



HAL
open science

Stability of Jahn Teller distortion in LaMnO₃ under pressure: An x-ray absorption study

Aline Y. Ramos, Hélio Tolentino, Narcizo M. Souza-Neto, Jean-Paul Itié,
Liliana Morales, Alberto Caneiro

► **To cite this version:**

Aline Y. Ramos, Hélio Tolentino, Narcizo M. Souza-Neto, Jean-Paul Itié, Liliana Morales, et al.. Stability of Jahn Teller distortion in LaMnO₃ under pressure: An x-ray absorption study. *Physical Review B: Condensed Matter and Materials Physics (1998-2015)*, 2007, 75, pp.052103. 10.1103/PhysRevB.75.052103 . hal-00128117

HAL Id: hal-00128117

<https://hal.science/hal-00128117>

Submitted on 30 Jan 2007

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Stability of Jahn-Teller distortion in LaMnO₃ under pressure: direct study by x-ray absorption

Aline Y. Ramos* and Hélio C. N. Tolentino

*Institut Néel, UPR 2940 - CNRS, 25 av. des Martyrs, BP166, 38042 Grenoble, France and
Laboratório Nacional de Luz Síncrotron - LNLS,
P.O. Box 6192, 13084-971, Campinas, São Paulo, Brazil*

Narcizo M. Souza-Neto

*Laboratório Nacional de Luz Síncrotron - LNLS,
P.O. Box 6192, 13084-971, Campinas, São Paulo, Brazil and
Dept. de Física dos Materiais e Mecânica, DFMT-IF-USP, São Paulo, SP, Brazil*

Jean-Paul Itié

Synchrotron SOLEIL, L'Orme des Merisiers, Saint-Aubin, BP 48, 91192 Gif-sur-Yvette Cedex, France

Liliana Morales and Alberto Caneiro

*Centro Atómico Bariloche and Instituto Balseiro,
Comisión Nacional de Energía Atómica and Universidad Nacional de Cuyo, 8400 S.C. de Bariloche, Argentine.*

(Dated: 30th January 2007)

The local environment of manganese atoms in LaMnO₃ under pressure up to 15.3 GPa has been studied by x-ray absorption spectroscopy. For pressures below 8 GPa, no change is detected within the MnO₆ octahedra. Above this pressure a continuous reduction of the long Mn-O distance takes place, however the octahedral distortion persists over the whole pressure range. At 15.3 GPa the average Jahn-Teller splitting of the distances is reduced by about one third, indicating that a total removal of the local Jahn-Teller distortion would occur only for pressures around 30 GPa, where metallization is reported to take place. A hysteresis in the long distance reduction is observed down to ambient pressure, suggesting the coexistence of MnO₆ distorted and undistorted units.

PACS numbers: 61.50.Ks, 61.10.Ht, 75.47.Lx, 71.30.+h, 75.47.Gk,

The physics underlining the remarkable properties of the manganite LaMnO₃ and its doped alloys is rich and complex. The actual path followed by a given system towards ferromagnetism and/or metallization, through chemical substitution, thermal treatment or pressure involves modifications of a delicate balance between delocalization and carriers trapping not yet completely understood. In the ground state LaMnO₃ is an antiferromagnetic semiconductor crystallizing in an orthorhombic variant of the cubic perovskite structure space group *Pbnm*. The MnO₆ octahedra are distorted due to the Jahn-Teller (*JT*) effect of the Mn³⁺(*t*_{2g}³*e*_g¹) and the *Mn* - *O* distances are split into 4 short bonds and 2 long bonds. In the basal *ab* plane long and short *Mn* - *O* bonds alternates. The apical and basal short bonds have different length, however this additional distortion is not resolved by local probes such as real space high resolution diffraction and x-ray absorption spectroscopy. The local radial distribution actually seen by these probes corresponds then to the *JT* splitting, with 4 oxygens at short distances (*Mn* - *O*)_s ≈ 1.94 Å and 2 oxygens at the long distance (*Mn* - *O*)_l ≈ 2.15 Å. LaMnO₃ undergoes a transition at *T** ≈ 710-750 K from the *JT* distorted orthorhombic phase O to a high temperature nearly cubic O' phase¹. The transition is accompanied by abrupt changes in the electrical resistivity and Weiss constant². The cell distortion is nearly removed and the orbital ordering disap-

pears in the O' phase, but the *JT* distortion of MnO₆ octahedra persists at the local scale^{3,4,5}. The transition then happens as an order-disorder transition, in agreement with the thermodynamic calculations⁶. More recently Qiu and co-workers⁷ reported on neutron powder diffraction measurements showing that the *JT* distortion of MnO₆ octahedra is maintained also in the high temperature rhomboedral phase (*T* ≥ 1010 K) and suggested the presence of ordered clusters with strong antiferrodistorsive coupling.

New insights for the role of the *JT* distortion can be obtained by the exploration of its pressure dependence. In LaMnO₃, the application of an external hydrostatic pressure produces a reduction of the lattice distortions and an enhancement of the carrier mobility^{8,9}. The *Mn* - *O* - *Mn* angle - tilt angle among two adjacent octahedra - is reported to decrease monotonically. The short bond distances (*Mn* - *O*)_s are weakly shortened with increasing pressure, the largest effect being a shortening of the long distance (*Mn* - *O*)_l. Besides, resistivity measurements⁸ reveal that the system keeps its insulating nature at room temperature up to 32 GPa where it undergoes an insulator-metal transition. From the extrapolation of x-ray diffraction data, it has been inferred that local *JT* distortion completely vanishes around 18 GPa⁸. However *ab initio* calculations using soft pseudo-

potentials recently predicted the conservation of local octahedral distortion well above this value¹⁰. The authors predict the occurrence of a structural phase transition around 15 GPa leading to a situation with a mixture of polytypes of antiferromagnetic order¹¹. Based on mean field theoretical calculations, Yamasaki and coworkers¹² also claimed that pressure induced metal to insulator transition in LaMnO₃ is not of Mott Hubbard type. They show that, according to calculations combining local density approximations and mean field theories, both on site repulsion and Jahn-Teller distortion are necessary for LaMnO₃ to be insulating below 32 GPa. The issue of local distortion in LaMnO₃ at high pressure is then not fully addressed.

Although the use of x-ray absorption spectroscopy (XAS), has been conclusive to elucidate many critical points of the local structure of LaMnO₃ and its doped alloys^{13,14,15,16}, no pressure dependence of the local order has been reported yet by XAS measurements, neither in the XANES (x-ray absorption near edge structure) nor in the EXAFS (extended x-ray absorption fine structure) range. This is principally due to inherent experimental difficulties in the collection of the XAS high pressures data at the manganese *K* edge, arising both from the low transmission of the diamond cells and from the strong additional absorption due to the La *L* edges. These difficulties were partially overcome here by using perforated-diamond cell^{17,18} and taking advantage of the high stability of a dispersive XAS setup. This setup is based on a bent crystal monochromator that opens up the energy band pass necessary for the spectroscopy and focuses it to the sample position. The dispersing band-pass is collected by a linear detector giving rise to a full spectrum at once. There is no moving optical element during experiments and an extremely good accuracy in the energy scale can be achieved¹⁹. Above 8 GPa the long distance ($Mn - O$)_l is continuously reduced, however an octahedral distortion persists over the whole pressure range. A hysteresis in the long distance reduction is observed down to ambient pressure, suggesting the coexistence of MnO₆ distorted and undistorted units.

The pressure dependent x-ray absorption measurements at the Mn *K* edge were performed at XAS dispersive beamline²⁰ of the LNLS (Laboratório Nacional de Luz Síncrotron, Campinas, Brazil). A polycrystalline powder sample of LaMnO₃ was synthesized by the liquid-mix method previously described²¹. The as-made sample was annealed at $T = 1000$ °C under oxygen partial pressure $p(O_2)$ during 24 h and then quenched at liquid nitrogen temperature. The T and $p(O_2)$ values were chosen in order to give an LaMnO₃ oxygen stoichiometric compound, according to high temperature thermodynamic measurements²². The lattice parameters obtained from the Rietveld refinement agree with those of literature²³. A fine grained powder sample was loaded in a 100 μm diameter hole of an iconel gasket mounted on perforated diamond as support to the 500 μm thick diamond anvils¹⁸.

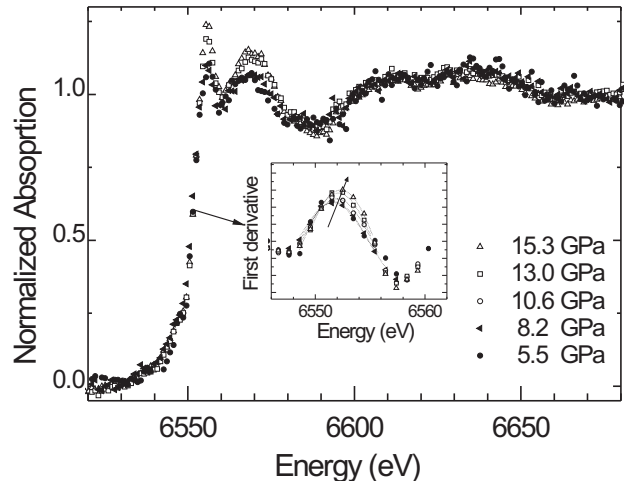


Figure 1: Mn *K* edge XANES spectra for LaMnO₃ at increasing pressures. Up to 8 GPa, the XANES are unchanged. Above 8 GPa, the structures just beyond the edge (6650 - 6680 eV) are slightly enhanced, whereas the absorption threshold is continuously shifted towards higher energies. The inset shows the spectra derivatives for a best observation of this last effect.

Silicone oil was used as pressure transmissive medium. Quasi-hydrostatic pressures up to 15.3 GPa were applied and controlled using ruby sphere with a precision of about 0.5 GPa. For each pressure, the cell was then realigned at the optical focus - of around 150 μm - and the spectra were collected in the transmission geometry. The beam path was set under vacuum, to reduce air absorption and beam harmonic contamination. To ensure the beam harmonic purity an additional gold coated mirror was added just behind the anvil cell and set to a grazing angle of 8 mrad. The energy resolution was about 1.5 eV, with energy calibration stable within 0.05 eV during the experiments.

The XANES spectra collected at various pressures exhibit slight modifications (Fig.1), indicating that the local atomic manganese environment is not drastically changed under external hydrostatic pressure. For pressures up to 8 GPa, no change at all is detected within the MnO₆ octahedra. In this pressure range a reduction of the cell volume has been observed by X-ray and neutron diffraction measurements^{8,9}. A continuous decrease of the $Mn - O$ bond length from ambient pressure to 8 GPa is also reported. Even if they are sharply contrasted, there may not be contradiction between the diffraction and the XAS results. On the first hand XRD provides the static periodicity of the structure averaged over a large spatial domain. On the other hand, as the characteristic time in the photoabsorption process is smaller than that the time corresponding to the thermal motion

of the atomic bonds, XAS probes the instantaneous short range structure around the absorbing atoms. A good example of complementarity of these approaches is given by LaMnO_3 on crossing the Jahn-Teller transition temperature. Diffraction methods show that the three bond converge into a single bond length on crossing T^{*1} , whereas XAS gives no change of all in the three bonds lengths^{3,4,5}. The dynamic nature of the Jahn-Teller transition has been deduced from the confrontation of both experimental evidences. By analogy with the temperature dependent measurements, we may assume here that when an external pressure above 8GPa is applied the main effect should be then the rearrangement among the octahedra, with possibly the formation of domains of disordered distortions, while the local instantaneous octahedron keeps almost unchanged. Such rearrangement would result in a reduction of the coherence length of the dynamical spatial fluctuations, and yield, in diffraction measurements, to the derivation of smaller average static values for the $Mn - O$ bonds.

Above 8 GPa, we observe a continuous shift of the absorption threshold towards higher energies, (Fig.1 and inset), along with the enhancement of the structures just above the edge (6650 - 6680 eV). As the manganese formal valence (Mn^{3+}) keeps unchanged with pressure, the edge shift (δE) expresses here modifications in the repulsive nearest neighbors potential arising from change in $Mn - O$ bonds in the coordination shell²⁴. Besides, as shorter bond lengths correspond to higher edge energies, the edge position is determined by the long distance $(Mn - O)_l \approx 2.15 \text{ \AA}$. δE is then here essentially related to the specific reduction in this bond. At the same time, the structures close to the edge are enhanced. It should be pointed out that, due to the selection rules in x-ray absorption spectroscopy, the K edge transition originates from the core $1s$ state to the projected np (mainly $4p$) unoccupied states. The enhancement in the structures close to the edge accounts for a change in the hybridization concomitant with the increasing overlap of the wavefunctions when the hydrostatic pressure is applied. This indicates a reduction of the local distortion of the Mn sites, leading to a reduction of the e_g splitting and a partial delocalization of the e_g electrons. Our results then agree with the intuitive idea, also confirmed by the x-ray diffraction measurements^{8,9}, that the short bonds $(Mn - O)_s$ should be less reduced than the long $(Mn - O)_l$ ones by the application of an external pressure.

Ab initio XANES calculations²⁵ of LaMnO_3 - based structures with progressive reduction of the $(Mn - O)_l$ bond reproduce well the experimental features (Fig.2). We should report that in simulations where long and short bonds are reduced in a same proportion, the positive edge shift is correctly reproduced but the structures at the edge are not enhanced. The evolution of the XANES features reflects then a continuous reduction of the average JT distortion from 8 GPa up to 15.3 GPa. The

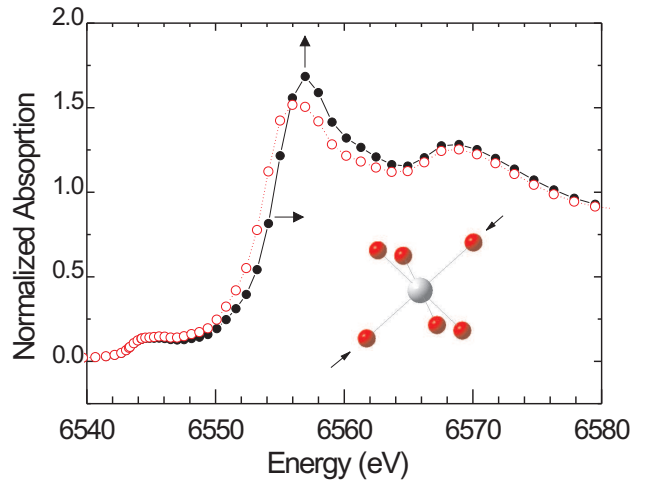


Figure 2: XANES simulations. Open symbols : LaMnO_3 structure. Plain symbols : LaMnO_3 variant where the $(Mn - O)_l$ bonds are reduced ($\approx 0.1 \text{ \AA}$). The edge is shifted ($\approx 0.9 \text{ eV}$) and the structures above the edge are enhanced.

evolution of the edge energy as a function of the applied pressure is given in figure 3. For small shifts the relationship between the reduction δR of a bond distance and the associated edge shift δE and is almost linear²⁴. An experimental calibration obtained for manganite systems²⁶ gives $\frac{\delta E}{\delta R} \approx 10 \text{ eV/\AA}$. We also obtain the similar calibration from our XANES simulations ($\frac{\delta E}{\delta R} \approx 9 \text{ eV/\AA}$, Fig.2).

The edge shift of 0.6 eV from 8 to 15.3 GPa corresponds then to a reduction in the long bond $(Mn - O)_l$ by about $0.06 \text{ \AA} \pm 0.02 \text{ \AA}$ over this range. Even if the short bonds $(Mn - O)_s$ were not reduced at all under pressure, the total suppression of the JT splitting would result in an energy shift at the edge of the order of 2 eV , which is not observed. Up to 15 GPa, the coherent JT distortion parameter $(\sigma_{JT})^2$ - defined as $\sigma_{JT} = \sqrt{1/6 \cdot \sum |R_i - R|}$ - decreases only by one third (Fig.3, inset). Local JT distortions are then present, even if the crystallographic structure suggests otherwise. Extrapolation of the data shows that the JT splitting would vanish only around 30 GPa, *i.e.* for applied pressures where the system is reported to undergo an insulator - metal transition⁸.

When the pressure is released down to ambient, a hysteresis is observed. At 9 GPa all features of the XANES spectrum are similar to those of the spectra at 13 GPa. Such hysteresis suggests the occurrence of a mixture of phases with close compositions and related structures²⁷. The hysteresis results in this case from the non-negligible elastic strain energy of coherent or semi-coherent interfaces that must be taken into account to describe the to-

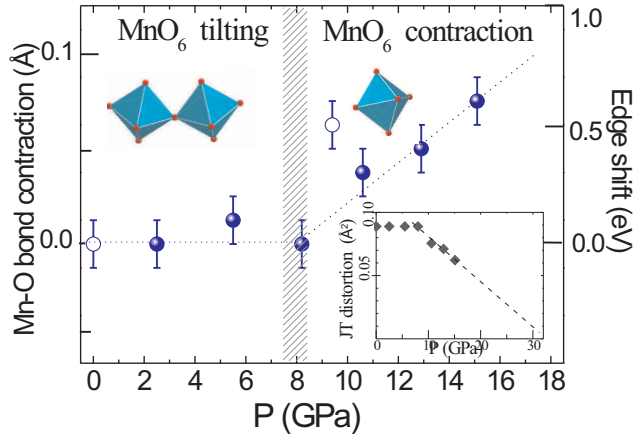


Figure 3: Relative changes in $Mn - O$ distance in the pressure range 0 to 15.3 GPa. The plain circles are obtained by increasing the pressure and the open circles correspond to the release. The curve in the inset gives an estimation of the JT distortion defined by $(\sigma_{JT})^2 = (\sqrt{1/6} \cdot \sum |R_i - R|)^2$ and predicts a suppression of the distortion above 30 GPa

tal Gibbs energy of the system. As the free motion of each MnO_6 octahedron is limited by the oxygen atoms shared with the adjacent units, electron phonon coupling may

involve several coupled Mn atoms²⁸. We should remind that XAS informs on the average Mn environment. The present data may then be examined considering that application of the pressure above 8 GPa, would induce the progressive formation of MnO_6 undistorted units coupled to distorted ones. The coexistence, at high pressures, of these larger polarons with the small MnO_6 JT distorted octahedra may account for the hysteresis behavior.

In summary, we studied the modifications in the average local distortion around the manganese atoms induced by application of high pressures in the range 0 to 15.3 GPa, from the modifications of x-ray absorption spectra in the near edge range. The MnO_6 octahedra keep unchanged by application of hydrostatic pressures up to 8 GPa, whereas above this value the long $(Mn - O)_l$ distance is continuously reduced. The Jahn-Teller bond splitting persists over the whole pressure range. The total quenching of this splitting is expected to take place only at pressures above 30 GPa, indicating that local JT distortion and insulator - to - metal transition should be closely related. A hysteresis in the XANES features when pressure is released suggests the coexistence of MnO_6 distorted and undistorted units.

Acknowledgments

This work is partially supported by LNLS/ABTLuS/MCT, CNPq and CNPq-CNRS agreement.

* Electronic address: aline.ramos@grenoble.cnrs.fr

- ¹ J. Rodríguez-Carvajal, M. Hennion, F. Moussa, A. H. Moudden, L. Pinsard, and A. Revcolevschi, Phys. Rev. B. **57**, R3189 (1998).
- ² J.-S. Zhou and J. B. Goodenough, Phys. Rev. B. **60**, R15002 (1999).
- ³ E. Araya-Rodriguez, A. Y. Ramos, H. C. N. Tolentino, E. Granado, and S. Oseroff, J. Magn. Magn. Mat. **233**, 88 (2001).
- ⁴ M. C. Sánchez, G. Subías, J. García, and J. Blasco, Phys. Rev. Lett. **90**, 045503 (2003).
- ⁵ R. A. Souza, N. M. Souza-Neto, A. Y. Ramos, H. C. N. Tolentino, and E. Granado, Phys. Rev. B **70**, 214426 (2004).
- ⁶ A. J. Millis, Phys. Rev. B. **53**, 8434 (1996).
- ⁷ X. Qiu, T. Proffen, J. F. Mitchell, and S. J. L. Billinge, Phys. Rev. Lett. **94**, 177203 (2005).
- ⁸ I. Loa, P. Adler, A. Grzechnik, K. Syassen, U. Schwarz, M. Hanfland, G. K. Rozenberg, P. Gorodetsky, and M. P. Pasternak, Phys. Rev. Lett. **87**, 125501 (2001).
- ⁹ L. Pinsard-Gaudart, J. Rodríguez-Carvajal, A. Daoud-Aladine, I. Goncharenko, M. Medarde, R. I. Smith, and A. Revcolevschi, Phys. Rev. B **64**, 064426 (2001).
- ¹⁰ G. Trimarchi and N. Binggeli, Phys. Rev. B **71**, 035101 (2005).
- ¹¹ T. Mizokawa, D. I. Khomskii, and G. A. Sawatzky, Phys. Rev. B. **60**, 7309 (1999).
- ¹² A. Yamasaki, M. Feldbacher, Y.-F. Yang, O. K. Andersen,

- and K. Held, Phys. Rev. Lett. **96**, 166401 (2006).
- ¹³ T. A. Tyson, J. MustredeLeon, S. D. Conradson, A. R. Bishop, J. J. Neumeier, H. Röder, and J. Zang, Phys. Rev. B **53**, 13985 (1996).
- ¹⁴ G. Subías, J. García, M. G. Proietti, and J. Blasco, Phys. Rev. B **56**, 8183 (1997).
- ¹⁵ C. H. Booth, F. Bridges, G. H. Kwei, J. M. Lawrence, A. L. Cornelius, and J. J. Neumeier, Phys. Rev. B **57**, 10440 (1998).
- ¹⁶ T. Shibata, B. A. Bunker, and J. F. Mitchell, Phys. Rev. B **68**, 024103 (2003).
- ¹⁷ A. Dadashev, M. P. Pasternak, G. K. Rozenberg, and R. Taylor, Rev. Sci. Instrum. **72**, 2633 (2001).
- ¹⁸ J. P. Itié, F. Baudelet, A. Congeduti, B. Couzinet, F. Farges, and A. Polian, J. Phys. Condens. Matter. **17** (2005).
- ¹⁹ H. Tolentino, E. Dartyge, A. Fontaine, and G. Tourillon, J. Appl. Crystallogr. **21**, 15 (1988).
- ²⁰ H. C. N. Tolentino, J. C. Cezar, N. Watanabe, C. Piamonteze, N. M. Souza-Neto, E. Tamura, A. Y. Ramos, and R. Neuschwander, Physica Scripta **115**, 977 (2005).
- ²¹ F. Prado, R. D. Sanchez, A. Caneiro, M. T. Causa, and M. Tovar, J. Solid State Chem. **146**, 418 (1999).
- ²² L. Morales and A. Caneiro, J. Solid State Chem. **170**, 404 (2003).
- ²³ F. Prado, R. Zysler, L. Morales, A. Caneiro, M. Tovar, and M. T. Causa, J. Magn. Magn. Mat. **196**, 481 (1999).

- ²⁴ C. Natoli, *EXAFS and Near Edge structure III* (Springer-Verlag, 1984), chap. 4, pp. 38–42.
- ²⁵ A. L. Ankudinov, B. Ravel, J. J. Rehr, and S. D. Conradson, *Phys. Rev. B* **58**, 7565 (1998).
- ²⁶ N. M. Souza-Neto, A. Y. Ramos, H. C. N. Tolentino, E. Favre-Nicolin, and L. Ranno, *Phys. Rev. B* **70**, 174451 (2004).
- ²⁷ N. Grunbaum, L. Moggi, F. Prado, and A. Caneiro, *J. Solid State Chem.* **177**, 2350 (2004).
- ²⁸ M. C. Sánchez, J. García, G. Subías, and J. Blasco, *Phys. Rev. B* **73**, 094416 (2006).