



**HAL**  
open science

## Ab initio of magnetic properties of cobalt and tetracobalt nitride Co<sub>4</sub>N

Samir F. Matar, A. Houari, A. Belkhir

► **To cite this version:**

Samir F. Matar, A. Houari, A. Belkhir. Ab initio of magnetic properties of cobalt and tetracobalt nitride Co<sub>4</sub>N. *Physical Review B: Condensed Matter and Materials Physics (1998-2015)*, 2007, 75 (24), 245109 (7 p.). 10.1103/PhysRevB.75.245109 . hal-00154321

**HAL Id: hal-00154321**

**<https://hal.science/hal-00154321v1>**

Submitted on 13 Jun 2007

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

# Ab initio studies of magnetic properties of cobalt and tetracobalt nitride $\text{Co}_4\text{N}$

S. F. Matar<sup>a)\*</sup>, A. Houari<sup>b)</sup> and M. A. Belkhir<sup>b)</sup>.

<sup>a)</sup>*ICMCB, CNRS, Université Bordeaux 1. 87 avenue*

*Dr A Schweitzer, 33600 Pessac Cedex, France.*

<sup>b)</sup>*Laboratoire de Physique Théorique, Département de Physique,*

*Université de Bejaia. 06000 Bejaia, Algeria.*

(Dated: May 21, 2007)

## Abstract

Magnetic properties and bonding analyses of perovskite structure  $\text{Co}_4\text{N}$  nitride have been investigated within density functional theory using both pseudo potential and all electron methods. In the same time, the structural and magnetic stability of pure cobalt in hexagonal close packed (HCP), face centered cubic (FCC) and body centered cubic (BCC) structures are reviewed. At equilibrium, non-spin polarized (NSP) and spin polarized (SP) calculations of the energy versus volume show that the ground state is ferromagnetic in both materials. HCP-Co is found to be more stable than the cubic ones. Magnetic moments of Co atoms in  $\text{Co}_4\text{N}$  nitride respectively belonging to two different crystallographic sites are studied over a wide range of the cubic lattice constant, and a comparison with the FCC-cobalt one is given. The volume expansion in the nitride indicates that the corner  $\text{Co}^I$  atoms show localized magnetism while face center  $\text{Co}^{II}$  atoms exhibit an itinerant behavior. Like in FCC-Fe/ $\text{Fe}_4\text{N}$ , a “low volume-low moment” and “large volume-high moment” behavior is observed for FCC-Co/ $\text{Co}_4\text{N}$ . The density of states of the  $\text{Co}_4\text{N}$  ferromagnetic ground state is interpreted within the rigid band model. The different bonding characters of  $\text{Co}^I$ -N versus  $\text{Co}^{II}$ -N are shown with help of electron localization function ELF plots and spin resolved chemical bonding criteria.

PACS numbers: 71.15.Mb, 71.15.Nc, 71.20-b, 75.10.Lp, 74.25.Ha, 73.43.Cd

---

\* Corresponding author (s.matar@u-bordeaux1.fr)

## I. INTRODUCTION

With respect to their thermal, mechanical and magnetic properties, the transition metal nitrides are known to be very interesting in many technological areas. For example, in the field of high density magnetic recording, iron nitrides are applied [1-5]. Unlike iron nitrides, the cobalt nitrides seem to be less explored and there are only few studies which treat their magnetic properties. The cobalt-nitrogen system was reported to contain many phases ranging from  $\text{Co}_4\text{N}$  to  $\text{CoN}$  [6-11]. In the cobalt-rich side, a cubic  $\text{Co}_4\text{N}$  was synthesised with a lattice constant of 6.7788 bohr (1 bohr=0.529 Å), but no magnetic data were given [7]. On the other side bulk  $\text{Co}_2\text{N}$  was prepared with an orthorhombic  $\text{Fe}_2\text{C}$ -type structure [10], whereas no structure was indexed in  $\text{Co}_2\text{N}$  thin films [11]. However, the case of  $\text{CoN}$  mononitride seems somewhat conflicting. Like what happens in the iron mononitride [12], it was early reported in cubic rocksalt structure [9], but recently it was found in a paramagnetic zinc-blende type one [13]. On the theoretical side a lack of data about the cobalt nitrides is noted. Except some density functional calculations on  $\text{CoN}$  [14, 15], no calculations are carried out for  $\text{Co}_2\text{N}$  and  $\text{Co}_4\text{N}$ . Nevertheless, for the latter some studies on antiperovskite cobalt-substituted  $\text{Fe}_4\text{N}$ , i.e.  $\text{CoFe}_3\text{N}$ , have been undertaken [16, 17].

From the structural point of view,  $\text{Co}_4\text{N}$  can be assimilated as an insertion of N atom in FCC-cobalt. In this context, a reinvestigation of cobalt metal, at the same time as  $\text{Co}_4\text{N}$  could be useful for a better understanding of the nitride magnetic properties; especially the Co-N interaction.

Many experimental and theoretical reports have been carried out in order to study the three known structural types of cobalt, namely body centered cubic (BCC), face centered cubic (FCC) and hexagonal close packed (HCP), see [18-26] and references therein. Under ambient conditions of temperature and pressure, Co is well known to crystallize in hexagonal close packed (HCP) structure. Experimentally a lattice constant in the vicinity of  $a=4.739$  bohr, a  $c/a$  parameters ratio between 1.615-1.623 and an atomic magnetic moment of  $\sim 1.6\mu_B$  were found [18-21]. These results were confirmed later by density functional based calculations [22-23]. Further the stabilization of cobalt in BCC-structure [20], a structural transformation between the two close packed structures: FCC and HCP are also known to occur in cobalt [21]. Recently, the stability of the two latter structures was investigated theoretically but BCC-Co was not included [25]. To our knowledge, the only work which

includes the three types of structures is the one of Min *et al* [23] but within the framework of local spin density approximation (LSDA) to the density functional theory [28]. In this context, it is important to remind the well known drawback of this energy functional in predicting the ferromagnetic BCC-ground state of iron [29] which was of a major impetus in the development of generalized gradient approximation (GGA) [30].

The lack of theoretical as well as experimental data concerning  $\text{Co}_4\text{N}$  (and in general on the Co-N system), has motivated us to carry out a first principles study. The aim of this present work is to assess the chemical role of nitrogen within  $\text{Co}_4\text{N}$  and to check whether the variation for the magnetic moments at different lattice sites follows the trends of cobalt metal in the same FCC-like structure through magnetovolume effects.

The present paper is organized as follows. In section II the computational framework of calculations and the methods are briefly presented. It is followed by a section, divided in two parts. The first part describes and discusses the phase stability and magnetic moments in cobalt; the second one is devoted to the  $\text{Co}_4\text{N}$  electronic structure and magnetic properties, where these are treated in details by means of partial density of states and chemical bonding assessments. Finally, section IV presents the conclusions.

## II. COMPUTATIONAL FRAMEWORK AND METHODS

Two computational methods were used in the framework of density functional theory.

For a description of the electronic and magnetic structures, the all electrons augmented spherical wave (ASW) method [31] was used. The exchange-correlation effects were accounted for within the generalized gradient approximation (GGA) using the parameterization of Perdew, Burke and Ernzerhof [32]. This scheme was preferred over the LDA (local density approximation) in view of the recent results on iron mono nitrides [12]. In the ASW method, the wave function is expanded in atom-centered augmented spherical waves, which are Hankel functions and numerical solutions of Schrödinger's equation, respectively, outside and inside the so-called augmentation spheres. In order to optimize the basis set, additional augmented spherical waves were placed at interstitial sites corresponding to middle of cube edges of the open (in the sense of non close packed) perovskite structure of  $\text{Co}_4\text{N}$ . The cobalt 4s, 3d and nitrogen 2s, 2p states were treated as valence states. The basis set was complemented by wave functions of s-, p-, and possibly, d-symmetry centered at the interstitial

sites which ensure for the covalence effects within the lattice. A sufficiently large number of  $k$  points was used to sample the irreducible wedge of the Brillouin zone, and the energy and charge differences between successive iterations were converged below  $\Delta E=10^{-8}$  Ryd. and  $\Delta Q=10^{-8}$ , respectively, in order to obtain accurate values of the magnetic moments and total energy differences between various ordered magnetic states.

To extract more information about the nature of the interactions between the atomic constituents from electronic structure calculations, the crystal orbital overlap population (COOP) [33] or the crystal orbital Hamiltonian population (COHP) [34] may be employed. Both approaches provide a qualitative description of the bonding, nonbonding and antibonding interactions between two atoms. A slight refinement of the COHP was recently proposed in form of the 'energy of covalent bond' (ECOV), which combines COHP and COOP to calculate quantities independent of the choice of the zero of potential [35]. Both COOP and ECOV give similar general trends, but COOP, when defined within plane-wave basis sets, exaggerates the magnitude of antibonding states. In the present work the ECOV description was used for the chemical bonding analysis. In the plots, negative, positive and zero ECOV magnitudes are relevant to bonding, antibonding and nonbonding interactions respectively.

In as far as no structural determination is available for  $\text{Co}_4\text{N}$  trends in cell volumes and ground state crystal structure were desirable in the first place. The equilibrium structures were obtained from a full geometry relaxation of nitride systems starting from  $\text{Fe}_4\text{N}$  structural setup and using a pseudo potential approach within the VASP code [36]. Ultra soft Vanderbilt pseudo potentials (US-PP) [37] built within GGA scheme [32] were used. The calculations were converged at an energy cut-off of 435.12 eV for the plane-wave basis set with respect to the  $k$ -point integration with a starting mesh of  $6*6*6$  up to  $12*12*12$  for best convergence and relaxation to zero strains. The tetrahedron method with Blöchl corrections [38] was applied leading from 141 to 1569 inequivalent tetrahedra. The Brillouin-zone integrals were approximated using a special  $k$ -point sampling.

### III. RESULTS AND DISCUSSIONS.

#### A. Cobalt metal.

The volume dependence of the total energy for spin polarized (SP) ferromagnetic and non-spin polarized (NSP -meaning a non magnetic spin degenerate system) configuration for the three cobalt phases: BCC, FCC and HCP are displayed in figure 1-a, 1-b and 1-c, respectively. The ferromagnetic state is largely preferred over the non-magnetic one in all 3 different structures. The lowest energy in the spin polarized configuration is obtained for HCP-Co which is the ambient form of cobalt. This structural stability in the spin-polarized case is better illustrated in figure 1-d, where the  $E(V)$  plots for the three structures are shown together in the SP ferromagnetic state. Between HCP and FCC cobalt, a small energy difference  $\Delta E(\text{SP-NSP}) \sim 0.2 \text{ eV/atom}$  is obtained; this agrees well with the low magnitudes reported experimentally and points to the possible coexistence of the two phases at some high pressure range [21]. Furthermore, considering the non magnetic state where FCC cobalt is more stable than HCP-Co, one should conclude that the spin polarization is in favour of the latter. Our calculations at low volume show that the HCP-phase remains (ferro)magnetic whereas FCC one begins to lose the magnetic character (becomes non magnetic) while being more stable than the former. This finding is in agreement with the recent experimental observations of a non magnetic FCC-phase preferred to the HCP one in cobalt under high pressure [21]. From figure 1-d, the theoretical equilibrium volumes are nearly equal for HCP and FCC phases ( $\sim 78.27 \text{ bohr}^3 \cdot \text{atom}^{-1}$ ), whereas the BCC one is slightly larger ( $\sim 79.51 \text{ bohr}^3 \cdot \text{atom}^{-1}$ ). In terms of total energy, and as confirmed by recent theoretical study [26], a BCC-cobalt phase as bulk material should not be stable at ambient pressure. Nevertheless, the experimentally stabilized BCC cobalt thin films [20] were explained to originate from such imperfections or substrate lattice constant mismatch at an early stage of growth. Furthermore a critical thickness above which the BCC phase begins to transform to HCP one was identified [27]. In HCP-cobalt, our  $c/a$  ratio (1.603) and magnetic moment ( $1.634 \mu_B$ ) are in good agreement with experiments [18-21] and in better agreement than recent full potential calculations [25]. For FCC cobalt no experimental report is available for the cobalt magnetic moment; on the other hand our computed value of  $\sim 1.67 \mu_B$  is in good agreement of theoretical results both in LSDA [23] and GGA [25] exchange-correlation functionals. As

for BCC-Co, it needs to be stressed that our  $1.757 \mu_B$  moment magnitude which agrees with computed LSDA [23] one, is larger than the only reported experimental value of  $1.53 \mu_B$  [20].

## B. Tetracobalt nitride $\text{Co}_4\text{N}$ .

### 1. Geometry optimization and electron localization from pseudo potential calculations.

We started the geometry optimization by using  $\text{Fe}_4\text{N}$  structural setup with a lattice parameter from experiment:  $a=7.165$  bohr. We recall that this structure can be viewed from a cubic perovskite lattice:  $\text{ABX}_3$  where A is a large cation at corner, B is a smaller cation at cube center and X is an anion at face centers. From this  $\text{Co}^I\text{Co}_3^{\text{II}}\text{N}$  is an anti-perovskite because  $\text{Co}^{\text{II}}$  replace X anion sites at face centers. Then the lattice geometry was fully relaxed for the cell volume, shape and atomic positions. The calculations were carried out for both non magnetic and spin polarized configurations. The resulting structure was found to be cubic with lattice constants:  $a_{NSP}(\text{Co}_4\text{N})=6.975$  bohr,  $a_{SP}(\text{Co}_4\text{N})=7.062$  bohr, with an energy difference of  $\Delta E(\text{SP-NSP})= -0.982$  eV favoring the spin polarized configuration. The experimental value of  $a_{exp.}(\text{Co}_4\text{N})= 6.779$  bohr [7] is significantly smaller and this calls for an interpretation.

One explanation can be found in the possibility of existence of N vacancies within the lattice. A geometry optimization with a nitrogen deficient system was carried out calling for a new cell set up. Starting from a supercell with 8 formula units leading to 32 Co and 8 N, the above calculated lattice constant of  $\text{Co}_4\text{N}$  was checked firstly. Then vacancies were created at nitrogen sites. With 2 N vacancies, a small change of lattice constant is obtained with  $a= 6.925$  bohr; with 4 nitrogen vacancies the calculated lattice constant becomes smaller with  $a = 6.762$  bohr. This magnitude comes close to the experimental value. When all 8 nitrogen sites are vacant, a lattice constant of  $a= 6.539$  bohr is reached. These results support the potentiality of a partly nitrogen deficient lattice in the sense of the existence of some nitrogen lacking domains with stoichiometries such as  $\text{Co}_4\text{N}_{1-x}$ , beside fully stoichiometric  $\text{Co}_4\text{N}$ . While such a result is not reported experimentally [7], we suggest that nitriding conditions calling for high nitrogen pressures can be useful to obtain the stoichiometric compound [39]. Although the supercell calculation results allowed to clarify

the hypothesis of nitrogen vacancies with the tetracobalt nitride lattice, it should be kept in mind that they assumed an ideal bulk lattice with an ordered distribution of nitrogen atoms and vacancies, while in experimental reality those should be randomly distributed with possible surface effects.

A quantity that can be extracted from calculations is the electron localization function ELF which allows determining the amount of localization of electrons with respect to the free electron gas distribution as described by Becke and Edgecomb [40]. Figure 2 shows a  $\langle 101 \rangle$  diagonal plane crossing all atomic sites of  $\text{Co}_4\text{N}$ . This illustration stresses the isolated character of the corner sites  $\text{Co}^I$  at  $a\sqrt{3}/2$  from N with respect to face center cobalt  $\text{Co}^{II}$  which are at  $a/2$  from N. Further there is a non negligible localization along  $\text{Co}^{II}$ -N- $\text{Co}^{II}$  direction. Nitrogen exhibits the largest localization (large white area around it) in agreement with the description of the system as a “nitride” from the chemical view point. On the contrary, the metal sites show no localization (dark areas) due to their electropositive character so that electrons are smeared out in between the lattice sites (light grey areas) like in a metal, which is the characteristics of the system under study.

### C. Analysis of the electronic and magnetic structure

In this section, we investigate in detail the magnetic and electronic properties of pure  $\text{Co}_4\text{N}$ . The chemical effects of nitrogen are studied by means of partial density of states (PDOS) and chemical bonding properties (covalence energy ECOV). Within FCC cobalt the result of nitrogen insertion is to redistribute the FCC lattice into two different sublattices:  $\text{Co}^I$  (cube corners) and  $\text{Co}^{II}$  (face centers). The same atomic arrangement is found within  $\text{Fe}_4\text{N}$  but intriguingly the lattice parameter of the pure  $\text{Co}_4\text{N}$  found experimentally ( $a= 6.7788$  bohr) [7] is much smaller than the  $\text{Fe}_4\text{N}$  one ( $a= 7.1777$  bohr) [22]. Moreover, Matar *et al* [16] have carried out an experimental study, reinforced by theoretical calculations, on cobalt-substituted  $\text{Fe}_4\text{N}$  and have shown that Co can substitute Fe in both corners ( $\text{CoFe}_3\text{N}$ ) and face centers sites ( $\text{FeCoFe}_2\text{N}$ ) with no change in  $\text{Fe}_4\text{N}$  lattice constant value. Furthermore, considering the iron and cobalt atomic radii (Co is slightly smaller than Fe:  $r_{\text{Fe}}=2.212$  bohr and  $r_{\text{Co}}=2.193$  bohr [16]), one can conclude that  $\text{Fe}_4\text{N}$  and  $\text{Co}_4\text{N}$  must have nearly the same magnitude of lattice constant. As commented in the previous section and demonstrated by geometry optimization for N deficient system, this may be due to the



presence of N vacancies in the elaborated samples. The theoretical lattice parameter of the system in the magnetically ordered state, which is very close to the  $\text{Fe}_4\text{N}$  one [22], agrees well with the experimental study of Co-substituted  $\text{Fe}_4\text{N}$  [16]. However it is larger than the value reported for  $\text{Co}_4\text{N}$  [7].

The values of total energy at equilibrium show that the ferromagnetic state is preferred over the non magnetic one with  $\Delta E(\text{SP-NSP}) = -1.4$  eV per fu. This value is somehow close to the energy stabilization of SP ground state from US-PP calculations because the latter method accounts for valence electrons only. A magnetic volume expansion is visible whereby the theoretical lattice constant increases from 7.010 bohr in the non magnetic state at equilibrium to 7.098 bohr in the ferromagnetic state so that one expects the system to be more compressible in the spin polarized configuration. In the latter, we have obtained magnetic moment values of  $1.967\mu_B$ ,  $1.486\mu_B$  and  $0.069\mu_B$  for  $\text{Co}^I$ ,  $\text{Co}^{II}$  and N respectively. With respect to the average value of  $1.6\mu_B$  of Co, the enhancement of the magnetic moment of corner  $\text{Co}^I$  underlines its isolated character on one hand and the reduction of  $\text{Co}^{II}$  moment points to the chemical role of neighboring nitrogen on the other hand. Experimentally no reports about  $\text{Co}_4\text{N}$  magnetic moments are available. Compared to the data on  $\text{CoFe}_3\text{N}$ , our magnetic moments agree with reference [16] for cobalt  $\text{Co}^I$  (situated in the corners) and for nitrogen; whereas a large difference exists for  $\text{Co}^{II}$ . This fact is expected because the previous authors have substituted only one of the three face centers irons, and thus their  $\text{Co}^{II}$  moments ( $0.842\mu_B$ , see [16]), are largely affected by the neighbouring irons. Nevertheless, our  $\text{Co}^I$  moment is smaller than  $2.26\mu_B$  computed in ref. [17] for  $\text{CoFe}_3\text{N}$ .

### 1. Comparison of FCC-Co versus $\text{Co}_4\text{N}$

The magnetovolume effect of nitrogen insertion within  $\text{Co}_4\text{N}$  is illustrated in figure 3. Moreover, since  $\text{Co}_4\text{N}$  can be seen as FCC-Co with N atom at the body center, we have included the variation of the moment in FCC-Co in order to allow a better check of this effect. One can distinguish two different parts in figure 3. At low volume (below FCC-Co equilibrium), the  $\text{Co}^I$  moment rises rapidly to high value reaching  $1.909\mu_B$  at the FCC-Co theoretical equilibrium volume (just below the  $\text{Co}_4\text{N}$  one); while  $\text{Co}^{II}$  moment increases somewhat slowly and a value of  $1.486\mu_B$  is reached at equilibrium. Beyond the  $\text{Co}_4\text{N}$  equilibrium volume, the  $\text{Co}^{II}$  moment still increases while  $\text{Co}^I$  one seems to stabilize at high

volume. This variation can be assimilated to the moment variation of FCC-Co. At low volume the latter resembles  $\text{Co}^I$  and increases rapidly to a high value with increasing volume. However, beyond its theoretical equilibrium the FCC-Co moment varies like  $\text{Co}^{II}$ . As a consequence it can be concluded that the face centers Co atoms ( $\text{Co}^{II}$ ) show itinerant magnetism while the corners ones ( $\text{Co}^I$ ) exhibit a localized behavior. The observed behavior can be explained in terms of Co-N bond  $p$ - $d$  hybridization in the two different cobalt sites  $\text{Co}^I$  and  $\text{Co}^{II}$ . While  $\text{Co}^I$  is far from the cube center, it seems almost not affected by the nitrogen atom and shows a great sensibility to the onset of volume variation by acquiring a large moment value and it begins to stabilize only at high volume. On the contrary,  $\text{Co}^{II}$  is subjected to the chemical presence of neighboring N atoms at  $a/2$  and the strong  $p$ - $d$  hybridization makes its moment vary slowly with the volume until a sufficient value (beyond the FCC-Co equilibrium) where its variation begins to be more pronounced. From this, it can be concluded that the two kinds of moment variation in FCC-Co are simultaneously present in  $\text{Co}_4\text{N}$  and this is due to the insertion of N atom. We note finally that this behavior is observed, with almost the same manner as in  $\text{Fe}_4\text{N}$  versus FCC-Fe [41,42].

## 2. Analyses of the density of states and bonding properties

Using equilibrium values of lattice constants determined above, we examine the spin ( $\uparrow$  and  $\downarrow$ ) and site projected density of states (PDOS) accounting for site multiplicity in the formula unit:  $1\text{Co}^I$ ,  $3\text{Co}^{II}$  and  $1\text{N}$ . This is shown in figure 4. Further, for sake of comparison we present the DOS of three studied varieties of Co in figure 5 at equilibrium lattice volumes (cf. fig. 1). From the very low spin  $\uparrow$  DOS at the Fermi level due to the filling of the  $\text{Co}(3d^\uparrow)$  sub band up to 5 electrons, the major feature of the latter is its strong ferromagnetic behavior in all three varieties; we note the resemblance of BCC-Co DOS with BCC-Fe DOS [43]. Thus from the point of view of strength of magnetic behavior cobalt is like FCC-Ni whose pressure for the disappearance of magnetic moment is  $\sim 4000$  kbar, contrary to BCC-Fe which is a weak ferromagnet [43].

In  $\text{Co}_4\text{N}$  nitride, the energy shift between spin  $\uparrow$  and  $\downarrow$  populations, due to the exchange splitting is larger for  $\text{Co}^I$  than for  $\text{Co}^{II}$ . From Hankel spherical wave function energies  $E_{Hankel}$  which, in the ASW formalism, are relative to the middle of the band, we compute the following exchange splitting magnitudes:  $\Delta E_{Hankel}(\text{Co}^I)=2.74$  eV and  $\Delta E_{Hankel}(\text{Co}^{II})=1.42$

eV while an intermediate value of  $\Delta E_{Hankel}(Co) = 1.66$  eV is obtained for FCC Co. The magnitudes of spin splitting are proportional to the magnetic moments. They can be interpreted by the more localized PDOS of  $Co^I$  as with respect to  $Co^{II}$  atoms which undergo the chemical pressure from nitrogen with a subsequent lower magnetic moment. Due to the electronegative character of nitrogen its PDOS are found at lowest energy within the VB, between -10 and -6 eV. One common feature between tetracobalt nitride and FCC-Co is the very low DOS at the Fermi level for majority spin which means that both Co sites in the nitride contribute to make the system behave as a strong ferromagnet just like FCC-Co. Particularly one can notice the low intensity peak at  $\sim 0.5$  eV below the Fermi level in the spin  $\uparrow$  panel which arises from the  $Co^{II}$ -N  $\pi^*$  bonding. This is a major difference with  $Fe_4N$  [41,42] where the corresponding peak due to  $Fe^{II}$ -N  $\pi^*$  lies exactly at the Fermi level and contributes to assign a weak ferromagnetic behavior for the iron based nitride.

For further explanations of the quantum mixing between the valence states, the two atom chemical interactions are examined within the covalent bond energy ECOV plotted for the spin polarized ferromagnetic state in figure 6. Negative and positive ECOV magnitudes designate bonding and antibonding interactions respectively. Looking firstly at the bonding within the Co metal sublattices, due to the large filling of Co d states the  $Co^I$ - $Co^{II}$  interaction is found half bonding/half antibonding. The exchange splitting causes this interaction to stabilize for majority ( $\uparrow$ ) spin at lower energy so that it is found within the VB (panel 5-a) while its destabilization to higher energy for minority  $\downarrow$  spin (panel 5-b) places the antibonding states on the Fermi level. This means that  $\downarrow$   $Co^I$ - $Co^{II}$  interaction is more bonding than  $\uparrow$  one. The bonding with nitrogen is clearly dominant for face center cobalt and is found mainly in the lower part of the VB with  $\sigma$  like bonding sharp peaks at  $\sim -9$  eV and more extended  $\pi$ -like bonding in the  $[-9, -6]$  eV range. They are followed by antibonding counterparts  $\pi^*$  around  $E_F$  as suggested in the DOS discussion and  $\sigma^*$  above  $E_F$  within the conduction band at  $\sim 4$  eV. From the relative energy position between the two panels of fig. 6,  $Co^I$ -N and  $Co^{II}$ -N are little affected by the exchange splitting and they are both dominantly bonding within the VB so that nitrogen plays a role in stabilizing  $Co_4N$  nitride.

## IV. CONCLUSIONS.

In this work, we have studied the electronic and magnetic properties of  $\text{Co}_4\text{N}$  compound with respect to pure cobalt in three different structures: HCP, FCC and BCC. For the latter, the obtained results were found in good agreement with experimental and theoretical ones. Such a reinvestigation, especially for FCC-cobalt was very useful to understand the magnetovolume effects in  $\text{Co}_4\text{N}$ . We showed that the magnetic moment variation in the latter could be assimilated to FCC-Co. The low volume (low moment) and large volume (high moment) states of FCC-cobalt are found to be present, at the same time, in the two cobalt sites  $\text{Co}^{II}$  and  $\text{Co}^I$  of  $\text{Co}_4\text{N}$ . The  $\text{Co}^I$  moment appears very sensitive to the onset of volume variation, like cobalt at low volume, and reaches saturation rapidly, whereas the  $\text{Co}^{II}$  one begins to increase significantly only beyond equilibrium and without fully saturating. This behavior (low volume-low moment and large volume-high moment) is explained, via the density of states and chemical bonding analyses, in terms of Co-N relative distances and the corresponding d-p hybridization. The  $\text{Co}^I$  atom at  $a\sqrt{3}/2$  from N atom is not affected and is found rather localized while the  $\text{Co}^{II}$  one which is at only  $a/2$  from N is submitted to a *chemical pressure* due to nitrogen and has itinerant behavior. Although similarities with  $\text{Fe}_4\text{N}$  can be traced out, the overall DOS features point to a strong ferromagnetic nitride for both Co sites.

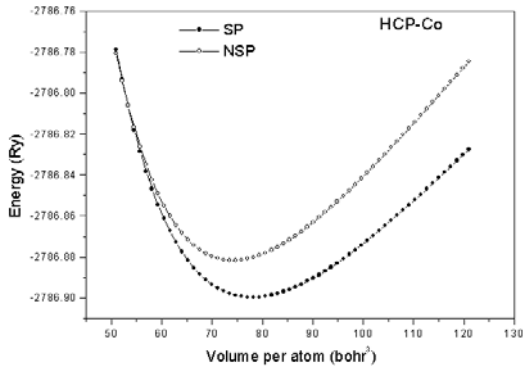
## V. ACKNOWLEDGEMENTS

Computations were partly carried out on main frame computers of the M3PEC-Mésocentre - University Bordeaux 1 Science and Technology (<http://www.m3pec.u-bordeaux1.fr>), financed by the “Conseil Régional d’Aquitaine” and the French Ministry of Research and Tecnology.

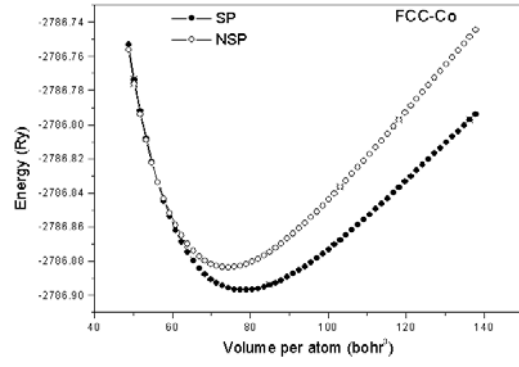
- 
- [1] D. Andriamandroso, G.Demazeau, M. Pouchard and P. Hagenmuller, J. Solid State Chem.**54**, 34 (1984).
  - [2] S. Matar, P. Mohn, G. Demazeau and B. Siberchicot, J. Phys. France **49**, 1761 (1988).
  - [3] S. Ishida and K. Kitawatase, J. Magn. Magn. Mater. **130**, 353 (1994).

- [4] A. Sakuma, J. Magn. Magn. Mater. **102**, 127 (1991).
- [5] R. Coehoorn, G.H.O. Daalderop and H.F.J. Jansen, Phys. Rev. B **48**, 3830 (1993).
- [6] M. Morito, O. Kenichi and I. Takashi, Appl. Phys. Lett **49**, 977 (1986).
- [7] Kiichi Oda, Tetsuo Yoshio and Kohei Oda, J. Mater. Science **22**, 2729 (1987).
- [8] M. Beshkova, G. Beshkov, M. Marinov, D. Bogdanov-Dimitrov, G. Mladenov, T. Tanaka and K.Kawabata , Mat. Manuf. Process **16**, 531 (2001).
- [9] O. Schmidt-Dumont and N. Kron, Angew. Chem. **67**, 231 (1955).
- [10] M. Hasegawa and T. Yagi, Sol. Stat. Comm **135**, 294 (2005).
- [11] W. De la cruz, O. Contreras, G. Soto and E. Perez-Tijerina, Rev. Mex. Phys **52**, 409 (2006).
- [12] A. Houari, S.F. Matar, M.A. Belkhir and M. Nakhl, Phys. Rev. B **75**, 064420 (2007).
- [13] K. Suzuki, T. Shinohara, F. Wagatsuma, T. Kaneko, H.Yoshida, Y. Obi and S. Tomiyoshi, Jour. Phys. Soc. Jap **72**, 1175 (2003).
- [14] H. Shimizu, M. Shirai and N. Zuzuki, J. Phys. Soc. Jap. **67**, 922 (1998).
- [15] P. Lukashev and Walter R. L. Lambrecht, Phys. Rev. B **70**, 245205 (2004).
- [16] S. Matar, L. Fournès, S. Chérubin-Jeannette and G. Demazeau, Eur. J. Solid State Inorg. Chem **30**, 871 (1993).
- [17] A. V. dos Santos and C. A. Kuhnen, J. Alloys and Compounds **321**, 60 (2001).
- [18] H. P. Meyer and W. Sucksmith, Proc. R. Soc. London, Ser. **A 207**, 427 (1951).
- [19] D. Bagayoko, A. Ziegler and J. Callaway, Phys. Rev. B **27**, 7046 (1983).
- [20] G. A. Prinz, Phys. Rev. Lett. **54**, 1051 (1985).
- [21] C. S. Yoo, H. Cynn, P. Soderlind and V. Iota, Phys. Rev. Lett **84**, 4132 (2000).
- [22] V. L. Moruzzi, P. M. Marcus, K. Schwartz and P. Mohn, J. Magn. Magn. Mater **54-57**, 955 (1986).
- [23] B. I. Min, T. Oguchi and A. J. Freeman, Phys. Rev. B **33**, 7852 (1986).
- [24] J. Donahue, *The structures of the elements* (Wiley, New York, 1976).
- [25] P. Modak, A. K. Verma, R. S. Rao, B. K. Godwal and R. Jeanloz, Phys. Rev. B **74**, 01210 (2006).
- [26] S. Fox and H. J. F. Jansen, Phys. Rev. B **60**, 4397 (1999).
- [27] Z. Ding, P. M. Thibado, C. Awo-Affouda and V. P. LaBella, J. Vac. Sci. Technol B **22**, 2068 (2004).
- [28] P. Honenberg and W. Kohn, Phys. Rev. **136**, B864 (1964); W. Kohn and L.J. Sham, Phys.

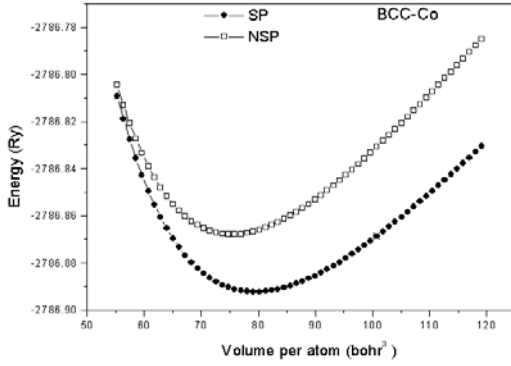
- Rev **140**, A1133 (1965).
- [29] J. Kübler, Phys. Lett. A **81**, 81 (1981).
- [30] J. P. Perdew and Y. Wang , Phys. Rev. B **33**, 8800 (1986).
- [31] V. Eyert, *The Augmented Spherical Wave Method – A Comprehensive Treatment, Lecture Notes in Physics* (Springer, Heidelberg, 2007).
- [32] J.P. Perdew, S. Burke, and M. Ernzerhof, Phys. Rev. Lett. **77** 3865 (1996).
- [33] R. Hoffmann, Angew. Chem. Int. Ed. Engl. **26**, 846 (1987).
- [34] R. Dronskowski and P. E. Blöchl, J. Phys. Chem. **97**, 8617 (1993).
- [35] G. Bester and M. Fähnle, J. Phys.: Cond. Matt. **13**, 11541 and 11551 (2001).
- [36] G. Kresse and J. Hafner, *Phys. Rev. B*, **47**, 558 (1993);  
G. Kresse and J. Hafner, *Phys. Rev. B*, **49**, 14251 (1994);  
G. Kresse and J. Furthmüller, *Comput. Mat. Sci.*, **6**, 15 (1996);  
G. Kresse and J. Furthmüller, *Phys. Rev. B*, **54**, 11169 (1996).
- [37] D. Vanderbilt, *Phys. Rev. B*, **41**, 7892 (1990).
- [38] P.E. Blöchl, O. Jepsen, and O. K. Anderson, *Phys. Rev. B*, **49**, 16223 (1994) .
- [39] G. Demazeau, A. Wang, S. Matar and J.D. Cillard. *High Pressure Research*, **12**, 343 (1994).
- [40] A. D. Becke and K. E. Edgecomb, J. Chem. Phys., **92**, 5397 (1990).
- [41] P. Mohn and S.F. Matar, J. Magn. Magn. Mater. **191**, 234 (1999).
- [42] S. F. Matar. J. Alloys and Compounds **345**, 72 (2002).
- [43] P. Mohn. *Magnetism in the solid State, An Introduction*, (Springer, Berlin, Heidelberg 2003).



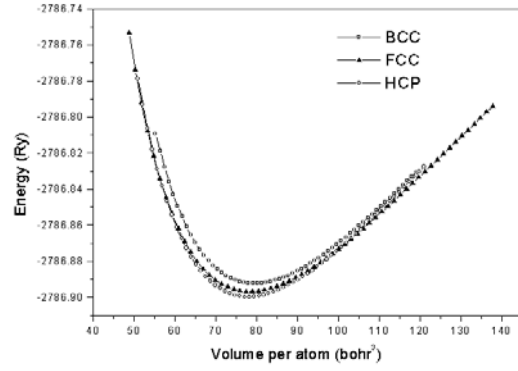
(a)



(b)



(c)



(d)

FIG. 1: Energy versus volume curves for non spin polarized (NSP ) and spin polarized (SP) cobalt in three allotropic varieties a) HCP, b) FCC, c) BCC and d) comparison of the three structures of Co in ferromagnetic ground state.

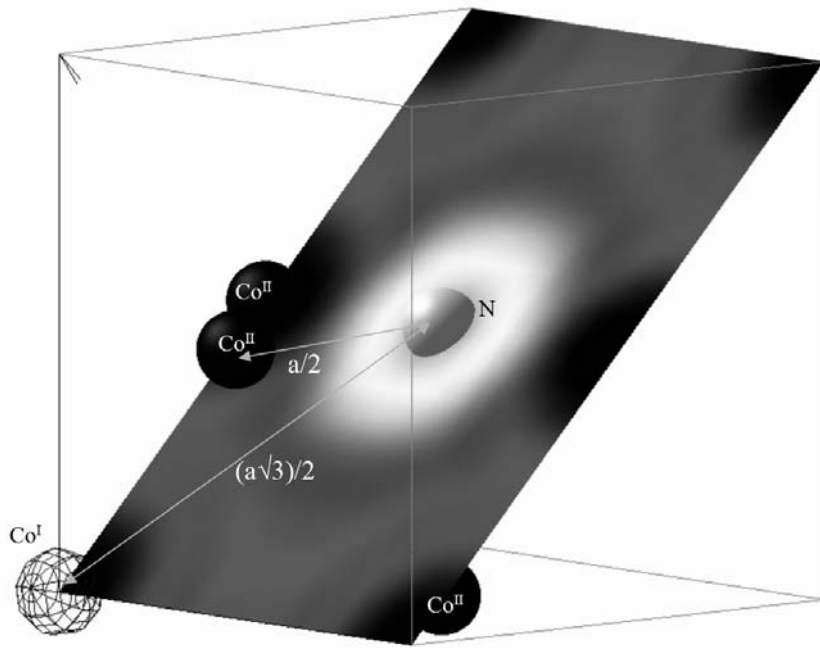


FIG. 2: Sketch of the anti perovskite cubic structure of  $\text{Co}_4\text{N}$  ( $\text{Co}^I\text{Co}_3^{\text{II}}\text{N}$ ) with a diagonal 101 plane showing electron localization function (ELF) contour plots. Strong localization (around N), free electron gas like distribution (within the lattice) and no localization (around cobalt sites) correspond to white, grey and black areas respectively.



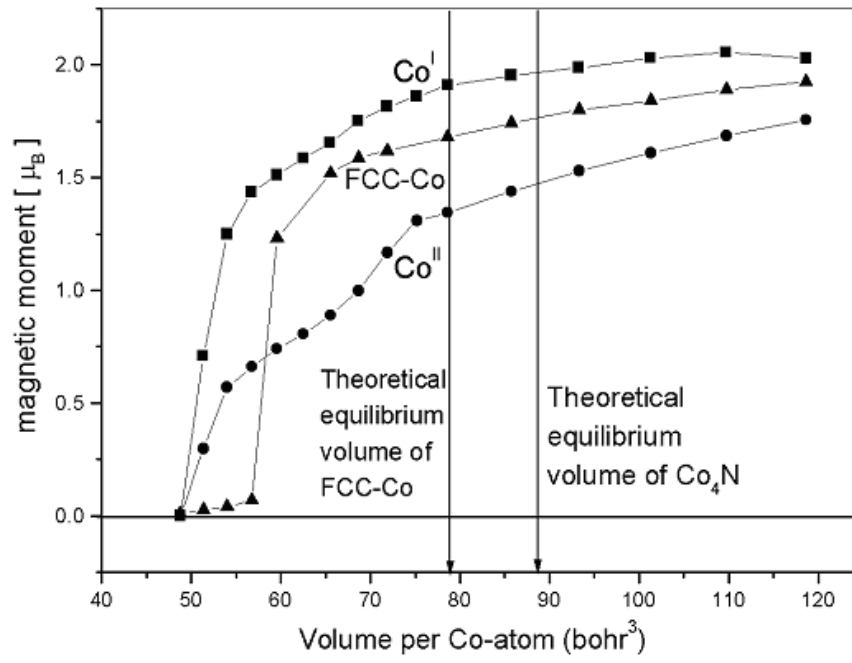


FIG. 3: Magnetic moment versus volume in  $\text{Co}_4\text{N}$  and FCC-Co.

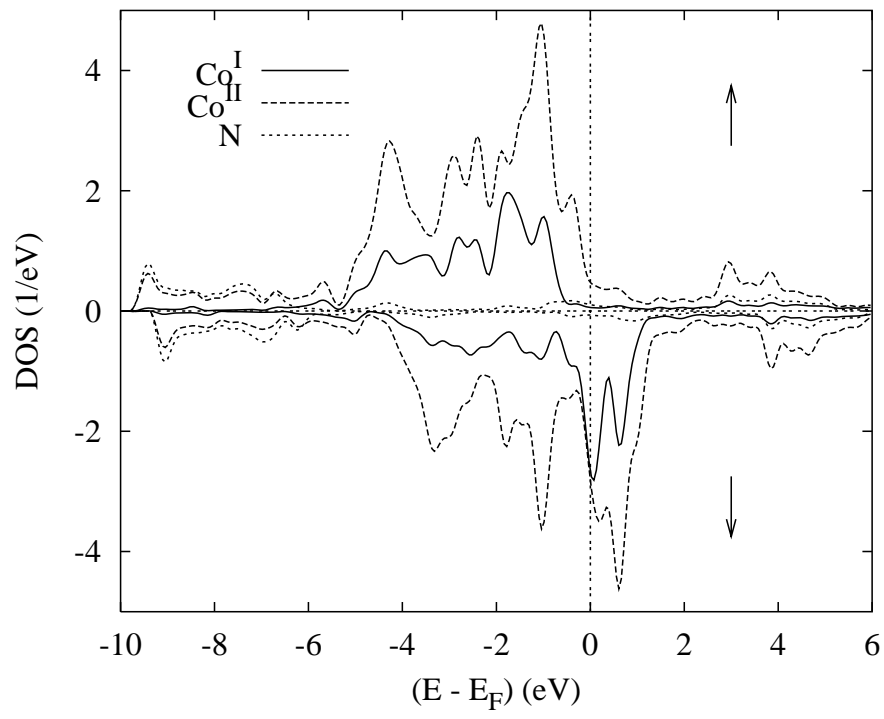
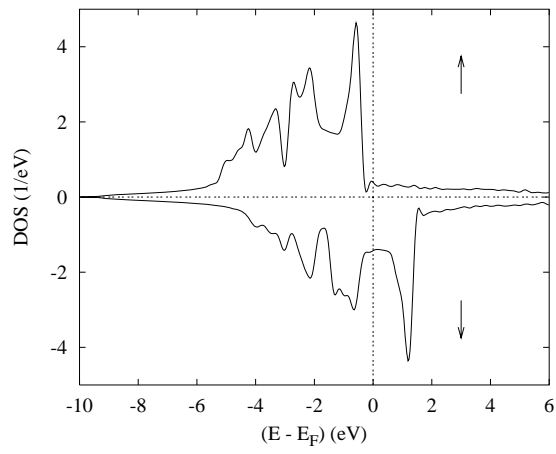
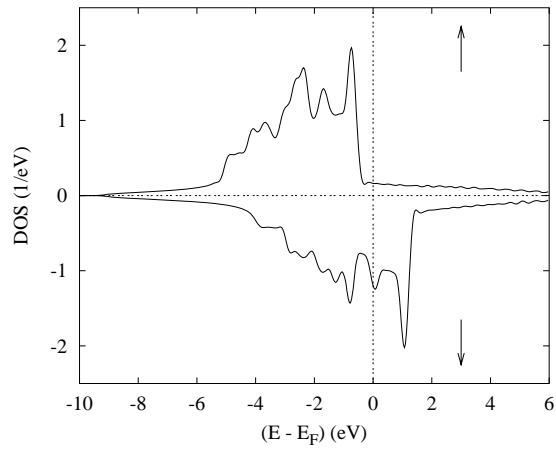


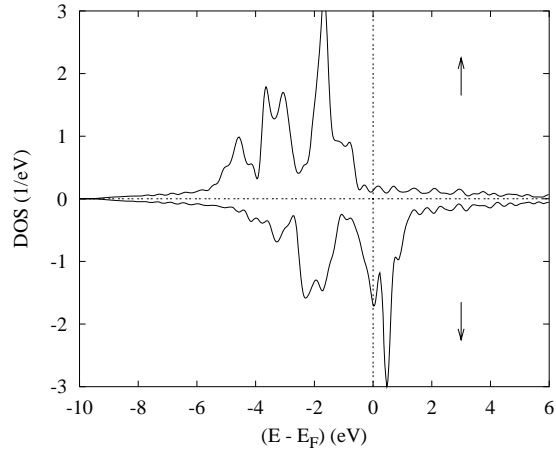
FIG. 4: Site and spin projected density of states (DOS) of spin polarized (SP)  $\text{Co}_4\text{N}$ .



(a)

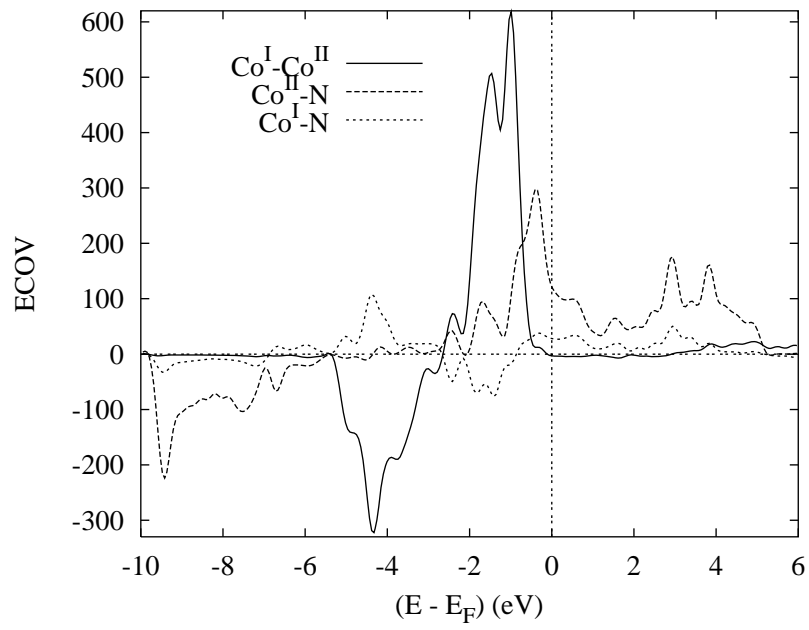


(b)

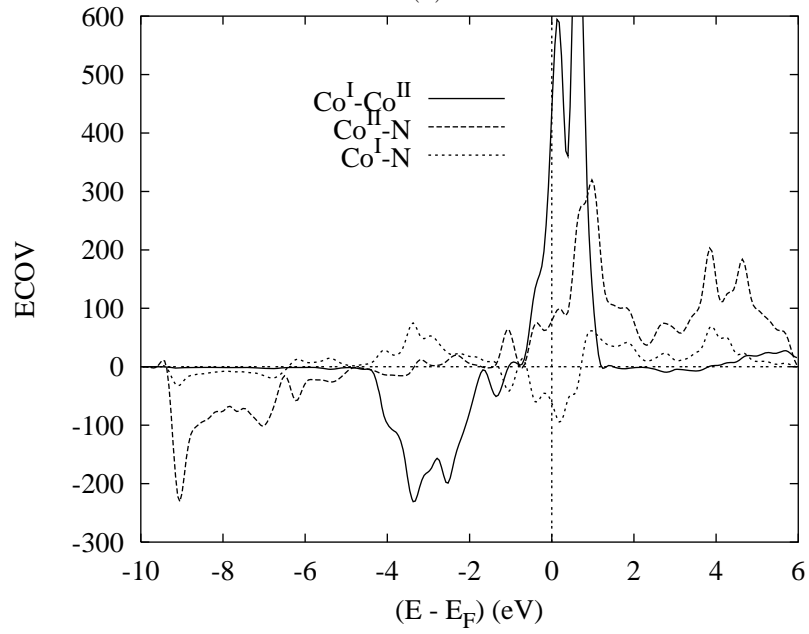


(c)

FIG. 5: DOS of spin polarized (SP) cobalt in three allotropic varieties (a) HCP, (b) FCC, (c) BCC



(a)



(b)

FIG. 6: Spin resolved chemical bonding from ECOV criterion within  $\text{Co}_4\text{N}$ : (a) majority spins ( $\uparrow$ ), (b) minority spins ( $\downarrow$ ).