Two new phosphates  $Fe_{0.5}TiO(PO_4)$  and  $Fe_{0.5}Ti_2(PO_4)_3$  have been prepared and characterised by X-ray diffraction, Mössbauer spectroscopy and magnetic measurements.  $Fe_{0.5}TiO(PO_4)$  is isostructural to the oxyphosphate  $Ni_{0.5}TiO(PO_4)$  while  $Fe_{0.5}Ti_2(PO_4)_3$  belongs to the Nasicon family. Their structures are formed by a 3 D framework of  $TiO_6$  octahedra and  $PO_4$  tetrahedra linked by corners.  $Fe^{2+}$  ions occupy isolated octahedral cavities situated between two  $TiO_6$  octahedra. The magnetic and Mössbauer parameters are typical of  $Fe^{2+}$  ions in octahedral sites with high spin configuration  $t_{2g}^4 e_g^2$ .

#### INTRODUCTION

The study of phosphates with octahedral  $[TiO_6]$  and tetrahedral  $[PO_4]$  groups has been intensive since the discovery of the Nasicon<sup>1</sup> (Na super ionic conductor) and the KTP<sup>2</sup> (KTiOPO<sub>4</sub>) materials. In the course of our investigation on these phosphate families we discovered two new phases: Fe<sub>0.5</sub>TiO(PO<sub>4</sub>) and Fe<sub>0.5</sub>Ti<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>. The present paper describes their preparation and characterisation by X-ray diffraction, Mössbauer spectroscopy and magnetic measurements.

#### EXPERIMENTAL

The powders of  $Fe_{0.5}TiO(PO_4)$  and  $Fe_{0.5}Ti_2(PO_4)_3$  were prepared by conventional solid state reaction techniques in silica tubes sealed under vacuum. Stoichiometric proportions of  $Fe_2O_3$ ,  $TiO_2$ ,  $TiP_2O_7$  and Fe have been mixed, ground and heated progressively up to 800°C for  $Fe_{0.5}TiO(PO_4)$  and 950°C for  $Fe_{0.5}Ti_2(PO_4)_3$ . These phosphates can also be obtained by exchange reaction.

The X-ray diffraction data were collected at room temperature with Philips PW 3040 ( $\theta$ - $\theta$ ) diffractometer using a graphite monochromator. The magnetic susceptibility determination was performed from 4.2 to 300 K in an applied magnetic field (0 < H < 3T), using a SQUID magnetometer (MPMS-S2, Quantum Design). Mössbauer measurements were performed using a constant acceleration HALDER-type spectrometer, with a <sup>57</sup>Co(Rh) source in transmission geometry. Isomer shift values refer to  $\alpha$ -Fe at 293 K. The spectra at 4.2 K and 293 K were recorded using a variable temperature cryostat.

#### **RESULTS AND DISCUSSION**

The structure of Fe<sub>0.5</sub>TiO(PO<sub>4</sub>) and Fe<sub>0.5</sub>Ti<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub> have been determined from powder X-ray diffraction data, by Rietveld refinement using Fullprof program<sup>3</sup>. Here we give a summary of this study, details of which will be published elsewhere. Fe<sub>0.5</sub>TiO(PO<sub>4</sub>) is isostructural to Ni<sub>0.5</sub>TiO(PO<sub>4</sub>)<sup>4</sup>. It crystallizes in the monoclinic system, space group P2<sub>1</sub>/c (No.14), with a = 7.4039(3) Å, b = 7.3838(3) Å, c = 7.4083(3) Å,  $\beta$  = 120.366(1), Z= 4. Fe<sub>0.5</sub>Ti<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub> is isostructural to Co<sub>0.5</sub>Ti<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub><sup>5</sup>. At room temperature its XRD pattern can be indexed assuming a hexagonal cell with the following parameters: a<sub>h</sub> = 8.509 ± 0.003Å and c<sub>h</sub> = 21.015 ± 0.004Å.

The structures of these phosphates (Figures 1, 2) are formed by a tridimensional framework of  $TiO_6$  octahedra and PO<sub>4</sub> tetrahedra linked by corners. The TiO<sub>6</sub> octahedra are rather regular and isolated from each other in Fe<sub>0.5</sub>Ti<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>, while in Fe<sub>0.5</sub>TiO(PO<sub>4</sub>) they are linked together by corners and form -Ti-O-Ti-O- infinite chains with alternating short and long Ti-O bonds. In both phases, Fe<sup>2+</sup> ions occupy isolated octahedral sites located between two TiO<sub>6</sub> octahedra and sharing with them two faces.



Figure 1: Structure of Fe<sub>0.5</sub>TiO(PO<sub>4</sub>)

Figure 2: Structure of Fe<sub>0.5</sub>Ti<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>

The thermal variation of the reciprocal magnetic susceptibility (Figure 3) between 4 and 300 K is linear. This paramagnetic behaviour shows that there is no magnetic interactions between divalent iron ions as expected from the structural study. Indeed the FeO<sub>6</sub> octahedra are isolated from each other and the Fe<sup>2+</sup>-Fe<sup>2+</sup> distances are longer than 5Å. The effective magnetic moment  $\mu_{eff} = 5.13 \ \mu\text{B}$  confirms the oxidation state (+II) of iron in octahedral sites with high spin configuration (Fe<sup>2+</sup>: t<sub>2g</sub><sup>4</sup>e<sub>g</sub><sup>2</sup>).

Figure 4 shows the Mössbauer spectra of  $Fe_{0.5}TiO(PO_4)$  and  $Fe_{0.5}Ti_2(PO_4)_3$  measured at 4.2 K and 293 K. The values of isomer shift ( $\delta$ ), quadruple splitting ( $\Delta$ ) and width at half maximum ( $\Gamma$ ) (Table I) are in agreement with the structural and magnetic results.



Figure 3: Thermal variation of the reciprocal susceptibility of  $Fe_{0.5}TiO(PO_4)$  (a) and  $Fe_{0.5}Ti_2(PO_4)_3$  (b)



#### Fe<sub>0.5</sub>TiO(PO<sub>4</sub>)

Fe0.5Ti2(PO4)3

Figure 4: <sup>57</sup>Fe Mössbauer spectra of  $Fe_{0.5}TiO(PO_4)$  and  $Fe_{0.5}Ti_2(PO_4)_3$  measured at 4.2 K (a) and 293 K (b)

Compound	T (K)	δ±0.01 (mm.s <sup>-1</sup> )	$\Delta \pm 0.01$ (mm.s <sup>-1</sup> )	$\Gamma_2 \pm 0.01$ (mm.s <sup>-1</sup> )	$\Gamma_1 \pm 0.01$ (mm.s <sup>-1</sup> )
$Fe_{0.5}TiO(PO_4)$	293	1.11	3.85	0.28	0.38
$Fe_{0.5}TiO(PO_4)$	4.2	1.24	4.02	0.30	0.30
$Fe_{0.5}Ti_2(PO_4)_3$	293	1.26	3.32	0.38	0.32
$Fe_{0.5}Ti_2(PO_4)_3$	4.2	1.36	4.21	0.30	0.30

Table 1: Mössbauer parameters of Fe<sub>0.5</sub>TiO(PO<sub>4</sub>) and Fe<sub>0.5</sub>Ti<sub>2</sub> (PO<sub>4</sub>)<sub>3</sub>

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