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Radio wavelength molecular observations of comets C/1999 T1 (McNaught-Hartley), C/2001 A2 (LINEAR), C/2000 WM₁ (LINEAR) and 153P/Ikeya-Zhang*

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ABSTRACT

We present a comparative study of the relative abundances of CO, CH_3OH , H_2CO , HCN, HNC, CS, H_2S , CH_3CN , SO and HNCO in comets C/1999 T1 (McNaught-Hartley), C/2001 A2 (LINEAR), C/2000 WM₁ (LINEAR) and 153P/Ikeya-Zhang, four of the brightest comets seen in 2001–2002. This investigation is based on millimetre/submillimetre observations made with the IRAM 30-m, SEST, CSO and Kitt Peak 12-m telescopes. Although these four comets are expected to originate from the Oort cloud, they present significant differences in molecular abundances, especially as regards to the most volatile species: CO and H_2S . In particular comet C/2000 WM₁ looks quite depleted in these volatiles, suggesting it may have a different origin than the others. Heliocentric variations of molecular relative abundance in the coma are also investigated. Significant increases in the CS/HCN and HNC/HCN production rate ratios with decreasing heliocentric distances are observed.

Key words. comets: general - radio lines: solar system - submilimetre

1. Introduction

The composition of cometary nuclei is of great importance for understanding their origin. For example, it is presumed that Oort-cloud comets were formed in the giant planet region (Jupiter-Neptune), before being expelled to the outer part of the Solar System. On the other hand, short period "Jupiterfamily" comets may have accreted directly in the Kuiper Belt beyond Neptune (Duncan et al. 2004). Having spent most of their time in a very cold environment, these objects should not have evolved very much since their formation. Thus, their composition provides clues to the composition in the regions of the Solar Nebula where they formed. The last decade has proven the efficiency of microwave observations in investigating the chemical composition of cometary atmospheres. About 20 different cometary molecules have now been identified at radio wavelengths (Bockelée-Morvan et al. 2004). Biver et al. (2002) presented a brief overview of the chemical diversity among 24 comets observed prior to 2002.

* Tables 1 to 5 are only available in electronic form at http://www.edpsciences.org

In the present paper, we extend this investigation to four of the brightest comets seen in 2001–2002. Comets C/1999 T1 (McNaught-Hartley), C/2001 A2 (LINEAR), C/2000 WM1 (LINEAR) and 153P/Ikeya-Zhang are all thought to originate from the Oort cloud but with likely different historical orbit evolutions. With orbital periods (before planetary perturbations) of 27000, 81000, 26000 and 360 years, respectively, all four comets are not new in the Oort sense (aphelia are well closer to the Sun than the mean distance to the Oort cloud), and have already experienced some alteration at previous perihelia. Comet 153P/Ikeya-Zhang is officially numbered as a short-period comet. It was previously observed by Johannes Hevelius in 1661 and likely also seen in 1273 and 877 (Hasegawa & Nakano 2003). C/2001 A2 and 153P have low inclination orbits (36° and 28° respectively) which cannot fully exclude an origin from the low inclination reservoir formed by the Kuiper Belt and Scattered Disk.

Section 2 presents the spectroscopic millimetre to submillimetre data obtained on these four comets. Section 3 discusses the various parameters and observational constraints used to derive the production rates. The comparison between the molecular abundances and the heliocentric evolution of the production rates is presented in Sect. 4.

2. Observations

C/1999 T1 and C/2000 WM₁ were discovered more than a year before perihelion so that coordinated radio observations could be scheduled in advance through regular allocations of telescope time. In contrast C/2001 A2 and 153P were observed on a target-of-opportunity time-line. These four comets were bright enough so that several other observing campaigns at infrared (e.g., Mumma et al. 2002) to UV (e.g., Feldman et al. 2002) wavelengths were conducted, providing complementary information to those presented here.

2.1. C/1999 T1 (McNaught-Hartley)

Comet C/1999 T1 (McNaught-Hartley) was discovered well ahead of perihelion, on 7 Oct. 1999 (McNaught & Hartley 1999) at 5.3 AU from the Sun. Being initially a southern object it moved northwards as it came to perihelion at 1.17 AU on 13 Dec. 2000. It reached perigee at 1.29 AU on 3 Feb. 2001. Although being an intrinsically active comet with a water production rate at perihelion reaching $Q_{\rm H_2O} = 10^{29}$ molec s⁻¹ (Crovisier et al. 2002), it did not become brighter than visual magnitude $m_1 = 7.7$, due to its relative large distance to the Earth and the Sun. The comet was first detected with the Swedish-ESO Submillimetre Telescope (SEST) 15-m radio telescope in Chile through its HCN J(3-2) line in September 2000. It was then observed at the Caltech Submillimeter Observatory (CSO) 10.4-m telescope on Mauna Kea on 5-7 Jan. 30, and 4-8 Feb. 2001 and at the Institut de Radio Astronomie Millimétrique (IRAM) 30-m facility in Spain, during the 24 Jan.-2 Feb. time interval. These latter observations suffered from bad weather on 24-30 January. The OH 18-cm lines were also daily observed with the Nancay radio telescope during the 16 Nov.-7 Dec. 2000 and 5-30 Jan. 2001 time intervals (Colom et al. 2004). Inferred OH production rates provide information on the water outgassing rate, which was also measured by the Submillimeter Wave Astronomy Satellite (SWAS) in Feb. 2001 (Bensch et al. 2004).

2.2. C/2001 A2 (LINEAR)

Comet C/2001 A2 was discovered on 15 Jan. 2001 by the Lincoln Near Earth Asteroid Research (LINEAR) project telescope of the Lincoln Laboratory (Massachusetts, USA). It was then only a non-promising faint 17th magnitude object at 2.3 AU from the Sun. But around 28 March it experienced a steep increase in brightness of 5 magnitudes followed by additional 1 to 1.5 magnitude short-lived outbursts around 11 May, 12 June and 12 July (Sekanina et al. 2002). These outbursts are likely connected to the release of fragments observed at the European Southern Observatory and elsewhere (Jehin et al. 2002, Sekanina et al. 2002). Comet C/2001 A2 reached naked eye visibility during two months with a peak brightness at $m_1 = 3.3$ in June. This was shortly after its perihelion on 24 May 2001 at $r_h = 0.78$ AU and before perigee on 30 June

at 0.24 AU. The initial surge in brightness made it a potentially interesting comet and target-of-opportunity observations were scheduled at IRAM on 8–10 July. It was also observed during short time intervals before sunrise at CSO on 16–19 June. Observations with the Kitt Peak National Observatory (KPNO) 12-m radio telescope on Kitt Peak (USA) were conducted on 5, 6, 11 and 12 June. The comet evolution was followed at Nançay between 2 April and 12 July, except during times of unfavourable OH-maser inversion. The peak outgassing rate occasionally exceeded 2×10^{29} molec s⁻¹ (Crovisier et al. 2002). The water line at 556.9 GHz was also observed and mapped with the Odin satellite on 27 April, and between 20 June and 7 July (Lecacheux et al. 2003).

2.3. C/2000 WM1 (LINEAR)

C/2000 WM₁ (LINEAR) was originally catalogued as a 18th magnitude asteroid when discovered on the 16th of November 2000 still at 5.8 AU from the Sun (Green 2000). This is the 46th comet discovery by LINEAR. With a perigee at 0.32 AU on 2 Dec. 2001 and a perihelion at only 0.55 AU on 22 Jan. 2002, it was a potentially interesting target. Around perigee it reached naked eye magnitude $m_1 \approx 5.5$ with $Q_{\rm H_2O} = 4 \times 10^{28}$ molec s⁻¹ (Lecacheux et al. 2003). Observations were scheduled around that time at IRAM (23-27 Nov.) and at CSO (3-8 Dec.), with cooperative weather. The 557 GHz water line was observed with Odin on 8 Dec. 2001 and 12 March 2002. Shortly after perihelion, on 30 Jan. 2002, the comet experienced a significant outburst of 3 magnitudes up to $m_1 = 2.8$. Nançay OH observations at 18 cm took place daily between 4 Oct. and 16 Dec. 2001 and resumed when the comet came back to higher declinations on 13 February till 20 April 2002 (Colom et al. 2004).

2.4. 153P/2002 C1 (Ikeya-Zhang)

This comet was co-discovered on 1 Feb. 2002 by two amateurs, Kaoru Ikeya (Japan) and Daquing Zhang (China) (Nakano & Zhu 2002) and was given the provisional designation C/2002 C1. It became quickly a bright object. It reached perihelion on 18 March at 0.51 AU and perigee on 29 April at 0.40 AU and with a visual magnitude of 3.4 at its brightest with a total outgassing rate around 9×10^{29} molec s⁻¹ (Dello Russo et al. 2004), it remained visible to the naked eye for nearly three months. It is a typical Halley-class comet. Soon after its discovery, its orbital period was determined to be close to 360 years and it was establish that this comet was likely the return of the historical comet observed in Europe in 1661 (Marsden & Nakano 2002).

Given the high interest of this new target, several observing programs were scheduled. At IRAM, observations took place at 3 periods (18–19 March, 29–30 April and 8–12 May) complemented by CSO observations (25–27 April) to follow the heliocentric evolution of the chemical abundances, especially the HNC/HCN ratio. The May run was hampered by bad weather which prevented an efficient deep search for molecular species only revealed in C/1995 O1 (Hale-Bopp) or C/1996 B2 (Hyakutake). The comet was also extensively observed with

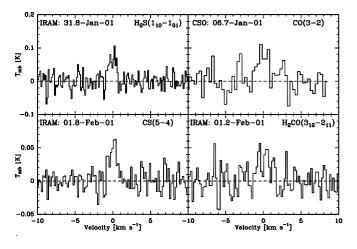


Fig. 1. Sample of molecular spectra obtained on comet C/1999 T1 (McNaught-Hartley).

Odin on 22–28 April: H_2O emission was mapped and $H_2^{18}O$ detected (Lecacheux et al. 2003). HCN *J*(3-2) was detected with the Kitt Peak 12-m on 29 March. Nançay OH 18 cm observations took place regularly between 29 Feb. and 20 June, excepted during periods when maser inversion was too unfavourable, especially in March.

2.5. Summary of observations

Table 1 lists the molecules and transitions included in our survey, in order of increasing frequency. The average half power beam width (HPBW) of each telescope used to observe the given molecular line is provided. Precise line frequencies and corresponding parameters (energy levels and line strength) were taken from the Cologne Database for Molecular Spectroscopy (CDMS, http://www.ph1. uni-koeln.de/vorhersagen/; Müller et al. 2005) and the JPL molecular Spectroscopy database (http://spec.jpl. nasa.gov/; Pickett et al. 1998).

Of the molecules listed in Table 1, eight species (HCN, CH₃OH, H₂CO, CS, H₂S, CH₃CN, HNC and CO) were searched for in all four comets. The five first species were detected in all comets: sample spectra of each comet are shown for HCN in Figs. 2, 4, 6 and 8, H₂CO, CS and H₂S in Figs. 1, 3, 5 and 7, and CH₃OH in Figs. 11, 12 and 13. In addition, CH₃CN was detected in C/2001 A2 and 153P (Fig. 10) and was marginal in C/2000 WM₁. HNC was detected in C/2001 A2 and 153P (Figs. 4 and 8) and CO was detected in C/1999 T1 and 153P (Figs. 1 and 7). Also, HNCO (Fig. 9) was detected in 153P and SO was marginally present in C/2001 A2 (Fig. 3). Finally, HC₃N, OCS and HCOOH were searched for in C/2001 A2 and 153P but no emission was detected beyond the 3- σ detection limit. A significant upper limit on the intensity of the HDO $(1_{10} - 1_{01})$ line at 509 GHz in 153P was also obtained at CSO.

Most lines observed or searched for in Table 1 were observed both at high resolution (20 to 100 kHz, in order to resolve the line with a resolution better than 0.1 km s⁻¹) and with a low resolution (1 MHz) wide band backend. Tables 2–5

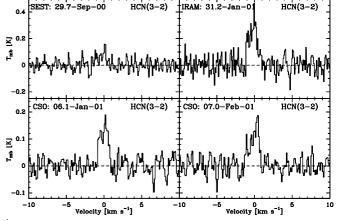


Fig. 2. HCN *J*(3-2) line at 265.9 GHz observed on comet C/1999 T1 (McNaught-Hartley).

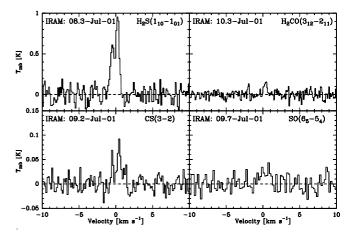


Fig. 3. Sample of molecular spectra obtained on comet C/2001 A2 (LINEAR).

provide the relevant information on the lines observed in these comets. For each observed line we provide the observing circumstances (date, heliocentric distance, geocentric distance and integration time: Cols. (1)–(4)) the molecule and transition, the integrated line intensity and rms or $3-\sigma$ upper limit and the velocity offset of the line. In the last column we give the positional offset that was used in the computation of production rates.

Ephemeris offsets were computed afterwards by comparing the ephemeris (either from the JPL's HORIZONS system or from the Minor Planet Center) used during observations and the latest available ephemeris. For some observations of HCN J(3-2) on C/2000 WM₁ at CSO and on 153P at IRAM we obtained coarse maps which helped us to pinpoint the maximum of brightness.

3. Data analysis

In most cases the production rates were derived following the methodology and parameters described in Biver et al. (1999a). To model the gas density in the coma, an isotropic steady-state outflow described by a Haser density radial profile is assumed.

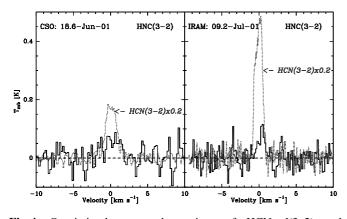


Fig. 4. Quasi-simultaneous observations of HCN J(3-2) and HNC J(3-2) lines on comet C/2001 A2 (LINEAR). The HCN J(3-2) line (dotted lines) intensity has been divided by 5 to fit with the vertical intensity scale in main beam brightness temperature.

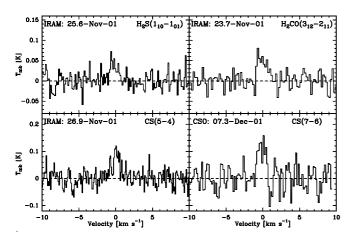


Fig. 5. Sample of molecular spectra obtained on comet C/2000 WM₁ (LINEAR).

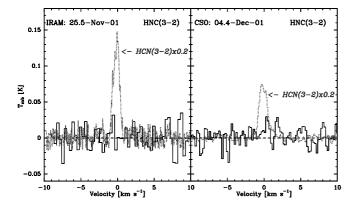


Fig.6. Simultaneous observations of HCN J(3-2) and HNC J(3-2) (not detected) lines on comet C/2000 WM₁ (LINEAR). The HCN J(3-2) lines (dotted lines) intensities have been divided by 5 to fit with the vertical intensity scale.

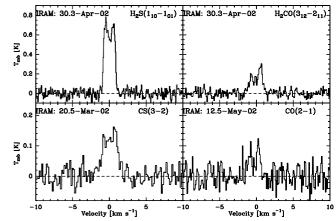


Fig.7. Sample of molecular spectra obtained on comet 153P/Ikeya-Zhang.

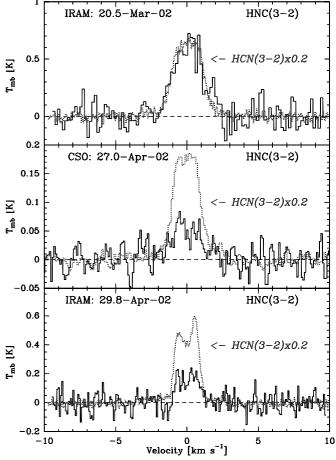


Fig. 8. Simultaneous observations of the HCN J(3-2) line (dotted line, plotted with an intensity divided by 5) and HNC J(3-2) line (plain line) on comet 153P/Ikeya-Zhang. One can readily see that the HNC/HCN line ratio decreased from about 20% in March to 5% at the end of April.

The velocity is assumed to be constant throughout the coma and deduced from line shapes. The number of molecules in the coma is also decreasing with distance to the nucleus due to photodissociation by solar UV (which scales as r_h^{-2}). The

populations of the molecular rotational levels are calculated from the model, taking into account collisional excitation with neutral gas at a constant temperature (see Biver et al. 1999a for assumed cross-sections), collisions with electrons (as

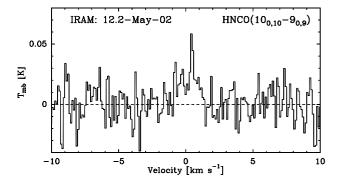


Fig.9. Detection of the HNCO line at 219.8 GHz on comet 153P/Ikeya-Zhang at IRAM.

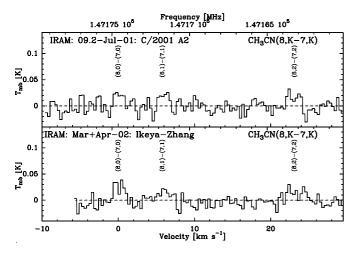


Fig. 10. Medium resolution spectra of three methyl-cyanide lines observed around 147 GHz in comets C/2001 A2 (LINEAR) and 153P/Ikeya-Zhang with IRAM. The velocity scale refers to the (8, 0)–(7, 0) line.

described in the same reference), and infrared pumping of vibrational bands for radiative excitation. Line intensities are then computed from a radiative transfer code by integrating the flux over the beam size of the antenna.

3.1. Photodissociation rates

Table 7 specifies the adopted HCN photodissociation rates at 1 AU from the Sun, based on the solar activity dependence as given in Crovisier (1994). The Lyman- α flux was estimated from the solar 10.7-cm flux monitored daily (www.sec. noaa.gov/ftpdir/indices/old_indices/), as suggested in Crovisier (1989). Solar activity was close to its maximum during the observations, so these lifetimes are smaller than in Biver et al. (1999a). The HNC lifetime is assumed to be the same as HCN. Other molecular lifetimes are not as sensitive to solar activity. The methanol photodissociation rate varied between 1.45 and 1.55×10^{-5} s⁻¹. Using the default CH₃OH value given in Biver et al. (1999a) leads to almost similar production rates. For H₂CO, CO, H₂S and CH₃CN, photodissociation rates are taken from Crovisier (1994), identical to those adopted in Biver et al. (1999a). In the case of CS we adopted a photodissociation rate $\beta_0(CS) = 2.0 \times 10^{-5} \text{ s}^{-1}$ at 1 AU from the

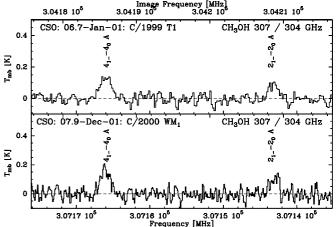


Fig. 11. Methanol lines simultaneously observed with the same tuning at 304 GHz (lower sideband) and 307 GHz (upper sideband) on comets C/1999 T1 (McNaught-Hartley) and C/2000 WM₁ (LINEAR) with the CSO. Frequency scale is upper sideband.

Sun, following recent constraints based on observations very close to the Sun presented in Biver et al. (2003). The effect of doubling $\beta_0(CS)$ (in comparison to Biver et al. 1999a) is investigated in Sect. 4.1. For SO we use $\beta_0(SO) = 1.5 \times 10^{-4} \text{ s}^{-1}$, from Bockelée-Morvan et al. (2000). Photodissociation rates for other molecules (OCS, HC₃N, HCOOH and HNCO) are also taken from Bockelée-Morvan et al. (2000).

3.2. Outflow velocities

The line shapes have been used to derive an estimate of the gas outflow velocity (v_{exp}) . Indeed, the line shapes are quite symmetric for most comets, suggesting isotropic outgassing. However, most lines observed in C/2000 WM₁ (Figs. 5, 6) are blue-shifted with Doppler shifts of -0.10 to -0.20 km s⁻¹ (Table 4, Col. 7). This indicates preferential sunward outgassing (the phase angle varied between 12 and 60°) but this asymmetry is not strong enough to justify the use of a fully asymmetrical model to derive production rates (no strong jet has been reported in the visible). The mean half-width at half maximum (HWHM) of the cometary lines with the best signalto-noise ratio was used to derive the expansion velocity (v_{exp}) . The adopted v_{exp} is actually about 10% lower than the HWHM to take into account spectral resolution and thermal broadening. We found $v_{exp} \approx 0.80$, 0.78, 0.75 and 0.86 km s⁻¹ at 1 AU, with a $r_{\rm h}^{-0.5}$ heliocentric dependence, for C/1999 T1, C/2001 A2, C/2000 WM1 and 153P, respectively. Values actually used are given in Table 7. The errors on the production rates resulting from rounding off or from the uncertainty on v_{exp} are less than 10%. The expansion velocity is actually expected to slightly increase throughout the coma (Combi et al. 2004), but in order to reproduce the observed line shape and given that molecules of very different lifetimes (e.g. H₂S and CO) have very similar line widths, the variation of the expansion velocity in the region of the coma sampled by the observations must be relatively small. Using a velocity profile such as those

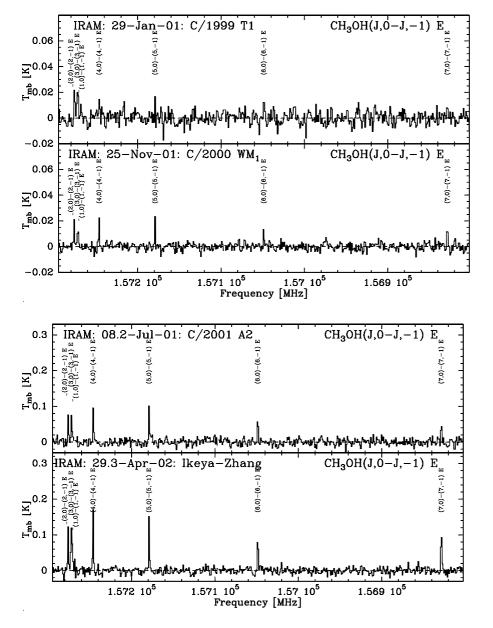


Fig. 12. Wide-band, low resolution spectra of methanol lines observed at 157 GHz on comets C/1999 T1 (McNaught-Hartley) and C/2000 WM₁ (LINEAR) with IRAM.

Fig. 13. Wide-band, low resolution spectra of methanol lines observed at 157 GHz on comets C/2001 A2 (LINEAR) and 153P/Ikeya-Zhang with IRAM.

of Combi et al. (2004) does not affect production rates by more than 10%.

3.3. Gas temperatures

When available, rotational temperatures (T_{rot}) of methanol or other species were used to derive the gas temperature *T*. Actual values measured in the four comets are given in Table 6. Some groups of methanol lines are particularly well suited to measure the gas temperature *T*: CH₃OH lines at 304/307 GHz (for T = 10-50 K) (Fig. 11), 157 GHz (for T = 10-80 K) (Figs. 12, 13) and 252 GHz (for T = 40-140 K) are the best. Other series of lines do not provide precise estimates of the gas temperature as they probe excitation conditions intermediate between thermal and fluorescence equilibrium. Inferred gas temperatures are given in Table 6, and the assumed values used to derive production rates are given in Table 7. The values we used follow roughly $T[K] = 60r_h^{-1}$ (153P and C/2001 A2) to $66r_{h}^{-1}$ (C/1999 T1 and C/2000 WM₁), which gives a good agreement to the measured values. A large uncertainty resides for the March observations of comet 153P at only 0.5 AU from the Sun. A steep increase (as $r_{\rm h}^{-1}$ or steeper) of the temperature of the coma when the comet approaches the Sun has been observed in a few comets (Biver et al. 1999a, 2000, 2002). However, below $r_{\rm h} \approx 0.6$ AU in comet Hyakutake, no significant further increase of T could be measured (Biver et al. 1999a). A weighted fit on the measurements of T in 153P (averaged by periods of 3 days) yields $T[K] = 57 \pm 4 \times r_{\rm b}^{-0.9 \pm 0.6}$, which extrapolates to 105 K at 0.5 AU. A temperature of 140 ± 10 K was actually measured in the infrared also at $r_{\rm h}$ = 0.5 AU (Dello Russo et al. 2004), but the infrared ro-vibrational temperatures tend to be higher than the gas temperature derived from radio measurements (see for example the values measured for C/2001 A2 around 9 July 2001 in Table 6 and in Dello Russo et al. (2005)). So we adopted 120 K as a compromise, which is consistent with the law $T[K] = 60r_{h}^{-1}$. In any

Table 6. Gas temperature measurements or collision rate constraints.

UT date	$\langle r_{\rm h} \rangle$	Molecular lines	Temperat	ture [K]	x_{ne}^1
[mm/dd.dd]	[AU]		Rotational	Kinetic ¹	
C/1999 T1 (M	cNaught-	Hartley) (2001)			
01/06.67	1.233	CH ₃ OH 304/307 GHz	52^{+36}_{-16}	60^{+40}_{-20}	
01/30.6-31.2	1.395	HCN(3-2)/(1-0)	21.1 ± 2.6		1.2 ± 0.4
01/31.2-31.2	1.398	HCN(3-2)/(1-0)	19.4 ± 2.0		0.55 ± 0.2
01/30.5	1.400	CH ₃ OH 157 GHz	39 ± 6	38 ± 6	
02/05.7-06.1	1.448	HCN(4-3)/(3-2)	32 ± 7		$1.4^{+1.8}_{-0.8}$
C/2001 A2 (LI	NEAR) ((2001)			
06/11.7	0.855	CH ₃ OH 157 GHz	>62	>50	
06/12.8	0.863	CH ₃ OH 145 GHz	37 ± 26		<1.1
07/08.2	1.143	HCN(3-2)/(1-0)	38 ± 5		$0.6^{+0.4}_{-0.3}$
07/08.23	1.144	CH ₃ OH 157 GHz	56 ± 5	51 ± 5	
07/10.2	1.169	HCN(3-2)/(1-0)	50 ± 6		$1.8^{+0.9}_{-0.6}$
C/2000 WM ₁ (LINEAF	R) (2001)			
11/24.0	1.343	HCN(3-2)/(1-0)	23.7 ± 2.3		0.3 ± 0.2
11/24.8	1.329	HCN(3-2)/(1-0)	58^{+21}_{-13}		14^{+18}_{-10}
11/25.4	1.320	CH ₃ OH 157 GHz	56 ± 7	50 ± 6	10
11/25.8	1.313	HCN(3-2)/(1-0)	23.0 ± 2.7		$0.2^{+0.15}_{-0.1}$
12/03.36	1.191	CH ₃ OH 304/307 GHz	36^{+9}_{-7}	35^{+9}_{-7}	
12/05.0	1.165	CH ₃ OH 242 GHz	74 ± 35	$150^{+\infty}_{-80}$	
12/07.7	1.121	HCN(4-3)/(3-2)	51^{+9}_{-7}		$1.5^{+1.3}_{-0.7}$
12/08.0	1.116	CH ₃ OH 304/307 GHz	48^{+18}_{-11}	46^{+17}_{-10}	
153P/Ikeya-Zh	ang (200	2)			
03/20.52	0.508	CH ₃ OH 157 GHz	99 ± 43	75 ± 35	
03/20.6	0.508	HCN(3-2)/(1-0)	67^{+16}_{-11}		$1.4^{+1.8}_{-0.9}$
04/25.59	0.977	CH ₃ OH 251 GHz	59 ± 5	72 ± 6	0.5
04/27.64	1.011	CH ₃ OH 251 GHz	87 ± 13	114 ± 20	
04/29.30	1.039	CH ₃ OH 157 GHz	62 ± 4	54 ± 4	
04/30.2	1.055	HCN(3-2)/(1-0)	40 ± 3		$0.9^{+0.3}_{-0.2}$
05/09.85	1.217	CH ₃ OH 157 GHz	40 ± 7	37 ± 6	-0.2
05/11.34	1.242	CH ₃ OH 157 GHz	35 ± 7	32 ± 6	
05/12.0	1.253	HCN(3-2)/(1-0)	31 ± 2		0.9 ± 0.2
05/12.08	1.254	CH ₃ OH 220/254 GHz	57^{+13}_{-8}	59 ± 10	

¹ A constraint on x_{ne} (see text) is given for observations that do not constraint the kinetic temperature but rather the collision rate.

case, in Sect. 4 we will investigate the influence of this uncertainty on T (±20 K for that period) on the production rates.

3.4. Collisional excitation

Table 7 also provides water outgassing rates measured with different means at the time of our observations, notably those derived from observations of the H₂O line at 556.9 GHz with the Odin satellite (Lecacheux et al. 2003; Hjalmarson et al. 2005). The Odin observations of H₂O are a good reference as the beam size (127") and type of lines (pure rotational) are more comparable to the observations analyzed in this paper than are the infrared, UV or decimetric observations used to derive the H₂O production rates. The computation of water production rates has been done with a similar model, too. We use these production rates and total collision cross-sections from Biver et al. (1999a) to compute neutral-neutral collision rates. The collisions with electrons are modelled as presented in Biver (1997) and Biver et al. (2000). Electron density and temperature are scaled according to the water production rate and the formulae in Biver (1997). Electron density is globally multiplied by a scaling factor " x_{ne} " independent of the distance to the nucleus

in the coma. This factor is constrained by the rotational temperatures that are sensitive to the collision rate rather than to the gas temperature (Table 6). The weighted average value for x_{ne} determined from the HCN observations is between 0.6 ± 0.5 and 1.1 ± 0.3 for the four comets with a mean value of ≈ 0.9 . On the other hand, a lower value (0.2) provides a good match to the spatial distribution of the intensity of the H₂O ($1_{10} - 1_{01}$) line observed in comets C/2001 A2 and 153P with the Odin satellite (Lecacheux et al. 2003; Biver et al. 2006). Derived x_{ne} are significantly sensitive to the assumed neutral-neutral crosssections. This may explain the different x_{ne} values found from HCN and H₂O observations. We thus adopted $x_{ne} \approx 0.5$ as a compromise for all comets.

3.5. Collisional versus radiative excitation of SO

We assume SO₂ as the main parent source of SO (Bockelée-Morvan et al. 2000), implying a parent scale-length of \approx 4500 km at $r_{\rm h} = 1.2$ AU. Infrared and UV pumping are likely marginally affecting the rotational populations due to relative weakness of the *g*-factors. Indeed, following Kim et al. (1999), the *g*-factor of the main A - X band should be on the

Table 7. Model parameters used.

UT J-4-	()	T			0	
UT date	$\langle r_{\rm h} \rangle$	Temperature	v_{exp}	$\beta_0(\text{HCN})$	$Q_{\rm H_{2}O}$	
[mm/dd-dd]	[AU]	in the coma	[km s ⁻¹]	$[s^{-1}]$	[molec s ⁻¹]	(Ref.)
	0	-Hartley) (2000/			20	
09/28-30	1.64	40 K	0.65	1.91×10^{-5}	4.0×10^{28}	(1)
01/03-07	1.23	55 K	0.70	1.80×10^{-5}	9.0×10^{28}	(1)
01/24-30	1.37	50 K	0.70	1.75×10^{-5}	6.5×10^{28}	(1)
02/00-02	1.41	45 K	0.70	1.71×10^{-5}	6.0×10^{28}	(1)
02/04-07	1.45	45 K	0.65	1.72×10^{-5}	5.5×10^{28}	(1)
C/2001 A2 (L	INEAR)	(2001)				
06/06	0.82	70 K	0.85	1.75×10^{-5}	1.0×10^{29}	(2)
06/11-12	0.86	70 K	0.85	1.80×10^{-5}	1.5×10^{29}	(2,3)
06/16	0.90	65 K	0.85	2.00×10^{-5}	1.2×10^{29}	(2,3)
06/17-19	0.92	60 K	0.80	2.00×10^{-5}	0.8×10^{29}	(2)
07/08-10	1.16	50 K	0.70	1.62×10^{-5}	0.4×10^{29}	(2)
C/2000 WM1	(LINEA)	R) (2001)				
11/23-27	1.33	50 K	0.65	1.80×10^{-5}	2.5×10^{28}	(4)
12/03-04	1.19	55 K	0.70	2.07×10^{-5}	3.7×10^{28}	(4)
12/05-08	1.14	60 K	0.70	2.05×10^{-5}	4.0×10^{28}	(4)
153P/Ikeya-Zl	hang (200)2)				
03/19-20	0.508	120 K	1.20	1.84×10^{-5}	7.0×10^{29}	(5)
04/24-25	0.96	63 K	0.88	1.75×10^{-5}	2.1×10^{29}	(6)
04/26-28	1.01	60 K	0.85	1.75×10^{-5}	1.8×10^{29}	(6)
04/29-30	1.05	55 K	0.82	1.71×10^{-5}	1.6×10^{29}	(6)
05/08-09	1.19	50 K	0.75	1.87×10^{-5}	1.0×10^{29}	(6)
05/11-12	1.25	48 K	0.75	1.87×10^{-5}	0.8×10^{29}	(6)

(1) From power law fit: 17.3 × 10²⁸ r_h^{-3.1} based on SWAS H₂O observations (Bensch et al. 2004).
 (2) From a power law fit 6.8 × 10²⁸ r_h⁻² based on Lecacheux et al. (2003) and Hjalmarson et al. (2005).

(3) Outburst from Crovisier et al. (2002).

(4) From Lecacheux et al. (2003) assuming $r_{\rm h}^{-2}$ dependence: $5.2 \times 10^{28} r_{\rm h}^{-2}$ molec s⁻¹.

(5) From Dello Russo et al. (2004).

(6) Power law fit $19.0 \times 10^{28} r_{\rm h}^{-4}$ based on Lecacheux et al. (2003) (Fig. 14).

Table 8. Comet C/1999 T1 (McNaught-Hartley) production rates.

UT date	$r_{\rm h}$	HCN	CH ₃ OH	СО	H ₂ CO	CS	H ₂ S	CH ₃ CN	HNC
[mm/dd.d]	[AU]			[1	0 ²⁵ molec	s ⁻¹]			
09/29.7	1.64	4.2 ± 0.6							
01/05.6	1.23	5.8 ± 0.7		2020 ± 330		6.8 ± 2.2			
01/06.6	1.23	7.1 ± 0.7	230 ± 10						
01/07.6	1.24		370 ± 190			<7.4			<1.9
01/25.5	1.35	5.8 ± 0.4	160 ± 30						
01/30.6	1.39	5.3 ± 0.4		<1370					
01/31.5	1.41	4.9 ± 0.6	115 ± 20	610 ± 70			33 ± 5	<3.1	< 0.7
02/02.1	1.42	2.9 ± 0.2	122 ± 10		19 ± 4	3.1 ± 0.4			
02/05.7	1.44	4.7 ± 0.4		700 ± 170		7.2 ± 1.5			
02/06.6	1.45	5.1 ± 0.4							< 0.4
02/07.6	1.46	4.7 ± 1.0							

order of 10^{-5} s⁻¹, which is also one of the main reasons why SO lines in the near UV (2500-2600 Å) have never been detected in comets in contrary to CS. In the case of SO observations in comets C/2001 A2 and 153P, more than 95% of the molecules in the beam are in a region (roughly less than 10000 km from the nucleus) where the collision rate with neutral gas is above 10⁻⁵ s⁻¹ (assuming a collisional cross-section $\sigma_c = 2 \times 10^{-14} \text{ cm}^{-2}$). Due to relatively large Einstein coefficients of the main lines $(1.335 \times 10^{-4} \text{ s}^{-1} \text{ for the } N_{\text{J}} = 5_6 - 4_5$ line), radiative decay should also dominate the UV pumping. In summary we have used the same excitation model as for other molecules but without UV radiative pumping. Assuming pure thermal equilibrium would only increase the production rates by ≈10%.

4. Results

4.1. Production rates and abundances

Production rates are given in Tables 8-11. Some are averages of production rates derived from several lines observed the same day (e.g. HCN J(3-2) and HCN J(1-0), or CH₃OH). We also

Table 9. Comet C/2001 A2 (LINEAR) production rates.

UT date	$r_{ m h}$	HCN	CH ₃ OH	CO	H ₂ CO	CS	H_2S	CH ₃ CN	HNC	SO
[mm/dd.d]	[AU]				[10 ²⁵	molec s ⁻¹]				
06/05.8	0.82					19 ± 2	77 ± 12			
06/11.8	0.86		434 ± 44							
06/12.8	0.86		393 ± 68		97 ± 18		119 ± 14			
06/16.7	0.90	13.7 ± 0.5								
06/18.6	0.92	8.2 ± 0.4							0.69 ± 0.10	
06/19.6	0.93	8.5 ± 0.6								
07/08.3	1.15	4.9 ± 0.2	112 ± 16	<149			46 ± 2			
07/09.3	1.16	6.2 ± 0.3				1.3 ± 0.2		1.1 ± 0.4	0.17 ± 0.03	4.7 ± 1.4
07/10.3	1.17	5.8 ± 0.6			9.5 ± 2.2					

Table 10. Comet C/2000 WM_1 (LINEAR) production rates.

				~ ~		-			
UT date	$r_{ m h}$	HCN	CH ₃ OH	CO	H_2CO	CS	H_2S	CH ₃ CN	HNC
[mm/dd.d]	[AU]				[10 ²⁵ 1	nolec s ⁻¹]			
11/23.1	1.36	1.6 ± 0.3							
11/24.1	1.34	1.8 ± 0.1			9 ± 1.6				
11/24.8	1.33	2.0 ± 0.3							
11/25.7	1.31	2.3 ± 0.2	30 ± 3	<38		0.5 ± 0.2	3.5 ± 0.5	0.4 ± 0.1	< 0.09
11/26.8	1.30	2.2 ± 0.2				1.2 ± 0.1			
12/03.3	1.19	3.1 ± 0.2	49 ± 6						
12/04.3	1.18	2.8 ± 0.2							< 0.10
12/05.3	1.16	2.9 ± 0.2	60 ± 17			1.8 ± 0.2			
12/06.3	1.15	3.1 ± 0.1			14 ± 2				
12/07.2	1.13	3.2 ± 0.3		<220		2.3 ± 0.3			
12/08.2	1.11	3.5 ± 0.2	60 ± 3		17 ± 5				

 Table 11. Comet 153P/Ikeya-Zhang production rates.

UT date	$r_{\rm h}$	HCN	CH ₃ OH	СО	H ₂ CO	CS	H_2S	CH ₃ CN	HNC
[mm/dd.d]	[AU]				[10 ²	⁶ molec s ⁻¹]			
03/19.7	0.51	10.0 ± 0.4				10.1 ± 1.1			
03/20.6	0.51	10.3 ± 0.7	104 ± 27	234 ± 52		9.8 ± 0.9		1.7 ± 0.6	2.6 ± 0.3
03/29.9	0.57	4.5 ± 0.5							
04/25.5	0.98	2.2 ± 0.2	49 ± 8						0.10 ± 0.02
04/26.5	0.99	2.1 ± 0.1							0.12 ± 0.03
04/27.5	1.01	2.0 ± 0.2	38 ± 13						0.09 ± 0.02
04/29.3	1.04	1.7 ± 0.0	38 ± 4			0.71 ± 0.05		0.21 ± 0.04	0.07 ± 0.01
04/30.3	1.06	1.7 ± 0.1	46 ± 14		5.8 ± 0.3		12.3 ± 0.2		0.09 ± 0.01
05/08.4	1.19	1.2 ± 0.1							
05/11.4	1.24	0.8 ± 0.1	22 ± 4			0.57 ± 0.09			HNCO:
05/12.1	1.26	0.8 ± 0.1	27 ± 6	34 ± 4					0.31 ± 0.05

took into account observational offsets and, when coarse maps were obtained, production rates obtained from the different positions were averaged (see data in Tables 2–5). Figure 14 shows the post-perihelion heliocentric evolution of molecular production rates in comet 153P. Water outgassing rates from other studies have been added for comparison. Several sources of possible uncertainties on these production rates have been investigated:

- Uncertainty on CS lifetime: the effect of using $\beta_0(\text{CS}) = 2.0 \times 10^{-5} \text{ s}^{-1}$, instead of $1.0 \times 10^{-5} \text{ s}^{-1}$ as in Biver et al. (1999a) is marginal ($\Delta Q/Q \le 10\%$). The most

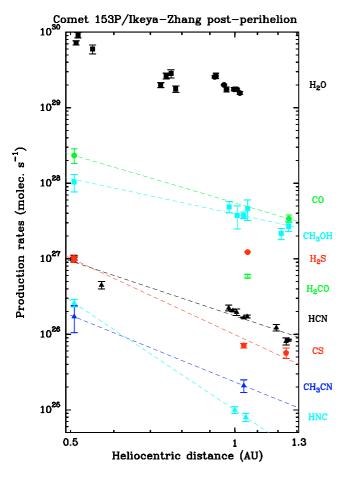


Fig. 14. Evolution of post-perihelion production rates in comet 153P/Ikeya-Zhang. Water production rates are from Odin (black dots, Lecacheux et al. 2003) and from infrared observations (black squares, Dello Russo et al. 2004).

sensitive cases are the observations of C/1999 T1 (for which CS production rates are increased by at most 15%) and 153P around perihelion (20% increase).

- Uncertainty on gas temperature: the larger uncertainty is for comets observed close to the Sun, i.e. within 0.8 AU, where gas temperatures are higher and harder to measure precisely. In fact, the average production rates based on multi-line observations (e.g. CH₃OH) are generally not so sensitive to the assumed temperature as they are based on rotational lines from a relatively wide range of energy levels. Otherwise the temperature uncertainty results in a 20% uncertainty in the production rates in the worst case.
- Uncertainty on collision rates: the electron density scaling factor x_{ne} is in some way a free parameter to get the best agreement between observations and model. Variations of x_{ne} in the range 0.2–1.0 will not change abundance ratios by more than 20%.

Figure 15 plots the result of the abundances relative to water from Table 12. It helps to see clearly where differences between comets are significant. The abundances found in comet C/1995 O1 (Hale-Bopp) (Bockelée-Morvan et al. 2000) are also plotted for comparison. These clearly show that for several molecules (H₂CO, H₂S, CS, HCN, HNC, SO and also

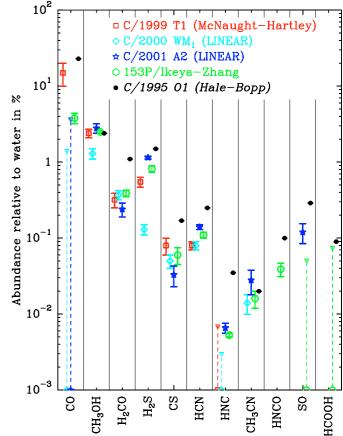


Fig. 15. Plot of the abundances relative to water of 11 molecules observed in the four comets or searched for in at least two of them. Data are from Table 12. For HNC and CS, whose abundances relative to water vary with heliocentric distance, values at $r_h \approx 1$ AU are plotted. Abundances measured in Hale-Bopp around perihelion are shown for comparison. Upper limits are drawn with $3 - \sigma$ error bars in dashed lines.

for HC_3N from Sect. 4.5) the abundances in comet Hale-Bopp were significantly higher than in any of the comets investigated in this paper.

4.2. Heliocentric variations of the abundances of CS and HNC

The observations of HNC and CS in these four comets strengthen the need to take into account variations with heliocentric distances. For C/2001 A2 and 153P, which were observed on a wider range of heliocentric distances than the others, significant heliocentric variations of the abundances of CS and HNC are found (Table 12).

The CS/HCN production rate ratio in comet 153P is proportional to $r_{\rm h}^{-0.7}$ (Fig. 14, Table 12), which is the same heliocentric dependence as found in comets Hyakutake (Biver et al. 1999a) and Hale-Bopp (Biver et al. 2002). In C/2000 WM₁, the CS/HCN ratio also increases with decreasing $r_{\rm h}$, following CS/HCN = $0.9 \times r_{\rm h}^{-2.6\pm0.6}$, but since the heliocentric distance changed only by 15% over the course of the observations, the slope is not very well constrained. The behaviour of C/2001 A2 in April–June 2001 was erratic due to frequent

Molecule	C/1999 T1 (McNH.)	C/2001 A2 (LINEAR)	C/2000 WM ₁ (LINEAR)	153P/IZ.
$\langle r_{\rm h} \rangle^1$	1.3	1.1	1.2	1.0
Abundance	e relative to water			
HCN	$0.08\pm0.01\%$	$0.14\pm0.01\%$	$0.08\pm0.01\%$	$0.11 \pm 0.01\%$
CH ₃ OH	$2.4\pm0.3\%$	$2.8\pm0.4\%$	$1.3\pm0.2\%$	$2.5 \pm 0.1 \times r_{\rm h}^{+0.8}\%$
CO	$15 \pm 5\%$	<3.7%	<1.4%	$3.8\pm0.6\%$
H ₂ CO	$0.32\pm0.07\%$	$0.24\pm0.05\%$	$0.37\pm0.05\%$	$0.39 \pm 0.04\%$
$H_2CO(p)$	$0.16\pm0.04\%$	$0.05 \pm 0.01\%$	0.09–0.18%	$0.10 \pm 0.01\%$
H_2S	$0.55\pm0.08\%$	$1.15\pm0.05\%$	$0.13\pm0.02\%$	$0.82\pm0.08\%$
CS	$0.08\pm0.02\%$	$0.07 \pm 0.01 \times r_{\rm h}^{-5}\%$	$0.05 \pm 0.01\%$	$0.067 \pm 0.016 \times r_{\rm h}^{-1.0}\%$
HNC	< 0.007%	$0.0066 \pm 0.0010 \times r_{\rm h}^{-2.9}\%$	<0.003%	$0.0053 \pm 0.0003 \times r_{\rm h}^{-2.6}\%$
CH ₃ CN	<0.052%	$0.028 \pm 0.010\%$	$0.014 \pm 0.004\%$	$0.016 \pm 0.004\%$
Abundance	e relative to HCN			
CH ₃ OH	31 ± 5	20 ± 2	16 ± 2	22 ± 6
CO	200 ± 80	<23	<17	36 ± 8
H ₂ CO	4.0 ± 1.0	1.6 ± 0.3	4.2 ± 0.5	3.6 ± 0.3
H_2S	6.3 ± 1.1	6.9 ± 0.7	1.6 ± 0.3	7.5 ± 0.5
CS	1.1 ± 0.2	0.2 ± 0.1	0.6 ± 0.1	$0.55 imes r_{ m h}^{-0.7}$
HNC	$< 0.08^{2}$	$0.052 \times r_{\rm h}^{-3.7}$	< 0.036 ³	$0.055 \times r_{\rm h}^{-2.2}$
CH ₃ CN	<0.6	0.19 ± 0.07	0.18 ± 0.04	0.14 ± 0.04
HC_3N		< 0.05		< 0.09
OCS	<125	<1.1		<2.0
HNCO		< 0.23		0.39 ± 0.08
SO		0.8 ± 0.2		<0.5
НСООН		<1.3		<0.8

Table 12. Compared relative production rates.

¹ Mean heliocentric distance (in AU) for the abundances, unless otherwise specified.

² at $r_{\rm h} = 1.42$ AU.

³ at $r_{\rm h} = 1.18$ AU.

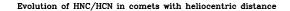
(p): parent distribution.

outbursts and fragment releases (Jehin et al. 2002). During the first half of June, Nançay OH observations showed variations of a factor of 3 or more in production rates from day to day. Thus, early June observations at the KPNO 12-m must be cautiously compared to June-July observations. Anyhow, they suggest a steep decrease of the CS/H₂O production ratio with heliocentric distance similar to that observed for 153P. The uncertainty on the CS photodissociation rate cannot explain the observed trend: as exposed in Sect. 4.1, the increase in the CS/HCN ratio in 153P between 1 and 0.5 AU can only be reduced from +65% down to +45% using a photodissociation rate twice lower. Snyder et al. (2001) even suggest a much higher photodissociation rate (5 times the value used here), probably unrealistic according to Biver et al. (2003): that would strongly increase the slope of the CS/HCN ratio versus heliocentric distance (to $r_{\rm h}^{-1.6}$). Otherwise, it is worth noting that this trend was also noticed with a different technique, i.e. from UV observations of comet 1P/Halley with IUE (Feldman et al. 1987; Meier & A'Hearn 1997). So the increase of the CS abundance in cometary comae close to the Sun is very likely and suggests that CS partially behaves as a low-volatility

molecule: either its expected main parent CS_2 is not easily released from the nucleus or another parent of CS (molecule, polymer or grains) releases additional CS only close to the Sun.

If we compare the CS/HCN production rate ratios at the same heliocentric distance (≈ 1.3 AU), then the average values are 1.1, 0.2, 0.5 and 0.5 for C/1999 T1, C/2001 A2, C/2000 WM₁ and 153P, respectively. This shows that C/1999 T1, only observed at this distance, is significantly richer in CS than the other comets.

The HNC/HCN production rate ratio exhibited a significant heliocentric dependence in comet Hale-Bopp (Biver et al. 1999b, Irvine et al. 1999). A production of HNC by chemical reactions in the coma was invoked by Rodgers & Charnley (1998): the presence of HNC in this comet and the increase of its abundance as r_h decreases was explained as due to the increase of outgassing rate and reaction efficiency. We find here again a steep ($r_h^{-2.2}$ to $r_h^{-3.7}$) evolution of the HNC/HCN ratio in comets 153P and C/2001 A2 that were \approx 50 times less productive than Hale-Bopp (Table 12). As shown in Fig. 16, the increase of the HNC/HCN production rate ratio at shorter heliocentric distances seems to be a common trend to all comets



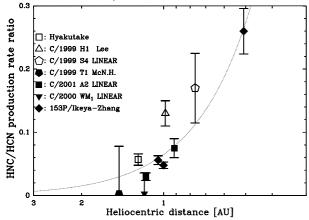


Fig. 16. The HNC/HCN production rate ratio as a function of heliocentric distance observed in 7 comets. The open symbols represent data previously published on comets of similar activity to those studied here: C/1996 B2 (Hyakutake) (Irvine et al. 1996), C/1999 H1 (Lee) (Biver et al. 2000) and C/1999 S4 (LINEAR) (Bockelée-Morvan et al. 2001). The dotted line corresponds to a weighted fit to all data yielding HNC/HCN = $0.06r_{\rm h}^{-2.1}$.

observed so far. However, according to Rodgers & Charnley (2001), the HNC/HCN ratio observed in such comets of lesser activity than Hale-Bopp cannot be explained by the same process.

Projecting a similar heliocentric dependence of the HNC/HCN ratio to the two other comets C/1999 T1 and C/2000 WM₁, all observations are compatible with a ratio of about 0.06 at 1 AU from the Sun (as compared to ≈ 0.2 in Hale-Bopp), varying roughly as $r_{\rm h}^{-2}$. The detection of HNC in C/2000 WM₁ by Irvine et al. (2003) (HNC/HCN = 0.08 at $r_{\rm h} = 1.08$ AU), a little closer to the Sun than were our observations, confirms this trend. The origin of HNC in cometary comae remains puzzling. It seems that the process releasing HNC in the coma is getting really efficient below $r_{\rm h} = 1$ AU, and it will be worth looking for measurements of the HNC/HCN production rate ratio at small heliocentric distances (Biver et al. 2003). If HNC is the photodissociation or thermal degradation product of a parent molecule or polymer (as proposed by Rodgers & Charnley 2001), one may expect the process to reach a maximum efficiency below some heliocentric distance and the HNC/HCN ratio will then reach a maximum value.

4.3. Evolution of the abundances of CO and CH₃OH relative to HCN and H₂O

In the case of comet 153P observations at 0.5 AU from the Sun, CO and CH₃OH abundances relative to HCN appear to be twice lower than around $r_h = 1.1$ AU. Given our assumed value of $Q_{\rm H_2O}$ in Table 7 the abundances of CO and CH₃OH relative to H₂O are also 1.3 and 1.7 times lower than at 1.1 AU. But the H₂O production rates were not measured with the same technique during all 153P observations: Dello Russo et al. (2004) finds $Q_{\rm H_2O} \propto r_h^{-3.2}$ while combining those data with Odin observation as in Fig. 14 yields $Q_{\rm H_2O} \propto r_h^{-2.1}$, which is the same slope as found for the CO production rate evolution with r_h .

So the ratios that significantly decrease towards the Sun are: CH₃OH/HCN, CH₃OH/H₂O and CO/HCN, while uncertainties on H₂O production rates are too large to be conclusive about the CO/H₂O and H₂O/HCN ratios. The decrease of the CO/HCN, CH₃OH/HCN and H₂O/HCN ratios with decreasing heliocentric distance were also observed in comet Hyakutake (Biver et al. 1999a). On the other hand the CH₃CN/HCN ratio does not vary with $r_{\rm h}$. A bias on the CO and CH₃OH production rates due to uncertainty on the gas temperature would be less than the observed trend. Possible explanations are not obvious, but if we assume that CO/H₂O does not really vary, HCN/H2O would then increase towards the Sun and one hypothesis would be that another source (e.g. HCN polymers) of CN-bearing molecules could become more efficient close to the Sun in releasing HCN, HNC and probably CH₃CN. The evolution of methanol abundance with heliocentric distance is still puzzling, but is observed for the third time: it was also noticed in Hale-Bopp (Biver et al. 1999b) that the CH₃OH/HCN ratio is higher at 2 AU than at 1 AU from the Sun. So it will be worth investigating in the future the evolution of the methanol abundance in cometary comae on a wide range of heliocentric distance.

4.4. Differences in composition between comets: is C/2000 WM₁ depleted in volatile species?

All four comets are different with respect to the abundances of HCN, CO, CH_3OH , H_2CO , H_2S and CH_3CN relative to water. Each exhibits at least one abundance ratio that distinguishes it from the others (Table 12, Fig. 15).

A question that is arising is the extent to which variations in coma abundances do reflects differences in nucleus ice composition. The nucleus is expected to be chemically differentiated in layers upon solar heating, with the upper layers depleted in the most volatile species. If this processing significantly altered chemical abundances and if depletion is directly linked to the sublimation temperature of the molecules, then abundances of species with high volatilities should be correlated. Similarly, if the composition of pre-cometary ices was governed by only volatility dependent condensation process, again, volatiles species should be correlated. Gibb et al. (2003) compared the CH₄ and CO abundances relative to water in a sample of 8 Oort cloud comets. These species have comparable volatilities, sublimating at 31 and 24 K, respectively. No apparent correlation is observed between CO and CH₄, with CO exhibiting much larger abundance variations than CH₄. These results suggest that the coma deficiency in hyper-volatiles in Oort cloud comets is not mainly related to chemical aging and also argue against temperature as the main factor controlling the composition of pre-cometary ices (Gibb et al. 2003).

Among the molecules studied in this paper, H_2S is the most volatile species after CO and CH₄, with a sublimation temperature ~57 K. As done for CH₄, it is interesting to study how CO and H_2S abundances correlate with each other. Figure 17 plots the H_2S/H_2O versus CO/H₂O relative abundances for seven comets: the four studied here plus three others in which both CO and H_2S were observed (from

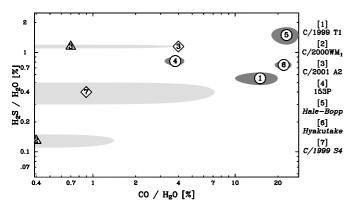


Fig. 17. Abundances of H_2S and CO relative to water in seven comets. Ellipses correspond to uncertainty domains ($\pm 1\sigma$ on each axis). In the case of CO, $3 - \sigma$ upper limits obtained in the radio are shown with light shaded ellipses. Circles (with comet number) correspond to the measured radio values, diamonds the infrared ones (Gibb et al. 2003; Mumma et al. 2002) and triangles the UV measurements (Weaver et al. 2003).

Biver et al. 1999a,b, 2000 and Bockelée-Morvan et al. 2001). CO was not detected in the radio in several comets but significant upper limits were obtained and its abundance was also estimated from UV or infrared observations (Weaver et al. 2003; Magee-Sauer et al. 2003; Mumma et al. 2002; Gibb et al. 2003). From Fig. 17, we can notice that:

- C/1999 T1 is CO-rich, with a CO abundance relative to HCN or water at least 4 times higher than in any of the three others. It has also an average H₂S abundance and belongs to the group of the volatile-rich comets like Hyakutake and Hale-Bopp (upper right corner).
- 153P has a lower CO abundance than the previous ones but is also relatively rich in H₂S. 153P is also a CH₄-poor comet (Gibb et al. 2003). Yet, even though it may have been originally volatile rich, given its relatively short orbital period, 153P possibly lost part of its CO and CH₄ after many passages close to the Sun.
- C/2001 A2 is relatively rich in H₂S, but has less CO than the previous comets. CO/H₂O abundance ratio published in the literature do not agree: 0.7% from UV observations (Weaver et al. 2003), but measured during an outburst and $\approx 4\%$ from infrared data (Gibb et al. 2003). In fact both studies infer about the same CO production rate at close times, but do not have the same reference H₂O production rate, because of the outburst. This comet may have had a similar history as 153P.
- C/2000 WM₁ is H₂S-poor, with the lowest H₂S/HCN or H₂S/H₂O ratio observed in any comet so far. This ratio is at least 3 times lower than any of the other comets. Although this determination relies on the observation of a single H₂S line, the detection is secure and was confirmed over several days with simultaneous HCN observations. It is also depleted in CO. Even if we only obtained an upper limit in the radio, it is a low one (CO/H₂O < 1.4%, Table 12) and UV observations suggest one of the lowest CO abundance found in any comet (0.4%, Weaver et al. 2003). To strengthen the case for a volatile-depleted comet, the

methanol abundance in C/2000 WM₁ is also clearly lower in this comet than in the three others (Fig. 15). Finally, Gibb et al. (2003) found that C/2000 WM₁ is also depleted in methane in comparison to the three other comets, Hale-Bopp, Hyakutake and C/1999 H1 (Lee). Only C/1999 S4 has a lower CH₄ abundance.

In summary, CO/H₂O and H₂S/H₂O abundance ratios do not correlate. While no CO-rich (>10%) comet has yet been observed with a low H₂S content, comets with CO mixing ratios ~5% or less exhibit high or low H₂S mixing ratios. 153P, despite its numerous passages close to the Sun, is still H₂S rich. C/2000 WM₁ is the first comet ever observed presenting a strong depletion in both CO, CH₄ and H₂S. The strong depletion in volatiles observed in comet C/1999 S4 led Mumma et al. (2002) to argue that this comet formed close to the Sun, in the Jupiter region: if so, C/2000 WM₁ might be another good candidate.

4.5. Abundances or upper limits of other molecules

The bottom of Table 12 provides results obtained for other molecules previously observed only in comets Hale-Bopp and Hyakutake. OCS, HC_3N and SO production rates were computed with the same model as for the main species, while a purely thermal model is used for HNCO and HCOOH. Abundances relative to HCN or upper limits are in most cases in the range of values measured in comets Hyakutake and Hale-Bopp:

- The 3- σ upper limits inferred for the HC₃N/HCN ratio in comets C/2001 A2 and 153P are comparable to the abundance measured in comet Hale-Bopp (Bockelée-Morvan et al. 2000). But upper limits on the HC₃N/H₂O ratio are three times and twice lower, respectively, than the ratio measured in comet Hale-Bopp. This suggests that the HC₃N abundance relative to water was relatively high in comet Hale-Bopp in comparison to other comets. A similar trend is also observed for the HC₃N/CH₃CN abundance ratio which is <0.3 in C/2001 A2 and <0.6 in 153P while slightly above one in Hale-Bopp.
- Significant upper limits on the OCS abundance were also obtained in comets C/2001 A2 and 153P. They are comparable to the values measured in Hyakutake (OCS/HCN ≈ 1) and Hale-Bopp (OCS/HCN = 1.6) (Bockelée-Morvan et al. 2000).
- HNCO, detected both in comets Hyakutake (Lis et al. 1997) and Hale-Bopp, was detected for the third time in 153P (Fig. 9). Its inferred abundance relative to HCN (Table 12, Lis et al. 1997; Bockelée-Morvan et al. 2000) is 0.4 in all three comets. The HNCO/H₂O production rate ratio is however lower than in Hyakutake and Hale-Bopp (0.04% versus 0.06 and 0.1%, respectively). The upper limits on the HNCO abundances in C/2001 A2 (HNCO/HCN < 0.23 and HNCO/H₂O < 0.035%) are in the low end range of the values measured in Hyakutake, Hale-Bopp and 153P.
- SO has been previously detected only in comet Hale-Bopp. It is marginally detected in comet C/2001 A2 (LINEAR)

(Fig. 3, Table 3). The inferred abundance relative to water is $0.11 \pm 0.03\%$. The upper limit obtained in comet 153P/Ikeya-Zhang is even lower (SO/H₂O < 0.05%), suggesting that Hale-Bopp was rather rich in SO or SO₂ (0.2–0.3%, Bockelée-Morvan et al. 2000).

- HCOOH was searched for in comets C/2001 A2 and 153P through its $J_{Ka,Kc}(10_{0,10} 9_{0,9})$ line at 220.038 GHz (same tuning as for SO and HNCO: around 220 GHz, Tables 1, 3 and 5). The upper limits obtained for the HCOOH/H₂O production rate ratio (0.2% and 0.08% respectively) are comparable to the abundance in Hale-Bopp. When comparing to methanol, we get HCOOH/CH₃OH < 7% (C/2001 A2), <3% (153P) and 4% in Hale-Bopp, suggesting that these comets are not significantly richer in HCOOH than comet Hale-Bopp.
- NS was only detected in comet Hale-Bopp (Irvine et al. 2000). We attempted a deep search in 153P/Ikeya-Zhang of the $N = \frac{11}{2} - \frac{9}{2e}$ and $N = \frac{11}{2} - \frac{9}{2f}$ lines at 253.569 GHz and 253.971 GHz, respectively. Summing the two lines (Table 5), we obtain an upper limit on the column density $N_{\rm col} < 3.6 \times 10^{11} \text{ cm}^{-2}$. The origin and lifetime of NS are unknown and we assume a photodissociation rate in the range $0.1-2 \times 10^{-4} \text{ s}^{-1}$ as in Irvine et al. (2000). Assuming a Haser parent density distribution, the corresponding upper limits in production rate are in the range $1.3-1.8 \times 10^{25}$ molec s⁻¹. This corresponds to an upper limit on the abundance of NS relative to water in 153P of 0.02%. In comet Hale-Bopp, Irvine et al. (2000) found NS/H₂O = 0.02% while Crovisier et al. (2004) found $NS/H_2O < 0.01\%$ from the same 254 GHz lines as observed here. So our upper limit in 153P is comparable to the values obtained in comet Hale-Bopp, but comparison may be hazardous since the origin of this radical in cometary atmosphere is unknown. If we now assume that NS is produced in the coma with a parent scale-length of 10⁴ km or more, our upper limit becomes more than twice higher than the corresponding abundance in Hale-Bopp due to the proximity and small beam with which we observed 153P.

4.6. HDO in 153P/lkeya-Zhang

The deuterium abundance in water is an important parameter for investigating the origin of cometary material (see e.g., Altwegg & Bockelée-Morvan 2003). Variations among comets may be expected if they did not formed in the same regions of the Solar Nebula (Hersant et al. 2001). We searched for the HDO (110 - 101) line at 509.292 GHz in comet 153P/Ikeya-Zhang using the CSO. Due to unexpected losses in telescope efficiency at this frequency, the search was not as efficient as anticipated and the line was not detected. Meanwhile, the 3– σ upper limit derived for the HDO production rate is $Q_{\rm HDO}$ < 1.0×10^{26} molec s⁻¹. On the same date (26 Apr. 2002), the observations of H_2O and $H_2^{18}O$ with the Odin satellite yield a confident value of $Q_{\rm H_2O} = (1.8 \pm 0.2) \times 10^{29} \, \rm molec \, s^{-1}$ (Lecacheux et al. 2003). Thus HDO/H₂O < $(5.6 \pm 0.6) \times 10^{-4}$ or $(D/H)_{H_2O}$ < $(2.8 \pm 0.3) \times 10^{-4}$ in 153P/Ikeya-Zhang. Our measurement does not confirm the high $(D/H)_{H_2O}$ value of $\approx 1.5 \times 10^{-3}$ determined by Gibb et al. (2002) in the same comet from a tentative detection of a HDO infrared line. Our upper limit is significant as the (D/H)_{H2O} ratio measured in other comets are, 2.6–3.4 × 10⁻⁴ in 1P/Halley (Eberhardt et al. 1995, Balsiger et al. 1995), 2.9 ± 1.0 × 10⁻⁴ in Hyakutake (Bockelée-Morvan et al. 1998) and (3.3 ± 0.8) × 10⁻⁴ in Hale-Bopp (Meier et al. 1998). It does not exclude the (D/H)_{H2O} to be lower in comet 153P than in other comets. Since comet 153P is CO-depleted compared to comets 1P/Halley, Hyakutake and Hale-Bopp, a low deuterium abundance could have suggested that comet 153P formed closer to the Sun than did the other three.

5. Conclusion

In the present study we reported the detection of 6 to 9 molecules in comets C/1999 T1 (McNaught-Hartley), C/2001 A2 (LINEAR), C/2000 WM₁ (LINEAR) and 153P/Ikeya-Zhang. In total, significant results on the relative abundances of up to 13 different molecules were obtained. This is one of the most extensive chemical survey of comets since Hyakutake and Hale-Bopp investigations. All comets discussed in the paper (including C/1999 S4 (LINEAR) and C/1999 H1 (Lee)) share a similar dynamical origin. They are long-period comets originating from the Oort cloud and are not at their first perihelion passage. 153P/Ikeya-Zhang may have experienced more alteration by solar heating, given its three recent recorded perihelion passages.

This study confirms previous evidences for chemical diversity among the Oort cloud population (Biver et al. 2002; Mumma et al. 2002). The variation of the H₂S and CO content between comets suggests that the deficiency is these hypervolatiles is not only related to comet aging upon solar heating. This is emphasized by the composition of C/2000 WM₁, which shows a much more severe depletion in both H_2S and CO than does 153P. Interestingly, C/2000 WM₁ is also less abundant in CH₄ than 153P (Gibb et al. 2003). Comparison between the H₂S and CO contents among comets shows that a deficiency in CO is not necessarily correlated with a deficiency in H₂S. Possibly C/2000 WM₁ formed in a relatively warm region of the solar nebula which partly prevented the condensation or trapping of H₂S and CO. But the absence of a clear correlation between volatility and abundance variations among comets (all parent molecules included) suggests that temperature and condensation were not the only factors that controlled the composition of cometary ices. Alternative explanations are debated as clathrate hydrates formation (Iro et al. 2003) or chemical processing in the Solar Nebula (Gail 2002).

Observations of HDO in comet 153P yield $(D/H)_{H_{2O}} < (2.8 \pm 0.3) \times 10^{-4}$, an upper limit equal to the D/H value measured in comets 1P/Halley, Hyakutake and Hale-Bopp. A low D/H ratio in 153P is thus not excluded. In addition, marginal detections or upper limits obtained on some of these comets suggest that CH₃CN, HC₃N, OCS, HNCO, SO and HCOOH may have been particularly abundant in comet Hale-Bopp.

The four comets studied in this paper were observed at several times to investigate the heliocentric evolution of molecular production rates. A few interesting results need to be emphasized:

- We confirm the heliocentric variation of the CS/HCN abundance ratio in $r_{\rm h}^{-0.7}$ observed in Hyakutake and Hale-Bopp from observations of 153P in the range $r_{\rm h} = 0.5-1.2$ AU.
- The monitoring of HNC/HCN in 153P and C/2001 A2 shows that this ratio is strongly $r_{\rm h}$ -dependent ($\approx r_{\rm h}^{-2.5}$ on average). All measurements obtained so far are consistent with HNC/HCN $\approx 6\%$ at 1 AU, the exception being comet Hale-Bopp ($\approx 15\%$ at 1 AU). Chemical processing in the coma was proposed to explain the same heliocentric trend observed in comet Hale-Bopp, but is inefficient in these less active comets (Rodgers & Charnley 1998). The HNC/HCN abundance in the coma seems related both to the heliocentric distance and (to a lesser extent) to the gas or dust production rate of the comet.
- We measure a factor of ≈2 decrease of the CH₃OH and CO abundances relative to HCN in 153P from 1.0 to 0.5 AU from the Sun. This was observed in comet Hyakutake and in C/2002 X5 (Kudo-Fujikawa) (Biver et al. 2003). However, at least part of this trend may be due to enhanced production of HCN at small heliocentric distances due to an extra source of CN-bearing species (since the HCN/CH₃CN does not vary).

The improvement of telescope sensitivity should further help in making extensive chemical investigations of comets. This will provide a more comprehensive view of the chemical composition and diversity of Solar System comets.

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N. Biver et al.: Comets C/1999 T1, C/2001 A2, C/2000 WM1 and 153P, Online Material p 1

Online Material

Table 1. Line frequencies and average beam sizes (HPBW).

	-				
Line	Transition	Freq.	IRAM	1 CSO F	KPNO
		[GHz]	["]	["]	["]
HCN*	1 – 0	88.632	26.6	-	
CH ₃ OH	$3_0 - 2_0 E$	145.094	17.0	_	41
CH ₃ OH	$3_{-1} - 2_{-1}E$	145.097	17.0	_	41
CH ₃ OH	$3_0 - 2_0 A^+$	145.103	17.0	_	41
CH ₃ OH	$3_2 - 2_2 E$	145.126	17.0	_	41
HC ₃ N	16 – 15	145.561	16.8	_	
OCS	12 – 11	145.947	16.8	_	
CS	3 - 2	146.969	16.6	_	40
CH ₃ CN	$8_3 - 7_3$	147.149	16.6	_	40
CH ₃ CN	$8_2 - 7_2$	147.163	16.6		40
CH ₃ CN	$8_{1} - 7_{1}$	147.172	16.6	_	40
CH ₃ CN	$8_1 - 7_1$ $8_0 - 7_0$	147.172	16.6	_	40
	$3_0 - 7_0$ $2_{11} - 1_{10}$		10.0	_	
H_2CO		150.498	15 4	_	39
CH ₃ OH	$7_0 - 7_{-1}E$	156.828	15.4	_	38
CH ₃ OH	$6_0 - 6_{-1}E$	157.049	15.4	-	38
CH ₃ OH	$5_0 - 5_{-1}E$	157.179	15.4	-	38
CH ₃ OH	$4_0 - 4_{-1}E$	157.246	15.4	-	38
CH ₃ OH	$3_0 - 3_{-1}E$	157.270	15.4	-	38
CH ₃ OH	$1_0 - 1_{-1}E$	157.272	15.4	-	38
CH ₃ OH	$2_0 - 2_{-1}E$	157.276	15.4	_	38
H_2S	$1_{10} - 1_{01}$	168.763	14.4	_	35
HNCO	$10_{0,10} - 9_{0,9}$	219.798	11.0		
SO	$N_J = 6_5 - 5_4$	219.949	11.0		
НСООН	$10_{0,10} - 9_{0,9}$	220.038	11.0		
CH ₃ OH	$8_0 - 7_1 E$	220.078	11.0		
H ₂ CO	$3_{12} - 2_{11}$	225.697	10.9	31	
CO	$\frac{3}{12}$ $\frac{2}{11}$ $2-1$	230.538	10.7	30.5	
	$5_0 - 4_0 E$	230.338 241.700	10.7	29	
CH ₃ OH					
CH ₃ OH	$5_{-1} - 4_{-1}E$	241.767	10.2	29 20	
CH ₃ OH	$5_0 - 4_0 A^+$	241.791	10.2	29	
CH ₃ OH	$5_1 - 4_1 E$	241.879	10.2	29	
CH ₃ OH	$5_2 - 4_2 A^+$	241.888	10.2	29	
CH ₃ OH	$5_2 - 4_E$	241.905	10.2	29	
CS	5 - 4	244.936	10.0	28.5	
CH ₃ OH	$9_3 - 9_2 A^{-+}$	251.360	9.8	28	
CH ₃ OH	$8_3 - 8_2 A^{-+}$	251.517	9.8	28	
CH ₃ OH	$7_3 - 7_2 A^{-+}$	251.642	9.8	28	
CH ₃ OH	$6_3 - 6_2 A^{-+}$	251.738	9.8	28	
CH ₃ OH	$5_3 - 5_2 A^{-+}$	251.812	9.8	28	
SO	$N_J = 5_6 - 4_5$	251.826	9.8	28	
CH ₃ OH	$4_3 - 4_2 A^{-+}$	251.867	9.8	28	
CH ₃ OH	$5_3 - 5_2 A^{+-}$	251.891	9.8	28	
CH ₃ OH	$6_3 - 6_2 A^{+-}$	251.896	9.8	28	
CH ₃ OH	$4_3 - 4_2 A^{+-}$	251.900	9.8	28	
CH ₃ OH CH ₃ OH	$4_3 - 4_2A$ $3_3 - 3_2A^{-+}$	251.906	9.8	28	
	$3_3 - 3_2 A$ $3_3 - 3_2 A^{+-}$	251.900 251.917		28 28	
CH ₃ OH			9.8		
CH ₃ OH	$7_3 - 7_2 A^{+-}$	251.924	9.8	28	
CH ₃ OH	$8_3 - 8_2 A^{+-}$	251.985	9.8	28	
CH ₃ OH	$9_3 - 9_2 A^{+-}$	252.090	9.8	28	
CH ₃ OH	$10_3 - 10_2 A^{+-}$	252.253	9.8	28	
NS	11/2 - 9/2e	253.571	9.7		
NS	11/2 - 9/2f	253.969	9.7		
CH ₃ OH	$2_0 - 1_{-1}E$	254.015	9.7		
HC ₃ N	28 - 27	254.700		27.5	
HCN*	3 - 2	265.886	9.4	26.5	22.5

Table 1. continued.

Line	Transition	Freq.	IRA	M CSO	KPNO
		[GHz]	["]	["]	["]
CH ₃ OH	$2_1 - 2_0 A^{-+}$	304.208	-	23.0	_
CH ₃ OH	$4_1 - 4_0 A^{-+}$	307.166	-	22.8	_
CS	7 – 6	342.883	-	20.6	_
CO	3 - 2	345.796	-	20.4	_
H_2CO	$5_{15} - 4_{14}$	351.769	-	20.0	_
HCN	4 – 3	354.505	-	19.8	_
HDO	$1_{10} - 1_{01}$	509.292	-	14.0	_

Notes: frequencies are from Müller et al. (2005) or Pickett et al. (1998) (see text).

"-" Means that this telescope is not equiped to observe this line.

* The SEST beam sizes at 89 and 266 GHz are 54" and 18.5" respectively.

 Table 2. Molecular observations in comet C/1999 T1 (McNaught-Hartley).

UT date	$\langle r_{\rm h} \rangle$	$\langle \Delta \rangle$	Int. time	Species (transition)	$\int T_b dv$	Velocity offset	Offset
[mm/dd.dd–dd.dd]	[AU]	[AU]	[min]		[K km s ⁻¹]	$[\text{km s}^{-1}]$	
SEST 15 m:				September 200			
09/28.64-30.74	1.641	2.056	280	HCN(3-2)	0.136 ± 0.018	-0.27 ± 0.10	4‴
09/28.64-30.74	1.641	2.056	280	HCN(1-0)	< 0.021		4‴
IRAM 30 m:				January–February			
01/24.46-24.53	1.345	1.303	60	HCN(3-2)	0.575 ± 0.085	-0.01 ± 0.08	4.7″
01/24.46-33.50	1.398	1.288	720	HCN(1-0)	0.063 ± 0.006	-0.08 ± 0.06	5.1"
01/24.46-33.50	1.395	1.291	500	CO(2–1)	0.058 ± 0.007	$+0.07 \pm 0.06$	5.1"
01/24.46-33.50	1.400	1.287	500	CH ₃ OH(1, 0–1, –1)E	0.033 ± 0.008	-0.41 ± 0.17	5.1"
				CH ₃ OH(2, 0–2, –1)E	0.029 ± 0.008	-0.35 ± 0.18	
				CH ₃ OH(3, 0–3, –1)E	0.058 ± 0.008	-0.13 ± 0.08	
				CH ₃ OH(4, 0–4, –1)E	0.045 ± 0.007	-0.04 ± 0.10	
				CH ₃ OH(5, 0–5, –1)E	0.040 ± 0.006	-0.34 ± 0.11	
				CH ₃ OH(6, 0–6, –1)E	0.033 ± 0.007	-0.07 ± 0.12	
				CH ₃ OH(7, 0–7, –1)E	0.028 ± 0.006	$+0.06 \pm 0.11$	
01/25.48-25.51	1.352	1.300	35	HCN(3-2)	0.380 ± 0.148		5.0"1
01/26.31-26.50	1.359	1.297	160	HCN(3-2)	0.425 ± 0.083	-0.11 ± 0.11	5.5"1
01/31.16-31.28	1.398	1.288	95	HCN(3-2)	0.433 ± 0.026	$+0.09 \pm 0.04$	5.0"
01/31.16-32.48	1.405	1.287	330	HNC(3-2)	< 0.076		5.1"
01/31.16-32.48	1.403	1.287	330	$H_2S(1_{10} - 1_{01})$	0.081 ± 0.012	-0.13 ± 0.08	5.0"
02/01.20-01.29	1.406	1.287	80	CS(3–2)	< 0.047		5.1"
02/01.20-02.50	1.412	1.287	205	CS(5–4)	0.083 ± 0.010	$+0.10\pm0.07$	5.2"
02/01.20-01.29	1.406	1.287	80	$H_2CO(3_{12} - 2_{11})$	0.053 ± 0.010	$+0.06 \pm 0.13$	5.1"
02/01.20-01.29	1.406	1.287	80	CH ₃ CN(8, 0+8, 1+8, 2)	< 0.076		5.1"
02/02.20-02.34	1.415	1.287	120	HCN(3-2)	0.266 ± 0.020	$+0.15\pm0.05$	4.7″
CSO 10 m:				January-February			
01/05.60-05.65	1.227	1.419	42.7	HCN(3-2)	0.186 ± 0.023	-0.06 ± 0.11	6.7″
01/05.66-05.72	1.228	1.417	58.7	CO(3–2)	0.154 ± 0.024	$+0.06 \pm 0.11$	6.0"
01/05.66-05.72	1.228	1.417	58.7	CS(7–6)	0.079 ± 0.025	-0.24 ± 0.23	6.0"
01/06.59-06.62	1.232	1.410	32.0	HCN(3-2)	0.240 ± 0.025	-0.02 ± 0.07	5.5"
01/06.63-06.71	1.233	1.410	72.0	CH ₃ OH(2, 1–2, 0)A–+	0.165 ± 0.018	-0.04 ± 0.05	5.5"
				CH ₃ OH(4, 1–4,0)A–+	0.220 ± 0.018	-0.14 ± 0.08	5.5"
01/07.61-07.68	1.238	1.402	64.0	HNC(3-2)	< 0.075		5.7"
01/07.69-07.71	1.238	1.401	16.0	CH ₃ OH(5, 0-4, 0)A+	0.175 ± 0.026	-0.16 ± 0.10	5.7"
				CH ₃ OH(5, -1-4, -1)E	0.080 ± 0.026	-0.32 ± 0.24	5.7″
				CH ₃ OH(5, 0-4, 0)E	0.139 ± 0.032		5.7"
				CH ₃ OH(5, 2–4, 2)E	< 0.096		5.7″
01/07.69-07.71	1.238	1.401	16.0	CS(5-4)	< 0.078		5.7″
01/30.60-30.65	1.393	1.289	52.0	HCN(3-2)	0.206 ± 0.017	$+0.37\pm0.06$	5.0"
01/30.66-30.71	1.393	1.288	48.0	CO(2-1)	< 0.049		5.0"
02/04.56-08.62	1.440	1.288	154.	CO(3–2)	0.070 ± 0.017	$+0.48\pm0.20$	5.8″
02/04.56-08.62	1.440	1.288	154.	CS(7–6)	0.087 ± 0.018	-0.16 ± 0.13	5.8″
02/05.58-05.64	1.444	1.288	50.7	HCN(3-2)	0.180 ± 0.020	-0.08 ± 0.07	6.0''
02/05.65-05.70	1.444	1.288	53.3	HCN(4-3)	0.226 ± 0.032	-0.06 ± 0.09	6.0''
02/06.58-06.61	1.452	1.289	37.3	HCN(3-2)	0.205 ± 0.018	$+0.00\pm0.06$	5.8″
02/06.62-07.70	1.457	1.290	150.	HNC(3-2)	< 0.018		5.9″
02/07.56-07.57	1.461	1.291	8.0	HCN(3-2)	0.188 ± 0.038	$+0.11 \pm 0.13$	6.0"

 1 Loss in efficiency due to frost on the antenna (beam degraded to $14^{\prime\prime}).$

Table 3. Molecular observations in comet C/2001 A2 (LINEAR).

UT date (2001)	$\langle r_{\rm h} \rangle$	$\langle \Delta \rangle$	Int. time	Species (transition)	$\int T_b dv$	Velocity offset	Offset
[mm/dd.dd–dd.dd]	(/h/ [AU]	(Δ) [AU]	[min]	Species (transition)	$[K \text{ km s}^{-1}]$	$[\text{km s}^{-1}]$	Onset
Kitt Peak 12 m:	[no]	[//0]	լոոոյ	June 2001:		[KIII 5]	
Kitt I Cak 12 III.				June 2001.			
06/05.71-05.77	0.815	0.419	60	$H_2S(1_{10} - 1_{01})$	0.114 ± 0.018	-0.04 ± 0.10	3″
06/05.79-05.87	0.816	0.418	84	CS(3–2)	0.167 ± 0.015	$+0.50\pm0.11$	6″
06/11.71-11.87	0.855	0.355	174	CH ₃ OH(1, 0–1, –1)E	0.059 ± 0.022	0.0 ± 0.0	11"
				CH ₃ OH(2, 0–2, –1)E	0.094 ± 0.024	0.0 ± 0.0	
				CH ₃ OH(3, 0–3, –1)E	0.114 ± 0.020	0.0 ± 0.0	
				$CH_3OH(4, 0-4, -1)E$	0.158 ± 0.027	0.0 ± 0.0	
06/12.71-12.75	0.863	0.346	30	$H_2S(1_{10} - 1_{01})$	0.251 ± 0.030	0.06 ± 0.08	<5‴
06/12.78-12.87	0.863	0.345	66	$H_2CO(2_{11} - 1_{10})$	0.074 ± 0.014	0.20 ± 0.14	5″
06/12.78-12.87	0.863	0.345	66	CH ₃ OH(3, 0–2, 0)A+	0.100 ± 0.023	0.0 ± 0.0	5″
				$CH_3OH(3, -1-2, -1)E$	0.094 ± 0.023	0.0 ± 0.0	
				CH ₃ OH(3, 0–2, 0)E	0.084 ± 0.021	0.0 ± 0.0	
				CH ₃ OH(3, 2–2, 2)E	0.055 ± 0.024	0.0 ± 0.0	
CSO 10 m: 06/16.66–16.67	0.807	0.210	107	June 2001:		+0.10 + 0.02	2.5//
	0.897	0.310	18.7	HCN(3-2)	2.095 ± 0.070 0.138 ± 0.020	$+0.10 \pm 0.03$	2.5"
06/17.64–19.68 06/18.63–18.64	0.916	0.294 0.294	104. 16.0	HNC(3-2)	0.138 ± 0.020 1.430 ± 0.060	$+0.12 \pm 0.12$	2.5" 2.5"
06/19.63-19.64	0.916 0.926	0.294	9.3	HCN(3–2) HCN(3–2)	1.430 ± 0.000 1.520 ± 0.110	$+0.08 \pm 0.03$ $+0.06 \pm 0.06$	2.5 2.5″
IRAM 30 m:	0.920	0.280	9.5	July 2001:		$\pm 0.00 \pm 0.00$	2.3
07/08.19-08.20	1.143	0.267	10	HCN(3-2)	2.530 ± 0.105	$+0.01 \pm 0.02$	4.2"
07/08.19-08.34	1.143	0.267	100	HCN(1-0)	0.209 ± 0.013	-0.07 ± 0.02	4.2"
07/08.19-08.34	1.144	0.267	100	CO(2–1)	< 0.076	0.07 ± 0.05	4.2"
07/08.19–08.27	1.144	0.267	60	$CH_3OH(1, 0-1, -1)E$	0.060 ± 0.013	-0.36 ± 0.12	4.2"
01/00119 00127		0.207	00	$CH_3OH(2, 0-2, -1)E$	0.194 ± 0.013	-0.01 ± 0.04	
				CH ₃ OH(3, 0–3, –1)E	0.202 ± 0.013	-0.10 ± 0.03	
				CH ₃ OH(4, 0–4, –1)E	0.207 ± 0.014	-0.09 ± 0.03	
				CH ₃ OH(5, 0–5, –1)E	0.194 ± 0.014	-0.09 ± 0.03	
				CH ₃ OH(6, 0–6, –1)E	0.195 ± 0.013	-0.11 ± 0.03	
				CH ₃ OH(7, 0–7, –1)E	0.158 ± 0.013	-0.05 ± 0.04	
07/08.21-10.35	1.157	0.274	355	HNC(3-2)	0.098 ± 0.017	$+0.29 \pm 0.10$	4.3"
07/08.30-08.34	1.145	0.268	40	$H_2S(1_{10} - 1_{01})$	0.926 ± 0.036	-0.06 ± 0.02	4.2"
07/09.17-09.33	1.157	0.274	140	HCN(1-0)	0.270 ± 0.011	-0.04 ± 0.02	4.3"
07/09.17-09.33	1.157	0.274	140	CH ₃ CN(8, 0–7, 0)	0.041 ± 0.008	$+0.07\pm0.09$	4.3"
				CH ₃ CN(8, 1–7, 1)	0.040 ± 0.008	$+0.11\pm0.10$	
				CH ₃ CN(8, 2–7, 2)	0.043 ± 0.008	-0.02 ± 0.11	
07/09.17-09.33	1.157	0.274	140	CS(3–2)	0.063 ± 0.007	$+0.16\pm0.06$	4.3"
07/09.17-10.30	1.163	0.277	220	$SO(6_5 - 5_4)$	0.029 ± 0.009	$+0.20 \pm 0.19$	4.3"
07/09.17-10.30	1.163	0.277	220	$HNCO(10_{0,10} - 9_{0,9})$	< 0.029		4.3"
07/10.16-10.18	1.169	0.280	30	HCN(3-2)	3.015 ± 0.050	-0.05 ± 0.04	4.3"
07/10.16-10.35	1.170	0.280	155	HCN(1-0)	0.220 ± 0.010	-0.12 ± 0.02	4.3"
07/10.16-10.35	1.170	0.280	75	$H_2CO(3_{12} - 2_{11})$	0.079 ± 0.018	$+0.35\pm0.14$	4.3"
07/10.16-10.35	1.170	0.280	155	OCS(12–11)	< 0.018		4.3"
07/10.16-10.35	1.170	0.280	155	HC ₃ N(16–15)	< 0.020		4.3"
07/10.21-10.30	1.170	0.280	80	$\text{HCOOH}(10_{0,10} - 9_{0,9})$	< 0.036		4.3"

Table 4. Molecular observations in comet C/2000 WM₁ (LINEAR).

				·			
UT date (2001)	$\langle r_{\rm h} \rangle$	$\langle \Delta \rangle$	Int. time	Species (transition)	$\int T_b \mathrm{d}v$	Velocity offset	Offset
[mm/dd.dd-dd.dd]	[AU]	[AU]	[min]		$[K km s^{-1}]$	$[\text{km s}^{-1}]$	
IRAM 30 m:				November 20			
11/23.03-23.06	1.359	0.387	35	HCN(3-2)	0.580 ± 0.092	-0.10 ± 0.08	5‴
11/23.03-24.21	1.345	0.376	320	HCN(1-0)	0.082 ± 0.007	-0.34 ± 0.05	4.5"
11/23.03-24.21	1.344	0.374	320	$H_2CO(3_{12} - 2_{11})$	0.069 ± 0.012	-0.30 ± 0.11	4.5"
11/23.03-27.14	1.317	0.356	615	CS(3–2)	0.020 ± 0.007	-0.19 ± 0.21	5.0"
11/23.03-27.14	1.317	0.356	615	CH ₃ OH(1, 0–1, –1)E	0.029 ± 0.004	-0.37 ± 0.09	5.0"
				CH ₃ OH(2, 0–2, –1)E	0.032 ± 0.004	-0.49 ± 0.09	
				CH ₃ OH(3, 0–3, –1)E	0.039 ± 0.004	-0.25 ± 0.06	
				CH ₃ OH(4, 0–4, –1)E	0.050 ± 0.004	-0.28 ± 0.05	
				CH ₃ OH(5, 0–5, –1)E	0.049 ± 0.005	-0.25 ± 0.06	
				CH ₃ OH(6, 0–6, –1)E	0.037 ± 0.004	-0.23 ± 0.07	
				CH ₃ OH(7, 0–7, –1)E	0.035 ± 0.005	-0.15 ± 0.09	
11/23.82-24.21	1.343	0.374	285	HCN(3-2)	0.689 ± 0.014	-0.16 ± 0.01	4.5"
11/23.98-27.14	1.321	0.358	555	$H_2S(1_{10} - 1_{01})$	0.059 ± 0.008	-0.24 ± 0.08	4.7"
11/24.78-24.80	1.330	0.364	20	HCN(3-2)	0.981 ± 0.052	-0.16 ± 0.03	4.6"
11/24.78-25.18	1.327	0.362	275	HCN(1-0)	0.069 ± 0.006	-0.34 ± 0.06	4.9″
11/24.78-26.13	1.330	0.356	520	CO(2–1)	< 0.022		4.6"
11/24.84-26.13	1.320	0.356	480	HNC(3-2)	< 0.044		5.0"
11/25.75-25.79	1.314	0.353	20	HCN(3-2)	0.813 ± 0.068	-0.05 ± 0.05	5.0"
			5	HCN(3-2)	0.777 ± 0.136	-0.39 ± 0.11	5.5"
			10	HCN(3-2)	0.489 ± 0.095	-0.05 ± 0.10	11.0"
11/25.75-26.02	1.313	0.352	180	CH ₃ CN(8, 0–7, 0)	0.016 ± 0.007		
				CH ₃ CN(8, 1–7, 1)	0.009 ± 0.007	-0.60 ± 0.22	5.0"
				CH ₃ CN(8, 2–7, 2)	0.015 ± 0.007		
11/25.75-26.13	1.312	0.351	245	HCN(1-0)	0.101 ± 0.007	-0.26 ± 0.04	5.0"
11/26.75-26.78	1.298	0.343	25	HCN(3-2)	0.957 ± 0.066	-0.27 ± 0.05	5.0"
11/26.75-27.14	1.295	0.341	485	CS(5-4)	0.118 ± 0.011	-0.01 ± 0.05	4.7"
CSO 10 m:				December 20			
12/03.18-03.25	1.193	0.317	48.0	HCN(3-2)	0.490 ± 0.015	-0.14 ± 0.12	9″
12/03.21-03.23	1.194	0.317	10.6	HCN(3-2)	0.580 ± 0.035	-0.06 ± 0.03	9.5″
			10.6	HCN(3-2)	0.308 ± 0.034	-0.05 ± 0.07	21"
12/03.28-03.40	1.191	0.317	104.	$CH_3OH(2, 1-2, 0)A-+$	0.170 ± 0.013	-0.18 ± 0.04	6‴
				CH ₃ OH(4, 1–4, 0)A–+	0.196 ± 0.013	-0.18 ± 0.04	
12/03.41-05.40	1.176	0.319	192	HNC(3-2)	< 0.021		5‴
12/04.17-04.21	1.178	0.319	32.0	HCN(3-2)	0.492 ± 0.021	-0.11 ± 0.03	5‴
12/04.21-04.22	1.178	0.319	8.0	HCN(3-2)	0.531 ± 0.041	$+0.07\pm0.05$	4‴
12/04.38-04.39	1.175	0.319	8.0	HCN(3-2)	0.588 ± 0.035	-0.08 ± 0.04	5″
12/04.40-05.35	1.165	0.322	133.	CH ₃ OH(5, 0–4, 0)A+	0.079 ± 0.012	-0.21 ± 0.10	5″
				CH ₃ OH(5, -1-4, -1)E	0.074 ± 0.012	-0.16 ± 0.10	
				CH ₃ OH(5, 0–4, 0)E	0.043 ± 0.013		
				$CH_3OH(5, 1-4, 1)E$	0.044 ± 0.014		
				CH ₃ OH(5, 2–4, 2)A+	0.049 ± 0.012		
				CH ₃ OH(5, 2–4, 2)E	0.084 ± 0.014		
12/04.40-05.35	1.165	0.322	133.	CS(5–4)	0.083 ± 0.011	-0.02 ± 0.08	5″
12/05.16-05.21	1.162	0.322	21.3	HCN(3-2)	0.620 ± 0.028	-0.06 ± 0.03	3.5"
			21.3	HCN(3-2)	0.382 ± 0.028	-0.08 ± 0.05	14.5"
12/05.35-05.37	1.158	0.323	16.0	HCN(3-2)	0.487 ± 0.033	-0.07 ± 0.04	5‴
12/06.18-06.35	1.145	0.327	24.0	HCN(3-2)	0.582 ± 0.024	-0.04 ± 0.03	4.5"
12/06.23-06.34	1.144	0.328	104.	$H_2CO(3_{12} - 2_{11})$	0.061 ± 0.010	$+0.04\pm0.10$	4.5"
	1.125	0.336	152.	CS(7–6)	0.149 ± 0.022	-0.26 ± 0.10	4‴
12/06.36-08.29	1 1 2 5	0.336	216.	CO(3–2)	0.104 ± 0.020	-0.20 ± 0.12	4‴
12/06.36-08.29 12/06.36-08.29	1.125		12.2	HCN(3-2)	0.526 ± 0.033	$+0.02\pm0.04$	4‴
	1.125	0.334	13.3				14 511
12/06.36–08.29 12/07.17–07.20	1.129		10.7	HCN(3-2)	0.462 ± 0.036	$+0.05\pm0.05$	
12/06.36-08.29		0.334 0.340		HCN(3–2) CH ₃ OH(2, 1–2, 0)A–+	$\begin{array}{c} 0.462 \pm 0.036 \\ 0.173 \pm 0.014 \end{array}$	$+0.05 \pm 0.05$ -0.39 ± 0.06	
12/06.36–08.29 12/07.17–07.20	1.129 1.116		10.7	HCN(3-2)			
12/06.36–08.29 12/07.17–07.20	1.129		10.7	HCN(3–2) CH ₃ OH(2, 1–2, 0)A–+	0.173 ± 0.014	-0.39 ± 0.06	14.5" 4.8" 4.8"

 Table 5. Molecular observations in comet 153P/Ikeya-Zhang.

UT date (2002)	$\langle r_{\rm h} \rangle$	$\langle \Delta \rangle$	Integration	Species (transition)	$\int T_b dv$	Velocity offset	Offset
[mm/dd.dd–dd.dd]	(/h/ [AU]	(Δ) [AU]	time [min]	Species (transition)	$[K \text{ km s}^{-1}]$	$[\text{km s}^{-1}]$	Onset
IRAM 30 m:	[AU]	[AU]	time [iiiii]	March 2002:			
03/19.74–19.78	0.507	0.802	35	HCN(3-2)	4.193 ± 0.167	-0.04 ± 0.09	6.5″
03/19.74–19.78	0.507	0.802	35	HNC(3–2)	0.398 ± 0.161	-0.19 ± 0.09	8.8″
03/19.74–19.78	0.507	0.802	35	CS(3–2)	0.306 ± 0.033	$+0.18 \pm 0.13$	6″
03/19.74-20.61	0.507	0.795	90	$CH_3OH(3, 0-3, -1)E$	0.086 ± 0.024	-0.02 ± 0.29	10"
05/17.74-20.01	0.507	0.775)0	$CH_3OH(5, 0-5, -1)E$ $CH_3OH(5, 0-5, -1)E$	0.000 ± 0.024 0.122 ± 0.025	-0.26 ± 0.22	10
				CH ₃ OH(3, 0–3, –1)E CH ₃ OH(7, 0–7, –1)E	0.122 ± 0.025 0.104 ± 0.025	-0.54 ± 0.29	
03/20.46-20.57	0.508	0.789	55	CS(3–2)	0.342 ± 0.019	$+0.06 \pm 0.07$	3.1″
03/20.10 20.37	0.200	0.702	10	CS(3-2)	0.308 ± 0.044	$+0.20 \pm 0.17$	7.0"
			25	CS(3-2)	0.266 ± 0.032	$+0.20 \pm 0.17$ $+0.48 \pm 0.15$	10.2"
			10	CS(3-2)	0.169 ± 0.045	$+0.42 \pm 0.33$	13.2"
03/20.46-20.57	0.508	0.789	55	$CH_3CN(8, 0-7, 0)$	0.063 ± 0.018	10.12 ± 0.55	3.3"
03/20.10 20.37	0.200	0.702	55	$CH_3CN(8, 1-7, 1)$	0.037 ± 0.018		0.0
				$CH_3CN(8, 2-7, 2)$	0.069 ± 0.018 0.069 ± 0.018		
				$CH_3CN(0, 2-7, 2)$ $CH_3CN(8, 3-7, 3)$	0.009 ± 0.010 0.045 ± 0.022		
03/20.46-20.57	0.508	0.789	55	$CH_3OH(1, 0-1, -1)E$	0.013 ± 0.022 0.053 ± 0.030		3″
03/20.10 20.37	0.500	0.702	55	$CH_{3}OH(2, 0-2, -1)E$	0.000 ± 0.000 0.100 ± 0.000		5
				$CH_3OH(2, 0-2, -1)E$ $CH_3OH(3, 0-3, -1)E$	0.100 ± 0.030 0.108 ± 0.033	$+0.11 \pm 0.30$	
				$CH_{3}OH(4, 0-4, -1)E$ CH ₃ OH(4, 0-4, -1)E	0.108 ± 0.033 0.188 ± 0.032	$+0.11 \pm 0.30$ $+0.40 \pm 0.20$	
				$CH_3OH(4, 0-4, -1)E$ $CH_3OH(5, 0-5, -1)E$	0.100 ± 0.032 0.104 ± 0.031	$+0.02 \pm 0.23$	
				$CH_3OH(6, 0-6, -1)E$	0.121 ± 0.031	$+0.27 \pm 0.23$	
				CH ₃ OH(0, 0–0, –1)E CH ₃ OH(7, 0–7, –1)E	0.121 ± 0.032 0.159 ± 0.028	-0.09 ± 0.16	
03/20.46-20.61	0.508	0.788	55	HCN(3-2)	6.801 ± 0.060	$+0.05 \pm 0.02$	2.6"
03/20.40-20.01	0.508	0.788	10	HCN(3-2) HCN(3-2)	4.209 ± 0.139	$+0.03 \pm 0.02$ +0.01 ± 0.08	2.0 7.7″
			10	HCN(3-2)	4.209 ± 0.139 2.822 ± 0.127	$+0.01 \pm 0.00$ $+0.15 \pm 0.11$	9.7″
			10	HCN(3-2) HCN(3-2)	2.322 ± 0.127 2.240 ± 0.135	$+0.10 \pm 0.11$ $+0.10 \pm 0.15$	10.9"
			10	HCN(3-2) HCN(3-2)	1.734 ± 0.128	$+0.10 \pm 0.13$ $+0.54 \pm 0.18$	12.5"
03/20.46-20.61	0.508	0.788	55	HOR(3-2) HNC(3-2)	1.666 ± 0.075	$+0.15 \pm 0.05$	4.6"
05/20.40-20.01	0.500	0.700	10	HNC(3–2)	1.266 ± 0.073	$+0.15 \pm 0.05$ $+0.15 \pm 0.14$	5.7"
			25	HNC(3–2)	0.672 ± 0.107	-0.13 ± 0.14 -0.13 ± 0.18	11.0"
			10	HNC(3-2)	< 0.507	0.15 ± 0.10	14.5"
03/20.64-20.68	0.509	0.786	40	HCN(1-0)	0.306 ± 0.019	-0.02 ± 0.07	2.5"
03/20.64-20.68	0.509	0.786	40	HNC(1-0)	0.063 ± 0.026	0.02 ± 0.07	2.5 4″
03/20.64-20.68	0.509	0.786	80	CO(2–1)	0.129 ± 0.020	$+0.33 \pm 0.24$	3″
IRAM 30 m:	0.007	0.700	00	April 2002:	0.12) ± 0.050	10.55 ± 0.21	5
04/29.24–29.37	1.039	0.405	100	HCN(3–2)	5.225 ± 0.046	-0.06 ± 0.02	2.5"
04/29.24-29.37	1.039	0.405	100	HNC(3–2)	0.271 ± 0.048	$+0.20 \pm 0.14$	2.5"
04/29.24–29.37	1.039	0.405	100	$CH_3CN(8, 0-7, 0)$	0.051 ± 0.012	10.20 ± 0.11	2.5"
0.1/2/12/12/12/10/	11007	01100	100	$CH_3CN(8, 1-7, 1)$	0.039 ± 0.012		2.0
				$CH_3CN(8, 2-7, 2)$	0.037 ± 0.012 0.037 ± 0.012		
				$CH_3CN(8, 3-7, 3)$	0.037 ± 0.012 0.035 ± 0.011		
04/29.24-29.37	1.039	0.405	100	$CH_3OH(1, 0-1, -1)E$	0.183 ± 0.017	-0.06 ± 0.07	2.5"
0 1/29.21 29.37	1.057	0.105	100	$CH_3OH(2, 0-2, -1)E$	0.105 ± 0.017 0.276 ± 0.017	-0.06 ± 0.07	2.5
				$CH_3OH(2, 0, 2, -1)E$ $CH_3OH(3, 0-3, -1)E$	0.308 ± 0.017	-0.11 ± 0.04	
				$CH_3OH(4, 0-4, -1)E$	0.300 ± 0.017 0.410 ± 0.018	-0.13 ± 0.03	
				$CH_3OH(5, 0-5, -1)E$	0.364 ± 0.017	-0.08 ± 0.04	
				$CH_3OH(6, 0-6, -1)E$	0.279 ± 0.017	-0.02 ± 0.01	
				$CH_3OH(0, 0-0, -1)E$ $CH_3OH(7, 0-7, -1)E$	0.279 ± 0.018 0.300 ± 0.018	-0.02 ± 0.05 -0.04 ± 0.05	
04/29.24-29.37	1.039	0.405	100	CS(3-2)	0.300 ± 0.013 0.178 ± 0.013	$+0.01 \pm 0.05$	2.5″
04/30.18-30.19	1.059	0.405	100	HCN(3-2)	5.779 ± 0.111	$+0.01 \pm 0.03$ $+0.00 \pm 0.04$	2.5
04/30.18–30.35	1.054	0.405	128	$H_2CO(3_{12} - 2_{11})$	0.361 ± 0.019	$+0.06 \pm 0.04$ $+0.06 \pm 0.04$	2.5"
04/30.18-30.35	1.056	0.405	128	$H_2CO(3_{12} - 2_{11})$ $H_2S(1_{10} - 1_{01})$	0.301 ± 0.019 1.106 ± 0.017	$+0.00 \pm 0.04$ $+0.01 \pm 0.01$	2.5 2.5″
04/30.18-30.35	1.056	0.405	128	$H_{2}S(1_{10} - 1_{01})$ HCN(1-0)	0.372 ± 0.013	-0.01 ± 0.01	2.5" 2.5"
04/30.18-30.35	1.056	0.405	138	$CH_3OH(4, 1-3, 2)E$	0.372 ± 0.013 0.071 ± 0.021	0.01 ± 0.03	2.5" 2.5"
04/30.20-30.35	1.056	0.405	115	HNC(3–2)	0.323 ± 0.031	$+0.14 \pm 0.07$	2.5″ 2.5″
07/00.20-00.00	1.050	0.405	115	111(C(J=2)	0.525 ± 0.051	10.17 ± 0.07	2.5

Table 5. continued.

UT date (2002) [mm/dd.dd–dd.dd]	$\langle r_{\rm h} \rangle$ [AU]	$\langle \Delta \rangle$ [AU]	Integration time [min]	Species (transition)	$\int T_b dv$ [K km s ⁻¹]	Velocity offset [km s ⁻¹]	Offse
IRAM 30 m:	[AU]	[AU]	time [mm]	continued May 200			
	1 102	0.425	20	continued – May 200		.0.00 . 0.10	8'
05/08.33–08.37 05/08.33–11.37	1.192	0.425	20 66	HCN(3–2)	2.324 ± 0.222 0.140 ± 0.035	$+0.08 \pm 0.18$ -0.08 ± 0.12	8 6'
	1.217	0.454	00	$CH_3OH(1, 0-1, -1)E$			0
				$CH_3OH(2, 0-2, -1)E$	0.151 ± 0.035	-0.06 ± 0.14	
				$CH_3OH(3, 0-3, -1)E$	0.236 ± 0.035	-0.36 ± 0.10 -0.25 ± 0.13	
				CH ₃ OH(4, 0–4, –1)E CH ₃ OH(5, 0–5, –1)E	0.202 ± 0.035 0.171 ± 0.035	-0.25 ± 0.15 -0.25 ± 0.15	
				$CH_3OH(5, 0-5, -1)E$ $CH_3OH(6, 0-6, -1)E$	0.171 ± 0.035 0.143 ± 0.035	-0.25 ± 0.15 +0.14 ± 0.14	
				$CH_3OH(0, 0=0, =1)E$ $CH_3OH(7, 0=7, =1)E$	0.143 ± 0.033 0.130 ± 0.035	$+0.14 \pm 0.14$ -0.12 ± 0.13	
05/11/22 11/27	1.242	0.443	21	HCN(3-2)	0.130 ± 0.033 2.014 ± 0.078	-0.12 ± 0.13 -0.07 ± 0.03	5.5
05/11.33-11.37	1.242	0.445	21	HCN(3-2) HCN(3-2)	1.791 ± 0.107	-0.10 ± 0.05 -0.10 ± 0.05	7.3
			o 11	HCN(3–2) HCN(3–2)	0.833 ± 0.093	-0.10 ± 0.03 +0.04 ± 0.08	14.5
05/11.33-11.37	1.242	0.443	21	HCN(3-2) HCN(1-0)	0.033 ± 0.093 0.178 ± 0.024	-0.18 ± 0.11	5.5
05/11.33–11.37	1.242	0.443	21	CS(5-4)	0.178 ± 0.024 0.336 ± 0.044	$+0.06 \pm 0.09$	5.5
05/11.55-11.57	1.242	0.445	19	CS(5-4)	0.350 ± 0.044 0.254 ± 0.049	-0.00 ± 0.09	11
05/11.97-11.98	1.253	0.447	5	HCN(3-2)	0.234 ± 0.049 2.910 ± 0.097	-0.09 ± 0.03	2
05/11.97-11.98	1.235	0.447	3	HCN(3–2) HCN(3–2)	2.310 ± 0.097 2.338 ± 0.128	-0.09 ± 0.03 -0.11 ± 0.04	2 5'
05/11.97-12.35	1.257	0.449	308	HCN(1-0)	0.232 ± 0.006	-0.11 ± 0.04 -0.18 ± 0.02	2
05/11.97–12.35	1.257	0.449	270	$CH_3OH(8, 0-7, 1)E$	0.232 ± 0.000 0.138 ± 0.007	-0.13 ± 0.02 -0.11 ± 0.04	2
05/11.97–12.35	1.255	0.448	270	HNCO(10, 0, 10–9, 0, 9)	0.043 ± 0.007 0.043 ± 0.007	$+0.14 \pm 0.04$	2
05/11.97-12.35	1.255	0.448	273	$SO(6_5 - 5_4)$	<0.018	$+0.14 \pm 0.11$	2
05/11.97–12.35	1.255	0.448	308	OCS(12-11)	<0.018		2
05/11.97-12.35	1.255	0.448	308	$HC_3N(16-15)$	<0.020		2
05/11.97–12.35	1.255	0.448	273	HCOOH(10, 0, 10–9, 0, 9)	<0.030		2
05/11.99–12.18	1.255	0.448	165	NS(11/2–9/2)e	<0.019		2
05/11.99-12.18	1.2.54	0.447	105	NS(11/2-9/2)c NS(11/2-9/2)f	<0.054		2
05/11.99-12.18	1.254	0.447	165	$CH_3OH(2, 0-1, -1)E$	0.099 ± 0.021	-0.05 ± 0.21	2
05/12.19–12.35	1.254	0.447	105	CO(2-1)	0.099 ± 0.021 0.114 ± 0.014	-0.03 ± 0.21 -0.04 ± 0.09	2
Kitt Peak 12 m:	1.237	0.449	170	March 2002:	0.114 ± 0.014	-0.04 ± 0.09	2
03/29.94–29.95	0.570	0.629	12.0	HCN(3-2)	1.560 ± 0.173		10
CSO 10 m:	0.370	0.029	12.0	April 2002:	1.300 ± 0.173		10
04/25.44-25.48	0.975	0.409	18.6	HCN(3-2)	2.541 ± 0.091	-0.02 ± 0.04	5
04/23.44-23.48	0.975	0.409	5.4	HCN(3–2) HCN(3–2)	2.541 ± 0.091 2.582 ± 0.110	-0.02 ± 0.04 -0.08 ± 0.04	6.5
			5.4	HCN(3–2) HCN(3–2)	1.520 ± 0.145	$+0.03 \pm 0.04$ $+0.03 \pm 0.09$	14.8
04/25.50-25.68	0.977	0.409	114	CH ₃ OH(3, 3–3, 2)A–+	1.520 ± 0.145 0.083 ± 0.016	-0.47 ± 0.17	4
04/25.50-25.08	0.977	0.409	114	$CH_3OH(3, 3-3, 2)A+-$ $CH_3OH(3, 3-3, 2)A+-$	0.083 ± 0.010 0.087 ± 0.018	-0.11 ± 0.14	4
				$CH_{3}OH(4, 3-4, 2)A+-$ $CH_{3}OH(4, 3-4, 2)A-+$	0.037 ± 0.013 0.107 ± 0.019	-0.11 ± 0.14	
					0.107 ± 0.019		
				$CH_{2}OH(4 + 3 + 4 + 2)A +$	0.126 ± 0.018	-0.18 ± 0.11	
				$CH_3OH(4, 3-4, 2)A+-$ $CH_2OH(5, 3, 5, 2)A+-$	0.126 ± 0.018 0.139 ± 0.020	-0.18 ± 0.11 -0.07 ± 0.11	
				CH ₃ OH(5, 3–5, 2)A–+	0.139 ± 0.020	-0.07 ± 0.11	
				CH ₃ OH(5, 3–5, 2)A–+ CH ₃ OH(5, 3–5, 2)A+–	0.139 ± 0.020 0.155 ± 0.018		
				CH ₃ OH(5, 3–5, 2)A++ CH ₃ OH(5, 3–5, 2)A+- CH ₃ OH(6, 3–6, 2)A++	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \end{array}$	-0.07 ± 0.11 -0.07 ± 0.09	
				CH ₃ OH(5, 3–5, 2)A–+ CH ₃ OH(5, 3–5, 2)A+– CH ₃ OH(6, 3–6, 2)A+– CH ₃ OH(6, 3–6, 2)A+–	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \end{array}$	-0.07 ± 0.11	
				CH ₃ OH(5, 3–5, 2)A–+ CH ₃ OH(5, 3–5, 2)A+– CH ₃ OH(6, 3–6, 2)A+– CH ₃ OH(6, 3–6, 2)A+– CH ₃ OH(7, 3–7, 2)A++	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \end{array}$	-0.07 ± 0.11 -0.07 ± 0.09 -0.18 ± 0.07	
				CH ₃ OH(5, 3–5, 2)A–+ CH ₃ OH(5, 3–5, 2)A+– CH ₃ OH(6, 3–6, 2)A+– CH ₃ OH(6, 3–6, 2)A+– CH ₃ OH(7, 3–7, 2)A++ CH ₃ OH(7, 3–7, 2)A+–	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \end{array}$	-0.07 ± 0.11 -0.07 ± 0.09	
				CH ₃ OH(5, 3–5, 2)A–+ CH ₃ OH(5, 3–5, 2)A+– CH ₃ OH(6, 3–6, 2)A+– CH ₃ OH(6, 3–6, 2)A+– CH ₃ OH(7, 3–7, 2)A+– CH ₃ OH(7, 3–7, 2)A+– CH ₃ OH(8, 3–8, 2)A–+	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \end{array}$	-0.07 ± 0.11 -0.07 ± 0.09 -0.18 ± 0.07	
				$CH_{3}OH(5, 3-5, 2)A-+ \\CH_{3}OH(5, 3-5, 2)A+- \\CH_{3}OH(6, 3-6, 2)A+- \\CH_{3}OH(6, 3-6, 2)A+- \\CH_{3}OH(6, 3-6, 2)A+- \\CH_{3}OH(7, 3-7, 2)A+- \\CH_{3}OH(7, 3-7, 2)A+- \\CH_{3}OH(8, 3-8, 2)A+- \\CH_{$	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \end{array}$	-0.07 ± 0.11 -0.07 ± 0.09 -0.18 ± 0.07	
				$CH_{3}OH(5, 3-5, 2)A-+ \\CH_{3}OH(5, 3-5, 2)A+- \\CH_{3}OH(5, 3-6, 2)A+- \\CH_{3}OH(6, 3-6, 2)A+- \\CH_{3}OH(6, 3-6, 2)A+- \\CH_{3}OH(7, 3-7, 2)A+- \\CH_{3}OH(7, 3-7, 2)A+- \\CH_{3}OH(8, 3-8, 2)A+- \\CH_{3}OH(8, 3-8, 2)A+- \\CH_{3}OH(9, 3-9, 2)A-+ \\CH_{$	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \end{array}$	-0.07 ± 0.11 -0.07 ± 0.09 -0.18 ± 0.07	
04/26 47 26 40	0.002	0.407	80	$CH_{3}OH(5, 3-5, 2)A-+\\CH_{3}OH(5, 3-5, 2)A+-\\CH_{3}OH(6, 3-6, 2)A+-\\CH_{3}OH(6, 3-6, 2)A+-\\CH_{3}OH(6, 3-6, 2)A+-\\CH_{3}OH(7, 3-7, 2)A+-\\CH_{3}OH(7, 3-7, 2)A+-\\CH_{3}OH(8, 3-8, 2)A+-\\CH_{3}OH(8, 3-8, 2)A+-\\CH_{3}OH(9, 3-9, 2)A$	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.072 \pm 0.019 \end{array}$	-0.07 ± 0.11 -0.07 ± 0.09 -0.18 ± 0.07 -0.18 ± 0.08	4
04/26.47-26.49	0.992	0.407	8.0 40.3	$CH_{3}OH(5, 3-5, 2)A-+ \\CH_{3}OH(5, 3-5, 2)A+- \\CH_{3}OH(6, 3-6, 2)A+- \\CH_{3}OH(6, 3-6, 2)A+- \\CH_{3}OH(6, 3-6, 2)A+- \\CH_{3}OH(7, 3-7, 2)A+- \\CH_{3}OH(7, 3-7, 2)A+- \\CH_{3}OH(8, 3-8, 2)A+- \\CH_{3}OH(8, 3-8, 2)A+- \\CH_{3}OH(9, 3-9, 2)A+- \\CH_{3}OH(9, 3-9, 2)A+- \\HCN(3-2)$	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.072 \pm 0.019 \\ 2.444 \pm 0.069 \end{array}$	-0.07 ± 0.11 -0.07 ± 0.09 -0.18 ± 0.07 -0.18 ± 0.08 -0.01 ± 0.03	
04/26.49-26.54	0.992	0.407	49.3	$CH_{3}OH(5, 3-5, 2)A-+ \\CH_{3}OH(5, 3-5, 2)A+- \\CH_{3}OH(6, 3-6, 2)A+- \\CH_{3}OH(6, 3-6, 2)A+- \\CH_{3}OH(6, 3-6, 2)A+- \\CH_{3}OH(7, 3-7, 2)A+- \\CH_{3}OH(7, 3-7, 2)A+- \\CH_{3}OH(8, 3-8, 2)A+- \\CH_{3}OH(8, 3-8, 2)A+- \\CH_{3}OH(8, 3-8, 2)A+- \\CH_{3}OH(9, 3-9, 2)A+- \\CH_{3}OH(9, 3-9, 2)A+- \\HCN(3-2) \\HNC(3-2)$	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.072 \pm 0.019 \\ 2.444 \pm 0.069 \\ 0.131 \pm 0.022 \end{array}$	-0.07 ± 0.11 -0.07 ± 0.09 -0.18 ± 0.07 -0.18 ± 0.08	4.5
04/26.49–26.54 04/26.55–26.69	0.992 0.994	0.407 0.407	49.3 128	$\begin{array}{c} {\rm CH_3OH(5, 3-5, 2)A-+}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm HCN(3-2)}\\ {\rm HNC(3-2)}\\ {\rm HDO(1_{10}-1_{01})}\\ \end{array}$	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.072 \pm 0.019 \\ 2.444 \pm 0.069 \\ 0.131 \pm 0.022 \\ < 0.855 \end{array}$	-0.07 ± 0.11 -0.07 ± 0.09 -0.18 ± 0.07 -0.18 ± 0.08 -0.01 ± 0.03 -0.10 ± 0.18	4.5 4
04/26.49-26.54	0.992	0.407	49.3 128 12.0	$\begin{array}{c} {\rm CH_3OH(5, 3-5, 2)A-+}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm HCN(3-2)}\\ {\rm HDC(1_{10}-1_{01})}\\ {\rm HCN(3-2)}\\ \end{array}$	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.072 \pm 0.019 \\ 2.444 \pm 0.069 \\ 0.131 \pm 0.022 \\ < 0.855 \\ 2.490 \pm 0.052 \end{array}$	-0.07 ± 0.11 -0.07 ± 0.09 -0.18 ± 0.07 -0.18 ± 0.08 -0.01 ± 0.03 -0.10 ± 0.18 -0.03 ± 0.02	4.5 4 3.4
04/26.49–26.54 04/26.55–26.69	0.992 0.994	0.407 0.407	49.3 128 12.0 2.7	$\begin{array}{c} {\rm CH_3OH(5, 3-5, 2)A-+}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm HCN(3-2)}\\ {\rm HDC(13-2)}\\ {\rm HDC(13-2)}\\ {\rm HCN(3-2)}\\ {\rm $	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.072 \pm 0.019 \\ 0.072 \pm 0.019 \\ 2.444 \pm 0.069 \\ 0.131 \pm 0.022 \\ < 0.855 \\ 2.490 \pm 0.052 \\ 2.265 \pm 0.117 \end{array}$	$\begin{array}{c} -0.07 \pm 0.11 \\ -0.07 \pm 0.09 \\ -0.18 \pm 0.07 \\ -0.18 \pm 0.08 \\ \end{array}$ $\begin{array}{c} -0.01 \pm 0.03 \\ -0.10 \pm 0.18 \\ -0.03 \pm 0.02 \\ -0.04 \pm 0.05 \end{array}$	4.5 4 3.4 5.5
04/26.49–26.54 04/26.55–26.69	0.992 0.994	0.407 0.407	49.3 128 12.0 2.7 6.7	$\begin{array}{c} {\rm CH_3OH(5, 3-5, 2)A-+}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm HCN(3-2)}\\ {\rm HDO(1_{10}-1_{01})}\\ {\rm HCN(3-2)}\\ {\rm HCN(3-2)$	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.072 \pm 0.019 \\ 0.072 \pm 0.019 \\ 2.444 \pm 0.069 \\ 0.131 \pm 0.022 \\ < 0.855 \\ 2.490 \pm 0.052 \\ 2.265 \pm 0.117 \\ 1.879 \pm 0.072 \end{array}$	$\begin{array}{c} -0.07 \pm 0.11 \\ -0.07 \pm 0.09 \\ -0.18 \pm 0.07 \\ -0.18 \pm 0.08 \\ \end{array}$ $\begin{array}{c} -0.01 \pm 0.03 \\ -0.10 \pm 0.18 \\ -0.03 \pm 0.02 \\ -0.04 \pm 0.05 \\ -0.06 \pm 0.04 \end{array}$	4.5 4 3.4 5.5 9.3
04/26.49–26.54 04/26.55–26.69 04/27.40–27.47	0.992 0.994 1.009	0.407 0.407 0.406	49.3 128 12.0 2.7 6.7 10.7	$\begin{array}{c} {\rm CH_3OH(5, 3-5, 2)A-+}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm HCN(3-2)}\\ {\rm HDC(13-2)}\\ {\rm HDC(13-2)}\\ {\rm HCN(3-2)}\\ {\rm $	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.072 \pm 0.019 \\ 0.072 \pm 0.019 \\ 0.131 \pm 0.022 \\ < 0.855 \\ 2.490 \pm 0.052 \\ 2.265 \pm 0.117 \\ 1.879 \pm 0.072 \\ 1.411 \pm 0.074 \end{array}$	$\begin{array}{c} -0.07 \pm 0.11 \\ -0.07 \pm 0.09 \\ -0.18 \pm 0.07 \\ -0.18 \pm 0.08 \\ \end{array}$ $\begin{array}{c} -0.01 \pm 0.03 \\ -0.10 \pm 0.18 \\ -0.03 \pm 0.02 \\ -0.04 \pm 0.05 \\ -0.06 \pm 0.04 \\ -0.03 \pm 0.05 \end{array}$	4.5 4 3.4 5.5 9.3 11.8
04/26.49–26.54 04/26.55–26.69	0.992 0.994	0.407 0.407	49.3 128 12.0 2.7 6.7 10.7 2.7	$\begin{array}{c} {\rm CH_3OH(5, 3-5, 2)A-+}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm HCN(3-2)}\\ {\rm HDC(13-2)}\\ {\rm HCN(3-2)}\\ {\rm HCN$	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.072 \pm 0.019 \\ 0.072 \pm 0.019 \\ 0.072 \pm 0.019 \\ 2.444 \pm 0.069 \\ 0.131 \pm 0.022 \\ < 0.855 \\ 2.490 \pm 0.052 \\ 2.265 \pm 0.117 \\ 1.879 \pm 0.072 \\ 1.411 \pm 0.074 \\ 2.649 \pm 0.084 \end{array}$	$\begin{array}{c} -0.07 \pm 0.11 \\ -0.07 \pm 0.09 \\ -0.18 \pm 0.07 \\ -0.18 \pm 0.08 \\ \end{array}$ $\begin{array}{c} -0.01 \pm 0.03 \\ -0.10 \pm 0.18 \\ -0.03 \pm 0.02 \\ -0.04 \pm 0.05 \\ -0.06 \pm 0.04 \\ -0.03 \pm 0.05 \\ -0.01 \pm 0.03 \end{array}$	4.5 4 3.4 5.5 9.3 11.8 3.3
04/26.49–26.54 04/26.55–26.69 04/27.40–27.47	0.992 0.994 1.009	0.407 0.407 0.406	49.3 128 12.0 2.7 6.7 10.7 2.7 5.3	$\begin{array}{c} {\rm CH_3OH(5, 3-5, 2)A-+}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm HCN(3-2)}\\ {\rm HDC(13-2)}\\ {\rm HDC(13-2)}\\ {\rm HCN(3-2)}\\ {\rm HC$	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.072 \pm 0.019 \\ 0.072 \pm 0.019 \\ 2.444 \pm 0.069 \\ 0.131 \pm 0.022 \\ < 0.855 \\ 2.490 \pm 0.052 \\ 2.265 \pm 0.117 \\ 1.879 \pm 0.072 \\ 1.411 \pm 0.074 \\ 2.649 \pm 0.084 \\ 2.190 \pm 0.055 \end{array}$	$\begin{array}{c} -0.07 \pm 0.11 \\ -0.07 \pm 0.09 \\ -0.18 \pm 0.07 \\ -0.18 \pm 0.08 \\ \end{array}$ $\begin{array}{c} -0.01 \pm 0.03 \\ -0.10 \pm 0.18 \\ -0.03 \pm 0.02 \\ -0.04 \pm 0.05 \\ -0.06 \pm 0.04 \\ -0.03 \pm 0.05 \\ -0.01 \pm 0.03 \\ -0.02 \pm 0.02 \end{array}$	4.5 4 3.4 5.5 9.3 11.8 3.3 8.3
04/26.49–26.54 04/26.55–26.69 04/27.40–27.47 04/27.47–27.55	0.992 0.994 1.009 1.010	0.407 0.407 0.406	49.3 128 12.0 2.7 6.7 10.7 2.7 5.3 4.0	$\begin{array}{c} {\rm CH_3OH(5, 3-5, 2)A-+}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm HCN(3-2)}\\ {\rm HDC(13-2)}\\ {\rm HDC(13-2)}\\ {\rm HCN(3-2)}\\ {\rm HC$	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.072 \pm 0.019 \\ 2.444 \pm 0.069 \\ 0.131 \pm 0.022 \\ < 0.855 \\ 2.490 \pm 0.052 \\ 2.265 \pm 0.117 \\ 1.879 \pm 0.072 \\ 1.411 \pm 0.074 \\ 2.649 \pm 0.084 \\ 2.190 \pm 0.055 \\ 1.579 \pm 0.063 \end{array}$	$\begin{array}{c} -0.07 \pm 0.11 \\ -0.07 \pm 0.09 \\ -0.18 \pm 0.07 \\ -0.18 \pm 0.08 \\ \end{array}$ $\begin{array}{c} -0.01 \pm 0.03 \\ -0.10 \pm 0.18 \\ -0.03 \pm 0.02 \\ -0.04 \pm 0.05 \\ -0.06 \pm 0.04 \\ -0.03 \pm 0.05 \\ -0.01 \pm 0.03 \\ -0.02 \pm 0.02 \\ -0.03 \pm 0.04 \\ \end{array}$	4.5 4 3.4 5.5 9.3 11.8 3.3 8.3 15.3
04/26.49–26.54 04/26.55–26.69 04/27.40–27.47	0.992 0.994 1.009	0.407 0.407 0.406	49.3 128 12.0 2.7 6.7 10.7 2.7 5.3	$\begin{array}{c} {\rm CH_3OH(5, 3-5, 2)A-+}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(5, 3-5, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(6, 3-6, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(7, 3-7, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(8, 3-8, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm CH_3OH(9, 3-9, 2)A+-}\\ {\rm HCN(3-2)}\\ {\rm HDC(13-2)}\\ {\rm HDC(13-2)}\\ {\rm HCN(3-2)}\\ {\rm HC$	$\begin{array}{c} 0.139 \pm 0.020 \\ 0.155 \pm 0.018 \\ 0.170 \pm 0.019 \\ 0.155 \pm 0.018 \\ 0.184 \pm 0.019 \\ 0.131 \pm 0.018 \\ 0.097 \pm 0.015 \\ 0.122 \pm 0.019 \\ 0.092 \pm 0.019 \\ 0.072 \pm 0.019 \\ 0.072 \pm 0.019 \\ 2.444 \pm 0.069 \\ 0.131 \pm 0.022 \\ < 0.855 \\ 2.490 \pm 0.052 \\ 2.265 \pm 0.117 \\ 1.879 \pm 0.072 \\ 1.411 \pm 0.074 \\ 2.649 \pm 0.084 \\ 2.190 \pm 0.055 \end{array}$	$\begin{array}{c} -0.07 \pm 0.11 \\ -0.07 \pm 0.09 \\ -0.18 \pm 0.07 \\ -0.18 \pm 0.08 \\ \end{array}$ $\begin{array}{c} -0.01 \pm 0.03 \\ -0.10 \pm 0.18 \\ -0.03 \pm 0.02 \\ -0.04 \pm 0.05 \\ -0.06 \pm 0.04 \\ -0.03 \pm 0.05 \\ -0.01 \pm 0.03 \\ -0.02 \pm 0.02 \end{array}$	4' 4.5' 4' 5.5' 9.3' 11.8' 3.3' 8.3' 15.3' 3.0' 5.0'

Table 5. continued.

	()	(•)	T ()	S : (1 :1:)	$\int T = 1$	V1 '4 0 4	000 /
UT date (2002)	$\langle r_{\rm h} \rangle$	$\langle \Delta \rangle$	Integration	Species (transition)	$\int T_b \mathrm{d}v$	Velocity offset	Offset
[mm/dd.dd–dd.dd]	[AU]	[AU]	time [min]		[K km s ⁻¹]	$[{\rm km}~{\rm s}^{-1}]$	
CSO 10 m:				April 2002 – continued:			
04/27.60-27.68	1.011	0.405	84.3	CH ₃ OH(3, 3–3, 2)A–+	0.017 ± 0.017		6''
				CH ₃ OH(3, 3–3, 2)A+–	0.069 ± 0.017		
				CH ₃ OH(4, 3–4, 2)A–+	0.113 ± 0.021		
				CH ₃ OH(4, 3–4, 2)A+–	0.076 ± 0.017		
				CH ₃ OH(5, 3–5, 2)A-+	0.103 ± 0.016	-0.15 ± 0.10	
				CH ₃ OH(5, 3–5, 2)A+–	0.104 ± 0.021		
				CH ₃ OH(6, 3–6, 2)A–+	0.127 ± 0.021		
				CH ₃ OH(6, 3–6, 2)A+–	0.108 ± 0.021		
				CH ₃ OH(7, 3–7, 2)A–+	0.085 ± 0.017		
				CH ₃ OH(7, 3–7, 2)A+–	0.138 ± 0.024		
				CH ₃ OH(8, 3–8, 2)A–+	0.087 ± 0.017		
				CH ₃ OH(8, 3–8, 2)A+–	0.133 ± 0.024		
				CH ₃ OH(9, 3–9, 2)A–+	0.081 ± 0.021		
				CH ₃ OH(9, 3–9, 2)A+–	0.107 ± 0.021		
				CH ₃ OH(10, 3–10, 2)A+–	0.102 ± 0.025		
04/27.60-27.68	1.011	0.405	84.3	$SO(5_6 - 4_5)$	< 0.036		6''
04/27.60-27.68	1.011	0.405	84.3	HC ₃ N(28–27)	< 0.036		6''